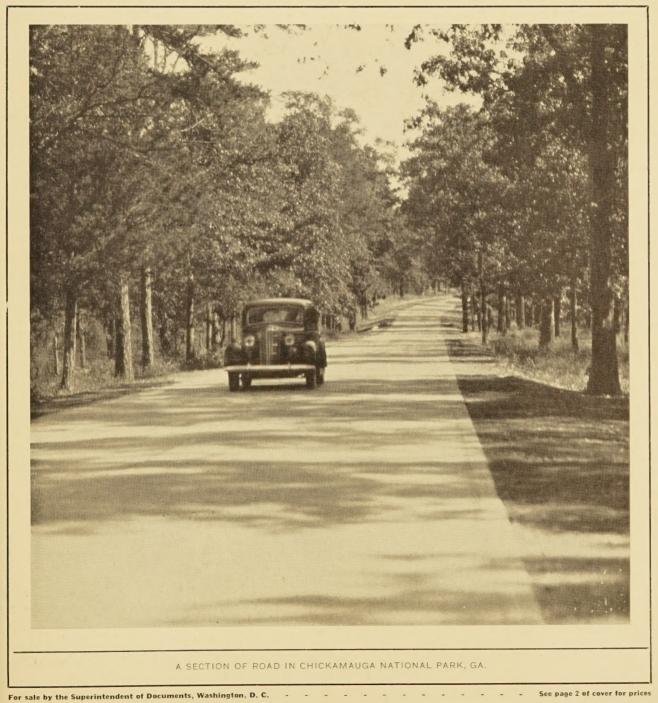


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The reports of research published in this magazine are necessarily qualified by the conditions of the tests from which the data are obtained. Whenever it is deemed possible to do so, generalizations are drawn from the results of the tests; and, unless this is done, the conclusions formulated must be considered as specifically pertinent only to described conditions.

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# THE EFFECT OF CURING CONDITIONS ON STRENGTH OF CONCRETE TEST SPECIMENS CONTAINING BURNT CLAY AGGREGATES

BY THE DIVISION OF TESTS, BUREAU OF PUBLIC ROADS

Reported by W. F. KELLERMANN, Associate Materials Engineer

THE TREND toward increased span lengths on highway bridges has stimulated efforts to decrease the dead load caused by the weight of the bridge floor. Results of a study of the relative strengths of test specimens of concrete bridge floor slabs made of several materials and placed by each of several methods have been published.1

Included among the concrete aggregates used was the artificial, lightweight aggregate known as "Havdite." The tests showed that there was a progressive retrogression in flexural strength for the Haydite concrete specimens for ages exceeding 90 days, whereas concrete made with siliceous gravel, crushed limestone, and quartz sand aggregates, cured under presumably the same conditions, continued to gain in strength. In view of the probability that this behavior of the Haydite specimens might be related to the moistureretaining qualities of the aggregate, it was decided to conduct further studies of Haydite concrete specimens. A supplementary series of laboratory tests under controlled curing conditions is discussed in this report.

Two series of concrete specimens, one containing fine and coarse Haydite as total aggregate and the other a natural sand and crushed limestone, were made and tested at various ages up to and including 1 year. In addition, a limited number of specimens made with Haydite aggregates were tested at an age of 428 days, and corresponding specimens made with natural aggregates were tested at 409 days. The mix used with the natural aggregates was 1: 2.1: 3.9 by dry loose volume (1: 2: 3.5 by dry-rodded volume). The mix used with the Haydite aggregates was 1: 1.5: 3 by dry loose volume, with 3 pounds of "Celite" added per sack of cement.

The latter mix was the same as that used in the bridge slab tests where the retrogression in flexural strength was noted. However, the Haydite used in the previous tests came from Jackson County, Missouri, while that used in these tests came from Erie County, New York. Though the materials were from different sources, both were manufactured by the same process and were submitted by the manufacturer as being representative of the type. The natural sand came from the Potomac River at Washington, the limestone from Martinsburg, W. Va. Physical characteristics of the aggregates are given in table 1, and the mix data are shown in table 2.

### MIXING AND TESTING PROCEDURES DESCRIBED

Because dried Haydite aggregate absorbs water rapidly, it was decided to presaturate all aggregates and use them in this condition. The procedure followed was to dry the materials in an oven, weigh out

the fine and coarse aggregates required for each batch in a watertight container, and immerse them for 48 hours in a measured quantity of water before mixing. By this procedure the total water used was accurately determined, yet thorough presaturation of the aggregates was accomplished. However, the net water-cement ratio reported for the Haydite concrete is only approximate, because of the difficulty of determining accurately the amount of water absorbed by the aggregates.

TABLE	1Gradin	and ph	usical :	properties	of aggregates

	Haydite fine aggregate	Potomac River sand	Haydite coarse aggregate	Martins- burg limestone
Retained on: 1½-inch sieve	Percent	Percent	Percent	Percent 20
34 inch sieve. 3√ inch sieve. 3% inch sieve. 4-mesh sieve. 8-mesh sieve. 16-mesh sieve. 30-mesh sieve.	14 48		$     \begin{array}{r}       17 \\       60 \\       98 \\       100 \\       100 \\       100     \end{array} $	49 74 87 100 100 100 100
50-mesh sieve 100-mesh sieve		86 97	100 100	100
Fineness modulus Physical properties:	2.91	3.12	6.58	7.36
Specific gravity <sup>1</sup> Absorption <sup>2</sup> percent Weight (dry loose) per cubic foot	$\begin{array}{c} 1.\ 63\\ 11.\ 4\end{array}$	$2.53 \\ 2.1$	1,42 7,8	2.71 0.13
VoidsPorcent	57 44	101 36	$\begin{array}{c} 45\\ 49\end{array}$	90 47

<sup>1</sup> Bulk specific gravity, A. S. T. M. definition E 12-27.
 <sup>2</sup> 70° cone method of Bureau of Public Roads.

TABLE 2.—Data on mixes used in making concrete specimens

Agg	gregates	Propor- tion by dry loose	Celite per sack of ce-	Wate tent sacl cem	z of	Con- sist- ency flow 1	Weight	Ce- ment factor	
Coarse	Fine	volume	ment	Total	Net	ПОМ т			
Haydite Limestone	Haydite Potomac River sand.	1:1.5:3.0 <sup>2</sup> 1:2.1:3.9	<i>Lb.</i> 3	Gal. 8.4 6.4	Gal. 5.9 5.8	140 165	<i>Lb.</i> <i>per</i> <i>cu. ft.</i> 104 151	Sacks per cu. yd. 7.2 5.7	

Fifteen ½-inch drops in 10 seconds.
 Proportions by dry-rodded volume 1:2:3.5.

The concrete was mixed with shovels in watertight pans, each batch being of sufficient size to make one flexure specimen (6- by 6- by 21-inch beam) or two 6- by 12- inch cylinders. After 1 day in the molds and 6 days in the moist room, the flexural specimens were subjected to the curing conditions indicated in table 3, and the compression specimens to the curing conditions shown in table 4. Moist curing was accomplished by storing the

<sup>&</sup>lt;sup>1</sup> The Effect of Materials and Methods of Placing on the Strength and Other Properties of Concrete Bridge Floor Slabs, by L. W. Teller and G. W. Davis. PUBLIC ROADS, vol. 12, no. 10, December 1931.

specimens in the moist room under standard conditions of humidity and temperature. The final wet curing consisted of immersion in water at laboratory temperatures prior to test.

Group no.	Initial moist curing	Air	Oven at 170° F.	Water at 70° F. prior to test	Age at test
	Days 7	Days	Days	Days	Days
	7	21			2
	28 7	173			2 18
	7	166		7	18
	7	169		4	18
	7	170		3	18
	7	171		2	18
	7	172		1	18
	173	7	7		18 1 18
	172 180	1	(		18
	.7	353			36
	7	346		7	36
	28	332			36
	180	180			36
	270	90			36
	28 180	325 173		77	36 36
	270	83		7	36
	353	7		· /	36
	352	l i	7		1 36
	359	1			36
	358	1	1		1 36
	360				36
	\$ 2 360	1	3 67		1 42
	4 360	1	3 48		1 4(
	$\left\{\begin{array}{c} 2 428 \\ 4 409 \end{array}\right.$				42 40
	409	421			40
	47	402			44

TABLE 3.—Curing conditions for concrete beam specimens

Oven-dried specimens cooled in air of laboratory for 1 day prior to test.
 Data for beams having Haydite aggregates.
 Dried in oven to constant weight prior to test.
 Data for beams having natural sand and crushed limestone aggregates.

TABLE 4.—Curing conditions and crushing strengths of cylinders

		Curi	ng					
	Age			Hayo	lite	Limes	tone	Strength ratio, Haydite
Group no.	attest	Moist room	Air	Crushing strength <sup>1</sup>	Coeffi- cient of varia- tion	Crushing strength <sup>1</sup>	Coeffi- cient of varia- tion	concrete to lime- stone concrete
1 2 3 4 12 13 25 A verage	Days 7 28 28 180 180 360 360	Days 7 -7 28 7 180 7 360	Days 21 173 353	Lb. per sq. in. 2, 300 3, 460 3, 970 5, 180 6, 490 5, 230 6, 850 4, 780	$\begin{array}{c} Percent \\ 3.5 \\ 6.2 \\ 4.6 \\ 1.4 \\ 5.6 \\ 4.4 \\ 1.9 \\ \hline 3.9 \end{array}$	$\begin{array}{c} Lb. \ per\\ sq. \ in.\\ 2, 280\\ 3, 510\\ 4, 040\\ 5, 270\\ 4, 250\\ 5, 410\\ \hline \hline \\ 4, 180\\ \end{array}$	Percent 3.8 2.1 5.4 3.7 1.8 4.4 3.2 3.5	Percent 101 99 98 116 123 123 123 127 114

<sup>1</sup> Each value represents the average of tests on 5 specimens.

Air curing was accomplished by storing the specimens in air in a frame structure which was heated only enough to keep the specimens from freezing. Twenty-four hours before test the air-cured specimens were moved to the laboratory in order to attain laboratory temperatures, and the final day was counted as part of the curing period. Oven-dried specimens were heated in an electric oven maintained at a temperature of 170°  $F. \pm 5^{\circ} F.$  and were brought to the laboratory 24 hours before test in order to cool. Each value given in tables represents the average of tests on five specimens in practically all cases.

The compression specimens were standard 6- by 12inch cylinders; the flexural specimens were 6- by 6- by 21-inch beams. Flexure specimens were weighed at the time they were fabricated, when they were removed from the molds, at each time the curing conditions were changed, and at the time of test. Thus it was possible to calculate the percentage of water remaining in each specimen at each change of curing condition and at test.

The flexure specimens were tested as simple beams on an 18-inch span with a side as molded, in tension. They were loaded at the third points with the apparatus described in PUBLIC ROADS (vol. 13, no. 11), January 1933. The compression specimens were tested in accordance with standard methods.

The percentage of original mixing water remaining in the flexural specimens at the periods indicated is shown in table 5. Similar data are not available for the cylinders, because it was necessary to cap them, thereby changing the weight as well as the composition of the individual specimens. The average flexural strength for each group tested is given in table 6; table 4 gives similar data as regards crushing strength.

#### HAYDITE CONCRETE RELATIVELY SLOW IN ABSORBING AND LOSING MOISTURE

Table 5 shows the amounts of water remaining in the beam specimens at various periods, expressed as percentages of the original mixing water. The results show that the Haydite concrete lost approximately 7 percent of its mixing water during the first 24 hours. After 6 days curing in the moist room the specimens had reabsorbed moisture until the water content averaged about 95 percent of the original amount. After 28 days the percentage had increased to about 97 percent, and complete saturation was reached at 180 days. For the limestone concrete the percentage of original mixing water at 24 hours averaged about 91 percent. After curing in the moist room for 6 days the water content was about 95 percent, and complete saturation was attained at 28 days. Complete saturation is considered to have been reached when the water content of the specimen was 100 percent of the original amount of water used.

The amounts of moisture lost by specimens cured in the moist room for 6 days, followed by curing in air for varying periods, are shown for groups 2, 4 to 9 inclusive, 13, 14, and 28 in table 5. It will be observed that for both aggregates the moisture loss reached a maximum after about 6 months' curing in air. Increasing the curing period beyond 6 months (groups 13, 14, and 28) appeared to have negligible effect insofar as drying was concerned.

Data for groups 5 to 9, inclusive, show the effect of immersing the specimens from 1 to 7 days in water after approximately 6 months curing in air. The initial curing period was 6 days in the moist room in each case. It will be observed that after 7 days of immersion (group 5) the limestone concrete had reabsorbed a considerably larger proportion of its original mixing water than had the Haydite concrete. Moreover, the rate of gain in weight after the first 24 hours in water appeared to be considerably less for the Haydite concrete.

Comparing groups 5 and 14 we find that for the limestone concrete, 7 days' curing in water was sufficient time for the specimens to regain all the moisture lost during the 6-month or 1-year curing in air. This was not true of the Haydite concrete, the amount of water reabsorbed after 7 days curing in water being substantially less than the amount present at the time the specimens were subjected to air curing.

#### TABLE 5.—Percentages of original mixing water remaining in concrete specimens after curing

# CONCRETE BEAM SPECIMENS CONTAINING HAYDITE AGGREGATES

	F	ercen	tage o	f origi	nal mi	xing v	tater re	emain	ing aft	er curi	ng in	
Group no.	Molds	Moist room			Air	0	ven	w	ater		oratory air	
10.	for 24 hours	Pe- riod	Wa- ter con- tent	Pe- riod	Wa- ter con- tent	Pe- riod	Wa- ter con- tent	Pe- riod	Wa- ter con- tent	Pe- riod	Wa- ter con- tent	At test
1	Per- cent 93	Days 6	Per- cent 94	Days	Per- cent	Days	Per- cent	Days	Per- cent	Days	Per- cent	Per- cent
2	93	6	96	20	73					1	71	94 71
3	92 94	27 6	96 96	172	62					1	62	96 62
5 6 7 8	93 92 93 92	6 6 6	96 94 95 94	166 169 170 171	62 60 62 59			7 4 3 2	82 81 81 80			82 81 81 80
9 10 11 12	94 93 94 93	6 172 171 179	97 100 100 100	172 6	62 95		63	1	81	1 1	95 64	81 95 64 100
13 14 15 16	93 93 94 93	6 6 27 179	96 94 98 100	352 346 331 179	61 60 70 83			7	84	1 1 1	61 70 84	61 84 70 84
17 18 19 20	93 93 93 93	269 27 179 269	100 96 98 99	89 325 173 83	88 69 83 88			7 7 7	87 92 95	1	88	88 87 92 95
21 22 23 24	94 96 93 94	352 351 358 357	101 101 100 101	6	97	7	74 90			1 1 1 1	96 75 97 90	96 75 97 90
25 26 27 28	94 94 95 93	359 359 427 6	$100 \\ 101 \\ 100 \\ 96$	352	60	67	29			1 69	30 58	$160 \\ 30 \\ 100 \\ 58$

	AND CRUSHED LIMESTONE														
1	92 90 93		96 93 100	20	76					1	75	96 75 100			
4	92	6	96	172	66					1	66	66			
5 6 7 8	91 92 90 91	6 6 6	96 95 94 96	166 169 170 171	$     \begin{array}{r}       67 \\       65 \\       66 \\       65     \end{array} $			7 4 3 2	95 95 92 91			95 95 92 91			
9 10 11 12	89 94 92 90	6 172 171 179	93 103 101 101	172 6	64 97	7	52	1	88	1 1	96 53				
13 14 15 16	92 92 92 92		96 96 100 100	352 346 331 179	67 67 70 80			7	97	1 1 1	67 70 79	67 97 70 79			
17 18 19 20	91 92 90 90	$269 \\ 27 \\ 179 \\ 269$	101 99 99 98	89 325 173 83	86 70 80 84			7 7 7 7	95 94 94	1	85	85 95 94 94			
21 22 23 24	92 91 90 91	352 351 358 <b>3</b> 57	$102 \\ 102 \\ 101 \\ 101 \\ 101$	6	98	7	63 88			1 1 1 1	97 63 97 88	97 63 97 88			
25 26 27	90 92 91	359 359 408	$     \begin{array}{c}       101 \\       103 \\       101     \end{array} $			48	32			1	34	$     \begin{array}{r}       101 \\       34 \\       101     \end{array} $			
28	92	6	95	352	66					50	63	63			

Data for groups 13, 15, 16, and 17 show the moisture content at 1 year, after the specimens had been subjected to moist curing followed by air curing for periods of various length. As might be expected, the percentage of water remaining at the end of the drying period decreased as the period of air curing increased. This

	145	1								Concrete
no.	test	Ini- tial moist	Oven	Air	Wa- ter	Mod- ulus of rup- ture	ficient	of rup	ficient	to lime- stone concrete
1	Days 7 28 180 180 180 180 180 180 180 180 180 360 360 360 360 360 360 360 360 360 36	Days 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	Days           7           7           7           7           67           48	Days 21 173 166 169 170 171 172 7 1 333 346 332 180 90 90 325 173 83 7 1 1 1 1	Days	$\begin{array}{c} Lb, \\ per\\ sq. in, \\ 386\\ 357\\ 533\\ 556\\ 556\\ 5544\\ 1572\\ 590\\ 1607\\ 391\\ 629\\ 512\\ 483\\ 488\\ 246\\ 296\\ 660\\ 585\\ 540\\ 666\\ 540\\ 666\\ 756\\ 376\\ 6779\\ 452\\ \end{array}$	$\begin{array}{c} Pct.\\ 6.6\\ 8.3\\ 1.6\\ 4.3\\ 11, 0\\ 7.1\\ 1.4.8\\ 8.8\\ 1.4.4\\ 13.2\\ 15.4\\ 6.5\\ 4.2\\ 10.9\\ 16.6\\ 6.5\\ 4.2\\ 10.9\\ 11.7\\ 7.6\\ 6.5\\ 4.2\\ 2.9\\ 11.7\\ 7.6\\ 6.5\\ 4.2\\ 2.9\\ 11.7\\ 7.6\\ 6.8\\ 2\\ 10.4\\ 8.7\\ 8.2\\ 10.4\\ 8.2\\ 10.4\\ 8.2\\ 10.4\\ 8.2\\ 10.4\\ 10.5\\ $	$\begin{array}{c} Lb,\\ per\\ sq.in,\\ 445\\ 591\\ 643\\ 586\\ 6573\\ 589\\ 600\\ 675\\ 1534\\ 1560\\ 770\\ 623\\ 555\\ 652\\ 652\\ 657\\ 652\\ 800\\ 589\\ 612\\ 577\\ 1631\\ 736\\ 591\\ \hline\end{array}$	$\begin{array}{c} Pct. \\ 6.2 \\ 4.5 \\ 12.8 \\ 12.8 \\ 12.9 \\ 4.2 \\ 19.0 \\ 1.6.2 \\ 5.9 \\ 1.6.2 \\ 5.9 \\ 1.6.2 \\ 5.8 \\ 1.0 \\ 1.6.5 \\ 8.1 \\ 1.1.5 \\ 8.6 \\ 4.7 \\ 1.1.5 \\ 8.6 \\ 7.1 \\ 5.6 \\ 114.9 \\ 7.8 \end{array}$	Pct.         87           81         90           86         94           95         97           98         90           73         66           82         82           90         71           37         72           68         72           50         82           60         82           60         82           60         82           60         82           60         82           60         82           60         82           60         92           76         97
	${2428 \\ 3409}$	77		421 402	*****	534	10.7	668	5.8	80
	(. 408	1		104				000	0.0	)

TABLE 6.—Curing conditions and flexural strengths of concrete

beams

Coarse aggregate

Limestone

624

7.8

79

Haydite

Curing

Age

Group

Average ...

<sup>1</sup> Values represent average of tests on 4 specimens, all other values represent average of tests on 5 specimens.

493

<sup>a</sup> Data for beams having Haydite aggregates.
<sup>a</sup> Data for beams having natural sand and crushed limestone aggregates.

was true also for groups 14, 18, 19, and 20 insofar as the amount of moisture remaining at the end of the air curing period was concerned. However, these specimens were subjected to 7 days of water curing following the air curing period. For the limestone concrete, this final immersion in water resulted in final moisture contents approximately the same as before air curing began. However, the final moisture content of the Haydite concrete specimens varied with the length of the air curing period—the shorter this period the greater the water content at the end of the final 7 days of water curing.

Data for groups 10 and 21 show for both types of concrete that the moisture loss after 7 days in air curing was approximately the same for specimens that were cured initially for 1 year in the moist room as it was for specimens cured for 6 months. On the other hand, 7 days of oven curing at 170° F. for specimens previously cured for 6 months in the moist room reduced their water contents by 43 percent, as compared with a reduction of only 33 percent for specimens cured for 1 year (groups 11 and 22, respectively). Data for groups 23 and 24 show that for specimens cured initially 1 year in the moist room, drying in the oven for 1 day drove off three or four times as much water as drying for 1 day in air.

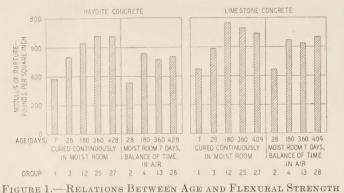
In order to determine how much water could be driven off at a temperature of 170° F., a group of both Haydite and limestone specimens were kept in the oven until they had reached constant weight. The results show (group 26) that a slightly greater percentage of

55

Strength ratio, Haydite

May 1937

CONCRETE BEAM SPECIMENS CONTAINING NATURAL SAND



OF HAYDITE AND LIMESTONE CONCRETE SPECIMENS.

water was lost by the Haydite concrete than by the limestone concrete, the percentages of water retained being 29 and 32, respectively. However, 67 days were required for the Haydite concrete to reach constant weight, as compared with 48 days for the limestone concrete.

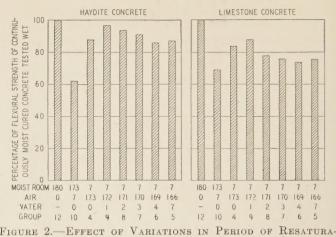
Results of the absorption tests are of interest principally as indicating the effect of type of aggregate on rate of change in moisture content resulting from drying or absorption. In general, Haydite concrete both lost and gained moisture at a slower rate than did limestone concrete. In this connection it should be remembered that the concrete made with Haydite aggregates contained 7.2 sacks of cement per cubic yard, while concrete made with natural aggregates contained but 5.7 sacks per cubic yard.

#### NONUNIFORM MOISTURE DISTRIBUTION FOUND TO AFFECT FLEXURAL STRENGTH

The results of tests for flexural strength at various ages and for the several conditions of curing investigated are given in table 6 and are shown graphically in figures 1 to 6, inclusive. The relations between flexural strength and age under continuous moist and continuous air curing conditions for both types of aggregate are shown in figure 1. It will be observed that under moist curing the Haydite concrete increased in strength up to 360 days, and specimens tested at the age of 428 days gave practically the same results as those tested at 1 year.

For limestone concrete under moist curing there was a distinct decrease in strength at 360 days as compared with strength at 180 days, with still further retrogression at 409 days. The retrogression in strength after 6 months for the limestone concrete is quite surprising in view of the fact that standard moist curing conditions obtained throughout the curing period.

In the case of specimens cured in air after 7 days of initial moist curing, we find little change in strength after 180 days for both types of aggregate. The maximum strength attained by the Haydite concrete specimens cured in air was about the same as the 28-day strength after moist curing. For the limestone concrete specimens cured in air the maximum strength was slightly higher than the 28-day strength after moist curing. Also, the 28-day strengths (after curing in air for 21 days) were about the same as the strengths of the 7-day, moist-cured specimens. In this case the normal gain in strength between 7 and 28 days was probably neutralized by the fact that specimens air dried for 21 days were only partially dry when tested; that is, while the outer shell of the specimen was fairly dry the core



TION ON FLEXURAL STRENGTH OF CONCRETE SPECIMENS AFTER CURING 180 DAYS AS INDICATED.

was still wet. The drying shell produced tensile stresses in the extreme fibers of the beam.

These initial stresses of course tended to reduce the load required to produce failure in bending, resulting in an observed modulus of rupture considerably lower than would have been obtained if the moisture content of the specimens when tested had been the same throughout. The fact that the strengths after 180 days in air curing increased materially probably resulted from the fact that during this period the specimens had an opportunity to dry uniformly, thus relieving these moisture stresses. That this so-called "shell" effect is of great importance in influencing the observed modulus of rupture will be evident from consideration of the results for specimens cured for 6 months and for 1 year, a discussion of which follows.

The results of flexure tests on specimens cured for 180 days under various curing and moisture conditions are shown in figure 2. Values in all cases have been expressed as percentages of the strength developed under continuous moist curing. It will be noted that specimens taken from the moist room at 173 days and cured in air for 7 days before testing attained only about two-thirds of the strength of specimens tested immediately upon removal from the moist room (groups 12 and 10). This decrease no doubt resulted from the partial drying of the specimens during the final 7-day period, as explained above. The effect was somewhat more marked in the Haydite concrete, the decrease in strength being 38 percent as compared with 31 percent for the lime-stone concrete.

The effect of partial drying is further illustrated by comparing the strengths of specimens cured in the moist room for 173 days, followed by 7 days in air, with those cured for 7 days in the moist room, followed by 173 days in air (groups 10 and 4). The air-cured specimens had much higher strengths, probably as a result of the absence of internal shrinkage stress in the concrete.

The effect of curing in water from 1 to 7 days just prior to testing the air-cured specimens is also shown in figure 2 (groups 5 to 9, inclusive). It will be observed that immersing the specimens for 1 day increased the strength, but that further immersion tended to lower the strength. This applied to both the Haydite and the limestone concretes. Here we have a condition where the dry specimen was absorbing water from the outside, resulting in an effect just the reverse of that previously described in which the wet specimen was

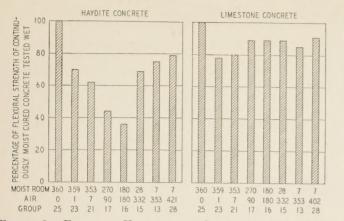


FIGURE 3.—EFFECT OF VARIATIONS IN PERIODS OF MOIST ROOM AND AIR CURING ON FLEXURAL STRENGTH OF CONCRETE SPECIMENS AFTER CURING 1 YEAR AS INDICATED.

drying. In this case the effect of moisture in the outer shell was to produce compressive stresses in the extreme fibers of the beam, thereby raising, instead of lowering, the observed modulus of rupture. That this is true is evidenced by the fact that specimens immersed for only 1 day in water had higher strengths than the specimens tested dry, but the strengths decreased as the period of immersion increased. The strength of Haydite concrete after 4 days of

The strength of Haydite concrete after 4 days of immersion was practically the same as that of the airdry specimens, indicating that moisture equilibrium had been established. For limestone concrete, the strengths after 2 days of immersion and longer were all somewhat lower than those of the air-dried specimens. For immersion periods of 3 days and longer there was little difference in strength. The results indicate that when curing has been in air, a 24-hour water immersion period prior to testing may not give true results because of incomplete moisture penetration.

#### EFFECT OF PERIOD AND TYPE OF CURING ON FLEXURAL STRENGTH STUDIED

The effect of variations in the relative amounts of moist and air curing on the flexural strength of specimens tested at the age of 1 year is shown in figure 3. It will be observed that, for Haydite concrete, increasing the period of air curing prior to test caused a progressive reduction in flexural strength up to and including 180 days of air curing. For this condition, the specimens developed only 36 percent of the strength of the continuously moist-cured concrete. However, further increasing the period of air curing up to 353 days resulted in increasing the strength progressively. An increase in the drying period up to 421 days still further increased the strength. It is believed, however, that the increase was caused more by the change in moisture condition than by the additional curing that the specimens received between 360 and 428 days.

The very peculiar behavior of the Haydite concrete here indicated is believed also to be caused by internal moisture stresses, as discussed above. There seems to be no other explanation that will cover the facts. In this connection it is interesting to note that, although the limestone concrete exhibited the same tendency, the maximum decrease in strength was not as great, and also that it occurred after only 1 day of drying. The moisture distribution appeared to be equalized, at least insofar as effect on internal stress was concerned, somewhere between 7 and 90 days of air curing, whereas in

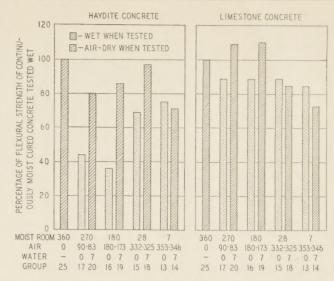


FIGURE 4.—EFFECT ON FLEXURAL STRENGTH OF RESATURA-TION FOR 7 DAYS PRIOR TO TESTING FOR CONCRETE SPECI-MENS AFTER CURING 360 DAYS AS INDICATED.

the Haydite concrete it did not occur until certainly well beyond the 180-day period.

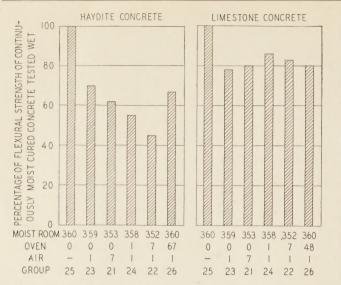
By comparing the results of the strength tests with the corresponding moisture losses at the time of test as shown in table 5 it will be noted also that the measured moisture loss appeared to bear no relation to variation in strength caused by moisture. For instance, groups 13, 15, 16, and 17 for limestone concrete showed practically the same strength although the percentage of total moisture retained varied from 67 to 85.

In order to study the effect of resaturation on the strength of the concrete cured for various periods in air, certain of the specimens cured as indicated in figure 3 were removed from air curing and immersed in water for 7 days prior to testing. The results are shown in figure 4, together with values for the corresponding airdried specimens. In studying these results it should be noted that, insofar as curing is concerned, each pair of values (for instance, groups 16 and 19) represent tests on specimens subjected, except for the final 7-day period, to identical curing conditions. Therefore, the variations noted must have been caused by some other factor, presumably the shell effect previously discussed.

#### FLEXURAL STRENGTH OF HAYDITE CONCRETE FOUND MOST SUSCEPTIBLE TO MOISTURE CHANGE

It is difficult to explain satisfactorily all of the trends observed in figure 4. This applies particularly to the results obtained on the specimens immersed in water for 7 days prior to test. This period would certainly seem sufficient to eliminate the shell effect. However, if we assume this to be true, then variations in strength of the resaturated specimens must be caused by variations in curing only. This would not explain the relative strengths shown by the Haydite concrete in groups 18, 19, and 20. It would certainly be reasonable to expect the strength to decrease with decrease in the length of the initial moist curing period, whereas, with the single exception of the specimens cured for 7 days in the moist room (group 14), just the reverse is true.

in the moist room (group 14), just the reverse is true. In the limestone concrete the relative values are as would be expected, except for the high strengths shown for groups 19 and 20. It is possible that in these groups the reabsorption of water actually produced compression





in the extreme fibers which, combined with the relatively long initial period of moist air curing, produced the exceptionally high strengths observed. In any event, it is obvious that, in the determination of the flexural strength of concrete, great care should be exercised to insure that the specimens are truly in equilibrium, insofar as internal moisture stresses are concerned, when they are tested.

In the discussion so far presented the specimens cured in air were partially dried by storing in an unheated room adjacent to the laboratory. When the outside temperature went below freezing, a window leading to the laboratory was opened so that with the exception of short periods the temperature in the curing room was kept above freezing.

Tests were made to determine the effect of rapid drying and also to determine what effect drying to constant weight would have on the flexural strength. A number of specimens were dried in an electric oven at a temperature of approximately 170° F. for periods of 1 and 7 days. Others were dried to constant weight, the time required varying for the individual specimens in each group. The 48 days noted for the limestone concrete and the 67 days noted for the Haydite concrete represent the average of five separate tests in each case. After removal from the oven the specimens were allowed to cool for 1 day in the laboratory before being tested. The results are shown in figure 5.

Referring first to the results for Haydite concrete, it will be observed that the strengths progressively decreased as the amount of drying increased up to 7 days of oven curing. Haydite specimens subjected to this treatment developed only 45 percent of the strength of moist-cured specimens of the same age, as compared with 83 percent for the limestone specimens. It will be noted also that, with the limestone concrete, the greatest decrease in strength occurred after 1 day of air drying, as against 7 days of oven and 1 day of air drying for the Haydite concrete. These results again point to the greater susceptibility of the Haydite concrete to moisture change and show also the effect of the slower drying action in concrete containing Haydite.

#### RECOMMENDATION MADE THAT FLEXURE SPECIMENS BE IMMERSED IN WATER FOR 48 HOURS BEFORE TESTING

Comparison of the flexural strengths of the Haydite and the limestone concrete specimens subjected to similar curing conditions shows that the average for all Haydite specimens tested wet was about 90 percent of the strength of the limestone concrete. On the other hand, the average strength of all Haydite specimens tested in various degrees of dryness was only about 70 percent of the strength of the limestone concrete cured and tested in like manner. This indicates that the strength of the Haydite concrete was affected to a greater degree by air curing than was the limestone concrete.

The degree to which the Haydite specimens were adversely affected varied over a wide range. For instance, the strength of Haydite specimens cured for 173 days in air after 7 days of moist curing was 86 percent of that of limestone specimens, or nearly equal to the 90 percent for the continuously moist-cured specimens. On the other hand, the Haydite specimens cured for 180 days in air after 180 days of moist curing had but 37 percent of the strength of limestone concrete similarly cured. It is apparent, therefore, that the time at which dry curing began had a marked effect on the ratio of the strength of the Haydite to the strength of the limestone concrete.

The previous discussion has been confined to the effect of the several variables on the flexural strength of the concrete tested. Additional tests were made to determine the effect on crushing strength of air curing as compared with moist curing. Tests were made at 7, 28, 180, and 360 days, on specimens cured the entire time in the moist room and on specimens cured in air after an initial moist curing of 7 days. No tests were made on specimens cured dry and tested wet or on specimens cured wet and tested dry as was done with the beam specimens. In other words, all of the compression specimens were probably in a stabilized moisture condition when tested.

The results for the Haydite concrete (fig. 6) indicate a progressive increase in strength for moist curing up to 180 days, with a small increase over the 180-day strength at 360 days. The same general trend is found for the specimens cured in air; that is, the strength at 360 days was approximately the same as that at 180 days. At each curing period the air-cured specimens had lower strengths than the moist-cured specimens, the difference becoming greater as the age at test increased. The results are in agreement with those found in the flexure tests; that is, the specimens cured in air and tested dry had lower strengths than those cured in moist air and tested wet.

The right-hand panel of figure 6 gives similar results for the limestone concrete. The results obtained are in general agreement with those found for the Haydite concrete; therefore they will not be discussed in detail. They are also in agreement with the flexure test results where the specimens were subjected to like curing treatments.

These tests indicate that the flexural strength of concrete containing aggregates of the type represented by Haydite is affected considerably more by variations in curing and moisture content at time of test than is concrete containing naturally occurring aggregates of average quality.

(Continued on p. 61)

# THE OLIENSIS SPOT TEST IMPROVED

Reported by R. H. LEWIS, Associate Chemist, and J. Y. WELBORN, Junior Highway Engineer, Division of Tests, Bureau of Public Roads

CINCE the publication in 1933 of Mr. Oliensis' paper "A Qualitative Test for Determining the Degree of Heterogeneity of Asphalts",1 the Bureau of Public Roads has devoted considerable time to the study of this test, which has proven to be a very useful one. The results of these studies corroborate the author's conclusions with respect to the scope and limitations of the test.

In tests by the Bureau on a large number of asphaltic materials from all producing centers, heterogeneity is found to be far more prevalent among the more fluid materials than among the refined, semisolid, paving asphalts, and the quantitative method recently de-veloped by Mr. Oliensis should be of great value in comparing the degree of heterogeneity of various asphaltic materials. The paper<sup>2</sup> describing this new method of test has been studied and the data obtained are presented in this report.

Clifford Richardson has stated: <sup>3</sup>

In residual pitches, at times some of the bitumen is found which is insoluble in cold carbon tetrachloride, and this is evidently due to the severe treatment which the material has suffered in the course of its production at very high tempera-tures. A determination of the amount is only valuable as an indication of the care which has been used in the preparation of such pitches.

For years, therefore, asphalt specifications, have required a high solubility in carbon tetrachloride as a protection against overheating in the refining process.

However, examination of fluid and semisolid asphalts from numerous sources has shown that many of the products that have been subjected to temperatures much higher than those used in normal steam refining have relatively high solubility in carbon tetrachloride. Many materials that are definitely cracking-coil residues produce a positive stain with both naphtha and xylene, while the majority of those that apparently have been inadvertently overheated in a steam or vacuum process, and some blends of cracked and uncracked residuals, give a positive stain with naphtha and a negative stain with xylene. In spite of their high solubility in carbon tetrachloride, these asphaltic materials are heterogeneous. It is apparent, therefore, as has been pointed out by Mr. Oliensis, that a high solubility in carbon tetrachloride does not indicate definitely that the material has not been overheated in the refining process.

The Bureau is now engaged in a detailed investigation of commercial grades of semisolid asphalts from the refineries of the major producers. This study involves among other things the exposure of the asphalts in films one-eighth inch thick to the action of sun and light for 15 weeks (approximately 900 sunlight hours).

Thirty-nine samples of 85-100 penetration asphalt have been studied, of which 16 samples gave negative stains both before and after exposure. Nine gave negative stains before exposure and showed varying degrees of heterogeneity after exposure. Twelve gave positive stains before and after exposure; and two materials that showed slight heterogeneity before exposure produced negative stains after 15 weeks of exposure. Perhaps these two are examples of the fugitive type which has been observed by Mr. Oliensis.

For the 23 samples that gave positive stains either before or after exposure, table 1 gives the base petroleum and the refining process as reported by the manufacturers. Results of the Oliensis spot test before and after exposure, the gilsonite and xylene equivalents before exposure, and the xylene equivalent after exposure are also listed.

#### METHOD OF DETERMINING GILSONITE AND XYLENE EQUIVALENTS DESCRIBED

In considering the data in table 1, and later in table 2, the heterogeneity of the materials is expressed quantitatively by limits representing the highest percentage of gilsonite or xvlene that would produce a positive stain and the lowest percentage of gilsonite or xylene that would produce a negative stain. The true xylene or gilsonite equivalent (the exact percentage of homogeniser necessary to produce a negative stain) is within these limits. Thus, a xylene equivalent of 12-16 means that with 12 percent or less of xylene in the standard naphtha the material developed a positive stain and that with 16 percent or more of xylene, a negative stain was obtained. A gilsonite equivalent of 50-60 means that when 50 percent of gilsonite was blended with the material under test and the blend tested with standard naphtha a positive stain was produced; when the blend of gilsonite and asphaltic material contained 60 percent of gilsonite a negative stain was obtained. The term "xylene insoluble" indicates that a positive stain was obtained with 100 percent of xylene and the material is beyond the range of the test, i. e., the heterogeneity of the material is indeterminable with this solvent.

Table 1 shows that only 2 of the 39 materials produced a positive stain with xylene, either before or after exposure. Eight of the 14 original materials that were heterogeneous had xylene equivalents of 8 or less, and gilsonite equivalents of 10 or less. Of the heterogeneous exposure residues of the nine materials that were originally homogeneous, seven had xylene equivalents of eight or less. For those materials that were heterogeneous originally, the increase in xylene equivalent in the exposed residues was quite variable.

Samples 7 and 8 are of special interest. Both were produced from the same Venezuelan crude, the first by the continuous vacuum steam distillation and the latter by batch steam distillation. After exposure sample 8, which gave a positive stain initially, had a higher xylene equivalent than sample 7, which gave a negative stain initially.

In table 1 of Mr. Oliensis' report,<sup>2</sup> he shows that the xylene equivalent for sample 13, representing a waxbearing residual, crude K, was greater than the gilsonite equivalent. Sample 23 of the materials tested by the Bureau was a wax-bearing residual, produced from a crude that tested positive in the spot test. Tests made on other semisolid products from this same refinery and on residues reduced in the laboratory with a high

<sup>&</sup>lt;sup>1</sup> Proceedings, American Society for Testing Materials, vol. 33, pt. II. <sup>2</sup> A Further Study of the Heterogeneity of Asphalt—A Quantitative Method, by G. L. Oliensis. Proceedings, American Society for Testing Materials, vol. 36, pt. II. <sup>3</sup> The Modern Asphalt Pavement, by Clifford Richardson. John Wiley & Sons, New York, 1905, 1 ed., p. 120.

Tests after exposure for 15 weeks in ½-inch film Tests before exposure Sam Base petroleum ple no. Refining process Standard naphtha stain Gilsonite equiva-lent Xylene equiva-lent Xylene equiva-lent Standard naphtha stain California Steam distillation Negative .... Positive .....  $\begin{array}{c} 0-2\\ 0-2\\ 12-16\\ 12-16\\ 28-32\\ 12-16\\ 12-16\\ 20-24\\ 12-16\\ 0-2\\ 0-2\\ 0-2\end{array}$ Vacuum distillation... Steam distillation... Vacuum distillation... -\_do\_\_ \_\_\_\_do\_\_\_\_ \_\_\_\_do\_\_\_\_ 0-1 50-60 0-224-28 2-4 Positive. \_\_\_\_do\_\_\_\_ enezuelan. Continuous vacuum steam distillation Batch steam distillation Steam distillation 5-8 Negative\_ Positive\_ 2-4 2-4 5-10 ---do. do Vacuum distillation. Negative. Positive. 0-1 Vacuum distillation  $\begin{array}{c}
0-2 \\
44-48 \\
2-4 \\
4-8 \\
(1)
\end{array}$  $\begin{array}{r}
 0-1 \\
 30-35 \\
 2-4 \\
 5-10
 \end{array}$ 72-76 Positive\_. Oklahoma. \_\_\_\_do\_\_\_\_\_ 8-12 (1) 4-8 do \_do..... do\_ Negative.... 60-80 do... Blend of Mexican and Oklahoma. Steam distillation Winkler-Kock shell still. 80-90 Positive ... Kansas\_\_\_\_\_ do... Steam distillation \_\_\_\_do\_\_\_\_\_ 15-25 0-2 12-16 0-2 ...do.... Negative. 32-36  $\frac{20}{21}$ Blend of Mexican and domestic Gulf coast 0-2 Negative. Positive crudes.  $\frac{22}{23}$ Positive... 2<del>-4</del> 36-40 \_\_\_do\_\_ Texas\_\_ 35-40 16-20 do...

TABLE 1.—Oliensis spot test results and gilsonite and xylene equivalents for samples of 85-100 penetration asphalt

<sup>†</sup> Xylene insoluble.

steam ratio from the crude oils used as base petroleum by this refinery, showed that they required approximately 35 percent of gilsonite to produce a negative stain. Although the other samples from this source were not tested for xylene equivalent, it is noted that the xylene equivalent for sample 23 was considerably lower than the gilsonite equivalent. This is not in agreement with the conclusion by Mr. Oliensis that gilsonite is more efficient than xylene in correcting heterogeneity caused by waxy bodies.

The spot test results for these representative asphalts show that 23 out of 39 samples, or about 60 percent, were homogeneous, and that 16, or about 40 percent, remained homogeneous after 15 weeks of exposure. All of the homogeneous materials that became heterogeneous upon exposure had xylene equivalents of 16 or less. Under the climatic conditions existing during the winter months at Madison, Ill., it is possible that heterogeneity might not have developed in these samples.

#### NEW METHOD USEFUL IN TESTING BOTH ORIGINAL MATERIALS AND RESIDUES FROM EXPOSURE

Under the same summer conditions to which the 85– 100 penetration asphalts were subjected, all fluid materials that were studied in this investigation developed heterogeneous residues within 5 weeks. In a previous report covering a study of the spot test on liquid asphalts, mentioned in Mr Oliensis' paper, the following statement was made: <sup>4</sup>

Since the results of the tests were based upon the appearance of the stain as interpreted by the observer, it is difficult if not impossible to distinguish between border-line materials or to express clearly the apparent degree of heterogeneity that may be indicated by the varying degrees of nonuniformity in the stain. The classification given should be understood to mean that, in the judgment of the observers, the materials and their residues gave stains that appeared either entirely uniform throughout or were only slightly nonuniform, having a slightly darker, more pronounced center, or else they had a definite dark to black center surrounded by a uniformly lighter colored stain, and were consequently classified as homogeneous, slightly heterogeneous, and heterogeneous respectively. The use of the xylene equivalent gives promise of overcoming past difficulties in the proper interpretation of the spot test results and appears to offer a ready means of determining the comparative degree of heterogeneity and the rate of its development in materials exposed to weathering.

In a report presented at the January 1936 meeting of the Association of Asphalt Paving Technologists <sup>5</sup> the behavior of five materials—three steam-distilled residuals (Mexican, midcontinent, and California), and two cracking coil residuals under exposure conditions was described. The results of the spot tests, determined on these materials after exposure in films 1/8, 1/16, and 1/32 inch thick for 5, 10, and 15 weeks, may help to show the possibilities of the quantitative method of determining heterogeneity suggested by Mr. Oliensis.

The two cracking coil residues gave positive spots with xylene, and the development of heterogeneity in these materials under exposure could only be detected by the steady development of organic matter insoluble in  $CS_2$  and  $CCl_4$ . The other three materials and their distillation residues gave negative stains. The xylene equivalents of the residues after various periods of exposure are shown in table 2.

It will be seen that, at each test period, the thinner the film thickness the greater the xylene equivalent. For each thickness of film, except the  $\frac{1}{4}$ - and  $\frac{1}{46}$ -inch films of the Mexican residual, the xylene equivalent increased with the time of exposure. The reason that the  $\frac{1}{4}$ - and  $\frac{1}{46}$ -inch films of the Mexican residual after 10 weeks of exposure had lower xylene equivalents than the corresponding films after 5 weeks of exposure is not known.

Data on the solubilities of some of these residues are of interest (table 3). It will be seen that the  $\frac{1}{22}$ -inch film of sample 2 after 15 weeks of exposure had more matter insoluble in CS<sub>2</sub> than two of the other residues, and more matter insoluble in CCl<sub>4</sub> than three of the other residues. However, it was readily dispersed in 68-72 percent xylene-naphtha solution, while the other samples all contained material insoluble in 100 percent xylene. This would seem to indicate a difference in the

<sup>&</sup>lt;sup>4</sup> Further Studies of Liquid Asphaltic Road Materials, by R. H. Lewis and W. O'B. Hillman. PUBLIC ROADS, vol. 16, no. 6, August 1935, p. 108.

<sup>&</sup>lt;sup>5</sup> A Report on the Weather-Resistant Properties of Certain Slow-Curing Liquid Asphaltic Materials. Proceedings of the Association of Asphalt Paving Technologists, Jan. 23, 1936.

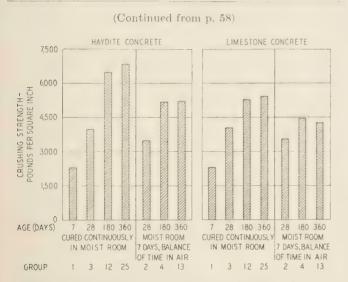
TABLE 2.—Xylene equivalents of residues of slow-curing liquid asphaltic materials (grade SC-2) after exposure

	Xylene equivalent of residues of										
Samples tested after exposure for—	Sample 1, 1	Mexican resid hickness of—	iual. Film	Sample 2, Fil	midcontinen m thickness (	t residual. of—	Sample 3, California residual. Film thickness of-				
	⅓-inch	以6-inch	1⁄32-inch	1⁄8-inch	1/16-inch	½₂-inch	!≴-inch	ŀí₀-inch	Já₂-inch		
5 weeks 10 weeks 15 weeks	4-8 0-2 8-16	8-12 4-8 (1)	12-16 (1) (1)	0-2 16-20 36-40	28-32 36-40 56-60	48-52 60-64 68-72	24–28 32–36 36–40	$36-40 \\ 44-48 \\ 48-52$	48-52 64-68 (1)		

<sup>1</sup> Xylene insoluble.

character of the insoluble materials in the several residues that affects their dispersion in the solvents used for determining heterogeneity.

The stain obtained in the original Oliensis spot test indicates only that the material is homogeneous or heterogeneous. The degree of heterogeneity is not determinable by this test and, as indicated in table 1, it automatically classes materials that have been only slightly overheated with products that have been subjected to excessive heat. This quantitative method, therefore, should prove extremely valuable in future studies of asphaltic materials.



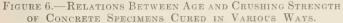


 TABLE 3.— Comparison of solubilities of exposed samples with xylene equivalents

Sample Do.	Time of exposure	Film thickness	Total or- ganic mat- ter insol- uble in CS <sub>2</sub>	Total or- ganic mat- ter insol- uble in CC1 <sub>4</sub>	Xylene equivalent
1 1 2 3	Weeks 10 15 15 15 15 15	Inch 1/32 1/10 1/32 1/32 1/32 1/32 1/32	Percent 1.88 1.42 2.88 1.65 1.21	Percent 1, 98 1, 67 3, 22 2, 06 1, 40	Insoluble. Do. Do. 68-72. Insoluble.

The wide variations in flexural strength of the Haydite concrete appear to result primarily from its tendency to absorb and to give off moisture at a slower rate than concrete containing normal aggregates, thus delaying the even distribution of moisture which is necessary in order to eliminate internal stresses.

The results emphasize the importance of controlling testing procedure so as to insure that specimens under test are free from stresses induced by an uneven distribution of moisture. They indicate that, where curing has been in air, immersion in water prior to test for more than the customary 24 hours may be necessary and that, when the specimens have been subjected to continuous moist curing until test, care should be exercised to prevent any drying prior to test.

Drying to constant weight in order to eliminate moisture stresses is not practicable because of the length of time required. For this reason it is recommended that all flexure testing be performed on saturated specimens, and that in cases where specimens have been subjected to drying prior to test they be immersed in water for at least 48 hours immediately before testing.

### SLIPPERINESS OF ROAD SURFACES STUDIED BY ENGLISH INVESTIGATORS

Apparatus for accurately measuring the slipperiness of various road surfaces has been developed at the Road Research Laboratory of the Department of Scientific and Industrial Research in England. The machine used consists of a motorcycle and sidecar in which the sidecar wheel can be set at an angle to the direction of travel.

The horizontal reaction of the road, developed by the wheel because of its position, and the load on the wheel. are measured. From data obtained with instruments carried in the sidecar, the ratio of these forces is obtained and expressed as a coefficient which is high for nonskid surfaces and low for slippery ones. The construction and operation of the motorcycle and sidecar and of the measuring apparatus are described in a bulletin<sup>1</sup> recently issued jointly by the Department of Scientific and Industrial Research and the Ministry of Transport. The apparatus, which has been in regular use for several years, is estimated to cost £475 (approximately \$2,350).

A bulletin<sup>2</sup> reporting results of the first of a series of studies of road friction has also been published recently. This bulletin summarizes the results of numerous measurements made on various types of road surfaces under different climatic conditions and at different speeds. Deductions and conclusions are drawn from the results obtained.

The skidding coefficient was found to be high for dry road surfaces free from loose material, and such surfaces can be considered as nonskid. On wet surfaces, the coefficient decreases as the speed increases. A value of 0.5 at 30 miles per hour may be regarded as reasonably safe; a value of 0.2 indicates that the surface affords insufficient resistance to skidding. Values of 0.2 or less are found most frequently on city streets.

The most slippery condition of a road is probably during a silver thaw, when the coefficient is as low as 0.1 at all speeds. Measurements taken on a snowcovered road showed values ranging from 0.5 to 0.2, the latter being found after the snow had become packed. The values for snow-covered roads were no lower than those found on some smooth, wet surfaces, but on snow-covered roads the lower values may persist down to speeds of 5 miles per hour.

On concrete surfaces the coefficient varied but little at different seasons of the year. Practically all other surfaces, when wet, were more slippery in summer than in winter. Tests made on wood- and rubber-block pavements showed them to have low coefficients.

The apparatus is now being improved to enable tests to be made at speeds higher than 30 miles per hour.

Road Research Bulletin No. 1, Measurement of the Non-Skid Properties of Road Surfaces, H. M. Stationery Office, 9d net.
 Road Research Technical Paper No. 1, Studies in Road Friction. 1, Road Surface Resistance to Skidding. Published by H. M. Stationery Office, 1s. 6d. net.

# PUBLIC ROADS

# MOTOR-FUEL CONSUMPTION, 1936

[Compiled for calendar year from reports of State authorities]

		ate per on—			Amoun	nts exem ment c		n pay-						unt taxed uced rates		iate amour ighway us	
State	On Jan. 1	On Dec. 31	Date of rate change	Net total consump- tion in State <sup>1</sup>	Federal and other public use <sup>3</sup>	Non- high- way use 4	Allow- ance for evapo- ration and other losses	Total	Gross <sup>6</sup> amount assessed for taxation	Amount subject to refund of entire tax	Net amount taxed	Amount taxed at full rate	Rate per gallon	Amount	1936	1935	Per- cent- age change
Alabama Arizona Arkansas California Conrecticut Delaware Florida Georgia Idaho Illinois Indiana Jowa Kansas Kentucky Louisiana Maine Maryland Massachusetts Minnesota Minnesota Minnesota Mississippi. Missouri Montana Nevada Nevada Nevada New Hampshire. New Jersey New Horska New Horska New Hampshire. New York North Carolina Orth Carolina Orth Carolina Orth Carolina North Carolina North Carolina North Carolina North Carolina North Carolina South Carolina Vermont Vermont Vermont Vermont	7653433554483362554483546	6 5	July 28	$\begin{array}{c} 1,000\\ gllons\\ gllons\\ 190,040\\ 96,543\\ 164,208,897\\ 2208,897\\ 2209,405\\ 50,766\\ 300,192\\ 303,968\\ 87,848\\ 1,191,915\\ 564,073\\ 460,298\\ 450,331\\ 228,333\\ 223,093\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 245,231\\ 134,521\\ 256,377\\ 101,645\\ 315,372\\ 103,667\\ 123,298\\ 103,667\\ 123,298\\ 103,667\\ 141,6043\\ 326,315\\ 1,008,063\\ 103,667\\ 141,6043\\ 266,315\\ 1,108,063\\ 83,358\\ 60,026\\ 316,556\\ 109,302\\ 316,557\\ 316,557\\ 1,008,063\\ 100,026\\ 316,556\\ 310,302$	3, 853 1, 412 3, 731 4, 971 3, 768 4, 496 3, 813 1, 274 1, 633 17 12, 193 4, 622	615 1, 993 478 10124,886 49, 925 1, 881 1, 189 4, 010 638	1,000 gallons 1,460 16,240 4,145 2,957 2,478 1,486 9,007 9,007 6,693 28,027 15,510 3,305 2,116 7,085 601 1,165 2,315 33,985 7,222 4,563 2,528 10,956 6,2,417	$\begin{array}{c} 44, 064\\ 13, 948\\ 5, 981\\ 1, 024\\ 12, 113\\ 5, 295\\ 4, 809\\ \hline \\ 2, 364\\ \hline \\ 133, 893\\ \hline \\ 10, 546\\ 1, 412\\ 3, 731\\ \hline \\ 82, 923\\ 21, 159\\ 7, 801\\ \hline \\ 1, 412\\ 3, 731\\ \hline \\ 82, 923\\ 21, 159\\ 7, 801\\ \hline \\ 1, 412\\ 3, 731\\ \hline \\ 82, 923\\ 21, 159\\ 7, 801\\ \hline \\ 8, 546\\ 8, 923\\ 3, 423\\ \hline \\ 7, 801\\ 6, 035\\ 5, 78, 76\\ 6, 044\\ 10, 888\\ 4, 663\\ 6, 055\\ 1, 009\\ 2, 3411\\ 6, 533\\ 13, 650\\ 22, 409\\ \end{array}$	$\begin{array}{c} 1,000\\ gallons\\ 194,040\\ 91,068\\ 148,276\\ 1,607,773\\ 194,949\\ 293,424\\ 49,742\\ 288,079\\ 298,673\\ 83,039\\ 1,191,915\\ 561,709\\ 460,298\\ 333\\ 212,547\\ 135,109\\ 460,298\\ 316,438\\ 228,333\\ 212,547\\ 133,109\\ 241,500\\ 654,309\\ 912,658\\ 462,658\\ 162,144\\ 567,750\\ 192,268\\ 163,133\\ 105,814\\ 208,132\\ 208,132\\ 208,133\\ 109,102\\ 208,133\\ 109,102\\ 211,915\\ 1,277,225\\ 122,289\\ 113,354\\ 849\\ 211,915\\ 1,277,225\\ 122,289\\ 113,354\\ 849\\ 211,915\\ 1,277,225\\ 122,289\\ 113,354\\ 849\\ 211,915\\ 1,277,225\\ 122,289\\ 108,133\\ 354,849\\ 211,915\\ 1,277,225\\ 122,289\\ 108,133\\ 351\\ 109,510\\ 222,2665\\ 1,084,654\\ 78,183\\ 57,697\\ 1316,556\\ 313,379\\ \end{array}$	8, 135 68, 583 35, 800 48, 754		79, 509 133, 844 1, 459, 993 172, 661 287, 218 287, 218 287, 218 287, 218 287, 218 287, 218 287, 218 287, 218 288, 079 298, 673 274, 701 1, 123, 332 228, 333 11 212, 547 127, 513 224, 319 624, 930 872, 976 404, 426 1511, 880 872, 976 404, 426 1511, 880 872, 976 404, 426 1511, 880 872, 976 404, 426 1511, 880 872, 976 404, 426 1548, 211 90, 447 1548, 9546 414, 1619, 122 222, 243 156, 082 22, 243 186, 092	(*) 	14, 392 <sup>9</sup> 203 <sup>12</sup> 5, 596 <sup>13</sup> 2, 217 ( <sup>14</sup> ) <sup>16</sup> 10, 264 <sup>16</sup> 8, 580 <sup>20</sup> 45, 098 <sup>21</sup> 773	$\begin{array}{c} 1,000\\ gallon_{28}\\ *199,040\\ (199,509\\ *148,276\\ 1,459,993\\ 172,661\\ 287,218\\ 47,067\\ *298,673\\ 74,701\\ 1,123,332\\ 525,998,673\\ 74,701\\ 1,123,332\\ 525,998,673\\ 74,701\\ 1,123,332\\ 47,01\\ 1,123,332\\ 47,01\\ 1,123,332\\ 47,01\\ 1,123,332\\ 47,01\\ 1,123,332\\ 47,01\\ 1,154\\ 438,47\\ 1,27,513\\ 226,536\\ 624,930\\ 151,880\\ 151,880\\ $	$\begin{array}{c} 1,000\\ gallons\\ *172,474\\ 67,333\\ *131,784\\ 1,340,137\\ 152,324\\ 268,781\\ 42,948\\ *256,609\\ *264,617\\ 63,743\\ 1,015,019\\ 42,948\\ *256,609\\ *264,617\\ 472,010\\ 386,489\\ *201,324\\ *204,324\\ 204,830\\ *205,308\\ *201,324\\ 204,830\\ *205,308\\ *201,324\\ 204,830\\ 574,937\\ 374,701\\ 114,532\\ 204,830\\ 574,937\\ 374,701\\ 123,201\\ 498,350\\ 77,338\\ *19,165\\ 225,48,323\\ 554,432\\ 554,432\\ 554,432\\ 57,987\\ 74,902\\ 90,959\\ 160,434\\ *1,171,499\\ 905,524\\ 60,966\\ 51,388\\ 272,169\\ 252,601\\ 9252,601\\ 752,200\\ 75$	$\begin{array}{c} 18,1\\ 12,59\\ 8,99\\ 9,6\\ 12,33\\ 8,99\\ 9,6\\ 12,29\\ 110,27\\ 111,45\\ 12,9\\ 110,27\\ 111,45\\ 11,39\\ 10,60\\ 13,55\\ 7,9\\ 23,22\\ 100,0\\ 16,80\\ 14,0\\ 10,7\\ 14,0\\ 10,7\\ 12,3\\ 10,60\\ 16,8\\ 12,4\\ 11,79\\ 16,0\\ 9,4\\ 11,79\\ 16,0\\ 9,7,8\\ 12,4\\ 11,79\\ 16,0\\$
West Virginia Wisconsin Wyoming Dist. Columbia_	4 4 4 2	4 4 4 2		$     \begin{array}{r}       315,302 \\       181,039 \\       503,969 \\       58,044 \\       126,837 \\     \end{array} $	27 457 3, 161 1, 748 29 4, 299		12, 521 293	457     15, 682     1, 748	$\begin{array}{c} 516, 748\\ 180, 582\\ 488, 287\\ 56, 296\\ 122, 245\end{array}$	25, 518 5, 952 35, 778 620	$\begin{array}{c} 268, 221 \\ 174, 630 \\ 452, 509 \\ 56, 296 \\ 121, 625 \end{array}$	$\begin{array}{c} 263, 221\\ 174, 630\\ 452, 509\\ 56, 296\\ 121, 625\end{array}$		• • • • • • • • •	174, 630 452, 509 28 *55, 482 121, 625	153, 105 405, 909 *47, 445 111, 983	$     14.1 \\     11.5 \\     16.9 $
Total		ghted te 3.85	average cents.	19, 653, 142	247, 037	215, 508	179, 075	641, 620	19, 011, 522	933, 996	18, 077, 526	17, 979, 340		98, 186	17, 993, 077	16, 264, 077	10. 6

<sup>1</sup> Export sales and other amounts not representing consumption in State have been eliminated as far as possible. Exemptions and refunds in a few States include small amounts of such deductible items not reported separately. In cases where States failed to report amounts exempted from taxation the gross amount taxed is shown in this column.
<sup>3</sup> In cases where refunds were made for Federal and other public uses, or for losses, the amounts have been included with the exemptions listed below, rather than with the refund gallonage.
<sup>3</sup> Federal use except as otherwise noted.
<sup>4</sup> In Kansas exemptions rather than refunds are made for all nonhighway uses. Other States, as shown, make both exemptions and refunds. Florida exempts motor fuel used in aviation but not other nonhighway uses.
<sup>3</sup> Is dates do not provide for exemptions or refunds for nonhighway use. The amounts entered for these States, indicated by stars, include both highway and nonhighway uses.

amounts entered for these states, indicated by state of adjacent State. Gallons <sup>6</sup> Within 300 feet of border tax rate is reduced to that of adjacent State. Gallons taxed at 2 cents, 2,739,000; at 4 cents, 11,534,000; at 5 cents, 119,000. <sup>7</sup> Federal use, 5,415,000 gallons; State use, 610,000 gallons. <sup>8</sup> Federal use, 414,000 gallons; State use, 610,000 gallons. <sup>9</sup> Motor fuel used in aviation. <sup>10</sup> Includes 113,859,000 gallons taxed at 5 cents per gallon and 98,688,000 gallons taxed et 7 cents per gallon. at 7 cents per gallon. <sup>12</sup> 3 cents per gallon refunded on nonhighway uses.

<sup>13</sup> I cent per gallon refunded on motor fuel used in vehicles licensed to operate exclusively in cities.
<sup>14</sup> 1½ cents per gallon refunded on motor fuel used in interstate aviation. Amount not reported.
<sup>15</sup> Excludes 1,464,000 gallons of aviation fuel taxed at 3 cents per gallon.
<sup>16</sup> 5 cents per gallon refunded on nonhighway uses.
<sup>17</sup> Federal use, 3,35,000 gallons; State, county, and municipal use, 47,801,000 gallons.
<sup>18</sup> Federal use, 11,880,000 gallons; State, county, and municipal use, 47,801,000 gallons.

lons.  $^{19}$  Includes 705,520,000 gallons taxed at 4 cents per gallon and 913,602,000 gallons taxed

<sup>19</sup> Includes 705,520,000 gallons taxed at 4 cents per gallon and 913,602,000 gallons taxed at 3 cents per gallon.
 <sup>20</sup> Does not include 69,610,000 gallons of liquid fuel (kerosene, fuel oil, etc.) taxed at 1 cent per gallon but not subject to the 3-cent tax on motor-vehicle fuel.
 <sup>21</sup> 4 cents per gallon refunded on motor fuel used in aviation.
 <sup>22</sup> Excludes 323,000 gallons of aviation fuel.

<sup>22</sup> Excludes 323,000 gallons of aviation fuel.
<sup>23</sup> 2 cents per gallon refunded on nonhighway uses.
<sup>24</sup> Revised figure.
<sup>25</sup> Federal use, 7,634,000 gallons; county and municipal use, 3,488,000 gallons.
<sup>26</sup> Excludes 271,000 gallons of aviation fuel.
<sup>27</sup> Public use not reported in full; amount includes 32,000 gallons Federal use and 425,000 gallons. State, county, and municipal use, on which refunds were paid.
<sup>28</sup> Excludes 814,000 gallons of aviation fuel.
<sup>29</sup> Includes 814,000 gallons of aviation fuel.
<sup>29</sup> Includes 814,000 gallons of aviation fuel.

# STATE MOTOR-FUEL TAX RECEIPTS, 1936

[Compiled for calendar year from reports of State authorities]

	Tax ra galle	nte per on—	Date of	Receipts from	n taxation	of motor fuel	Other r	eceipts in com	nnection w	ith motor-f	uel tax 1	Net total
State	On Jan. 1	On Dec.	rate change	Gross receipts	Refunds paid	Net receipts	Distribu- tors' and dealers' licenses	Inspection fees <sup>2</sup>	Fines and penalties	Miscel- laneous receipts <sup>3</sup>	Total	receipts
					1,000				1,000	1,000		
	Cents	Cents		1,000 dollars	dollars	1,000 dollars	1,000 dollars	1,000 dollars	dollars	dellars	1,000 dollars	1,000 dollar
Alabama Arizona	65	6 5		11,754 4,483	642	11,754 3,841	*	49			49 2	11,80 3,84
Arkansas	614	616		9,155	042	9, 155		80	4		80	9, 23
California	3	3		47, 429	4, 433	42,996	12				12	43,00
Colorado	4	4		7,855	1,022	6, 833						6, 83
Connecticut	3	3		8,967	185	8,782	53				53	8, 83
Delaware	4	4 7		1,962	109	1,853	33	359	*		3 392	1,85
Florida Georgia	6	6		19, 925 17, 493		19,925 17,493	33	998			392	20, 31 17, 49
daho		5		4, 094	402	3, 692	1		*	3	4	3, 69
llinois	3	3		35, 398	1,940	33, 458		359	2		361	33, 81
ndiana	4	4		22, 127	1,432	20, 695	*	458		1	459	21, 15
owa	3	3		13, 585	1, 389	12, 196						12, 19
Kansas Kentucky	35	3		9.372 11.273		9, 372 11, 273	6	100	3	42 1	148	9, 52 11, 27
Jouisiana	5	7	July 28	12, 121		12, 121		86	*	1	86	12, 20
Jaine	4	4	5 diy 20	5, 370	168	5, 202	*					5, 20
Iarvland	4	4		9,542	621	8,921						8, 92
fassachusetts	3	3		19, 348	900	18, 448						18, 44
1ichigan	3	3		26, 925	I, 190	25, 735	3		1		4	25, 73
Minnesota	3	3		13,994 9.577	1, 861 515	12, 133 9, 062	* 21	194	1	(i)	196	12, 32
Mississippi Missouri	2	2		11, 360	288	11.072		116		(.)	116	9, 00 11. I
Montana	5	5		5, 249	794	4, 455		110			110	4, 45
Nebraska	5	5		11, 443	225	11, 218		102			102	11, 32
Jevada	4	4		1, 166	86	1,080	*					1, 08
New Hampshire	4	4		3, 266	86	3, 180				1	1	3, 18
New Jersey	3 5	3 5		21,304 3,715	2, 217 326	19,087 3,389	16 21		*	3	19 21	19, 10 3, 41
New Mexico	4	2	July 1	57, 221	1, 587	55, 634	59			16	75	55, 70
North Carolina	6	6	July 1	20, 423	429	19,994	00	962		5	967	20, 96
orth Dakota	3	3		3, 088	843	2, 245	*	56			56	2, 30
)hio	4	4		5 45, 198	1,748	43, 450	* .					43, 43
klahoma	4	4		14,094	883	13, 211			5		5	13, 21
)regon	5 4	5		10,492	1, 284	9,208 49,364			$1 \\ 12$	9 7	10 19	9, 21
Pennsylvania Rhode Island	4 2	4		49,364 2,396	170	49, 364 2, 226	4		12	1	19	49, 38
outh Carolina	6	6		9,650	155	9,495		200			200	9,69
outh Dakota	4	4		4, 320	252	4,068		118			118	4, 18
ennessee	7	7		17, 177		17, 177		981			981	18, 15
'exas	4	4		43, 495	5,028	38, 467				4	4	38, 47
/tah/ermont	4	4		3,087		3,087 2,277	1				1	3,08
/ermont/irginia/	4	4		2,277 15,576	873	14, 703	* ****		11		11	2, 27 14, 71
Vashington	5	5		15, 612	1, 276	14, 705	*		11	9	9	14, 71
Vest Virginia	4	4		7,067	264	6, 803	7				7	6, 81
Visconsin	4	4		19, 299	1,468	17, 831		197			197	18, 02
Vyoming	4	4		2, 252		2, 252	2				2	2, 25
District of Columbia	2	2		2, 395	13	2, 382	11				11	2, 39
Total	{Weighted cents.	average	rate 3.85	723, 735	37, 104	686, 631	233	4, 417	38	101	4, 789	691, 42

Stars indicate amounts less than \$500.
 Fees for inspection of motor-vehicle fuel. Wherever possible fees for inspection of kerosene and other non-motor-vehicle fuels have been eliminated.
 Includes fees for motor-fuel carrier permits, refund or exemption permits, interest on deposits, and miscellaneous unclassified receipts.
 A special tax of 3 cents per gallon in Hancock County and 2 cents per gallon in Harrison County is imposed for sea-wall protection. The receipts from these taxes were \$130,000 in 1936. These receipts are distributed back to the respective counties.

<sup>5</sup> Ohio imposes a 3-cent tax on motor-vehicle fuel and a 1-cent tax on all liquid fuels. The receipts'from the 1-cent tax applicable to non-motor-vehicle fuels (kerosene, fuel oil, etc.) were \$689,000. These receipts have been eliminated from the total given, which represents a 4-cent tax on motor-vehicle fuel.

				AS OF AI	APRIL 30, 1937	37					
	1		COMPLETED		UNDER	ER CONSTRUCTION		APPROVE	APPROVED FOR CONSTRUCTION	NO	RALANCE OF
STATE	APPORTIONMENT	Estimated Total Cost	Federal Aid	Miles	Estimated Total Cost		Miles	Estimated Total Cost	Federal Aid	Miles	PROJECTS
Alabama Arizona Arizona	# 7.872.980 5.394.661 6.463.681	\$ 51,600 2,220,645	\$ 25,800 1,717,594	9.0 124.3	# 1,381,601 1,434,196 2,424,142	\$ 690,800 1,084,980 2,421,562	58.5 52.5 148.1	\$ 345,500 404,531 1.310.022	\$ 172,750 279,269 1,309,353	12.1 7.5 81.5	# 6,983,630 2,312,818 2,732,766
California Colorado Connecticut	14.366.891 6.911.198 2.388.339	6,264,184 3,972,008 771,224	3.560.738 2.109.120 385.224	162.2 160.1 14.2	9,318,237 3,581,611 745,158	5,172,708 1,995,118 370,176	228.3 129.3 8.7	1,298,097 1413,032	729.549	21.8	4,903,896 2,559,948
Delaware Florida Georgia	1,843,750 5,020,323 9,569,722	591,091 1,015,726	286,931 501,796 886,717	31.5	512.798 1.697.855 182.081	256,220 848,923	12.3 57.9	231,196 609,970	115,594 304,985	10.4 7.4	3,364,619
Idaho Illinois Indiana	15,564,720	2,645,269 8,553,819 5,660,729	1,534,328 4,259,908	256.6 136.7	6,816,788 Li 880 761	997.350 3,353,628	102.0 181.4	5,232,697	29,554 2,618,487		5,372,696
lowa Kansas Kentucky	9.757.950 10.005.211	7.219.533 3.798.047	3.513.360	193.8	4,434,057 5,730,694	2,838,900 2,838,900	138.4	3,690,910	871.299 1,845,444		3,342,721
Louisiana Maine Maryland	5,387,420 3,299,867 2,004,808	2,021,801	1,008,034 929,566	72.1 58.9	1,085,777	542.584		9,246,580	1,135,750	1	2,701,051
Massachusetts Nichigan Minnesota	5,255,300 11,562,296	333.935 9.270.093	166,968 h.559,604	332.2	4.335,990 5.972,169	2,167,995 2,985,181	20.3	511,780 511,780 2,388,450	255.890	64.2	2,664,447
Mississippi Missouri Montana	6,635,344 11,479,090	4,724,554	2,345,944	470.6	7.913.139	700,100	84.5 269.0	3,558,864	1,774,035	-	1,2,1,5,1 4,246,884 3,415,820
Nebraska Nevada New Hamnshire	7,809.353	3,255,004	1, 594, 331	190.8	2,343,573 1,673,167	1,191,149	261.5 54.4	2,007,072 2,007,072 9,600	1,003,517	251.7	3,990,995
New Jersey New Mexico New York	5,054,295 6,030,708	2,41,508 2,466,551 3,832,624	1,230,502 2,351,638	34.7	2,276,352 1,673,162	96,014 1,060,881 1,017,929	24.9 98.3	20,880	1,010,055	148.5	1,341,700 2,752,472 1,651,086
North Carolina North Dakota Ohio	5,914,683	2,494,670 197,000	1,243,481	312.9	214,800	2, 336, 652 124, 266	309.3	2,043,920 1,135,190	932,330	89.2	4,549,754
Oklahoma Oregon Pennsylvania	8,880,547 6,182,079 16,199,80h	3,374,231 3,002,674	1,756,840	116.8	2,556,806 3,745,458	2,245,576	90.1 145.5	2,215,616 1,021,690	557,118	88.9	1,584,510 1,584,518
Rhode Island South Carolina South Dakota	1,843,750 5,103,525 6,162,747	217,017 226,395 626,395	106,747 262,900 775,874	51.7 188.6	790,556 4,416,239 3320,467	395,278 395,278 1,832,523	277.44 6.6 277.44	777,410	36,754	1.1 75.6 80.7	1,304,972 2,706,400
Tennessee Texas Utah	7,949,380 23,506,431 1,274,740	2,494,552 13,290,909	1,245,146 6,622,202 1 LEZ 520	107.1 721.9 128.1	1.368.594 10.948.324	5,410,037	146.0 736.6	1,011,028 3,628,734	505,519 1,783,920	10,1 220.7	5,514,423 9,690,272
Vermont Virginia Washington	6,887,569 6,907,615	1,301,001 3,158,260 L 72L 620	1,575,817	6.29	3,241,793	417,110 1.581,803	26.4 129.6	155,340	552.277	30.0	7,177,671
West Virginia Wisconsin Wyoming	4,107,201 9,197,557 4,722,322	863,017 3,829,205 3,161,808	1,831,756 1,946,187	160.6	4,250,178	2,070,628	23.9 134.0	5,217,170 3,217,170	335, 146 1, 295, 675 716, 00h	76.7	2,999,498 3,999,498
Puerto Rico Hawaii	625,000 1,843,750	29,953	14.542	. 08	51	μ2μ.300	17.2			-	625,000 1,404,908
TOTALS	368.750.000	152.434.761	79.281.572	8 267 7	1 KE NZE DEC	26 15)1 715	E OOL O	1070 404 34	70 007 711	0 000 0	002 310 071

		BALANCE OF	FUNDS AVAIL PROJECTS	# 80,1495 73,560 55,616	6 45,490 1,478,300 266,0320	135,989			5 83,939 0 14,242 16,644		36.065						-	1 92,261	131,042 0 81,480 5 113,476		6 2,497 108,037 17.522	5.789
CTS		N	Miles				C•C+	0°5 0°5	11.	12	ູ	1.1	ت ما ما		2.3 16.1	55	93.	17.	1.0 r	*#	ູ	
PROGRAM HIGHWAY PROJECTS propriation act of 1935)		APPROVED FOR CONSTRUCTION	Works Program Funds		<b># 61.700</b>	11,076	7,396	37.924 144.075 27.528	314,892 157,053 62,100	441,384 439,338	4,650 120,248	8,900 36,658 60,987	67.729 4.767	96.573 96.573 83.314	305.778 27.700 86.471	190,540 90,738 25,002	3.277.909	277,801	492,461 60,475 76,141	18,090 45,533	145,080 2,999	
ACT OF 1935)		APPROVE	Estimated Total Cost		\$ 63.950 106 Eliz	19,870	365.1	37.924 60.176 27.528	314,901 171,682 62,100	475,216 812,060	5,766	8,900 38,755 100,210	67.729 4.767	105,542	305.778 33.878 86.471	193,060 90,798 25, 007	3,615,290	324,807	192,461 76,061 76,141	18,212 45,533	52,460 2,999	
ATION A			Miles	29.1 17.4 64.5	29.3	21.4	2.7 2.7 41.3	105.7 101.2 59.5	27.6 55.9 14.3	10.8	4.8 85.5	75.1 32.8 10.2	80.3	16.4	33.4 81.0 90.1	140.0 69.5	81.3	87.4 80.7	43.6 28.0	103.8	64.1 15.1 28.4	8.5
API	~	UNDER CONSTRUCTION	Works Program Funds	\$ 771,036 239.795 594,386	1.868.531 153.494	276,827 676,031	56.927 1.239.242	2.099.308 1.432.139 1.545.720	689,216 1,468,068 317,194	465,489 2.333.754	291,871 743,326	1.206,169 1.547.473 254,250	1,036,383	2,278,707 398,230	1,657,560 1,834,445 875,313	4,114,612 1,114,123	2,835,918	921,756	1.375.681 792.468 754.943	162.427 582.608 227.272	1.541.511 368,403 546,972	298,335
	D BY THE EMERGENCY RELIEF AS OF APRIL 30, 1937	UNDE	Estimated Total Cost	# 771,036 354,525 595,560	2.109.674 153.511 207.175	355, 632	56,927 1,239,320	2.099.308 1.492.589 1.557.217	689.216 1,612,242 717,194	465,489 2,687,114	336.871	1.207.208 1.732.190 254.250	1.036.387 352.584	2,291,862 398,230	1,657,560 1,872,245 878,804	4,162,047 1,126,782	2,893,643	946,605 749,357	1.375.681 896.406 385.607	220,948 594,789 201, 815	1.677.410 509.195 549.869	322,161
MERGENCY			Miles	109.2 178.5 294.7	230.4 98.1	1.17	182.9 182.9 1429.9	122.8 424.6 319.2	315.3 110.2 57.7	17.7 2.5	287.9 815.4	159.6 743.8 185.1	283.8 89.1	14.5	134.5 208.3 231.4	139.8 324.1	91.1	138.7	92.0 1,078.9 169.5	20.6 927.8	30.7 328.1 124.2	8.8 9.8
BY		COMPLETED	Works Program Funds	\$ 3,299,584 2,256,486 2,702,059	5,772,207 1,763,469 10,65,771	1,871,739	2,132,180 7,431,432	2,802,643 3,503,036 3,421,727	2.638.223 1,251,066 1.280.861	473,816 216,783	5,968,828 4,397,312	2,169,455 4,332,974 3,324,379	2.517.334	716.597 2.249,149	9.037.819 2.775.072 1.551.437	3,204.702 3,204.702	1.741.190	1.377.378	2.193.276 11.054.928 1.527.294	2,799,244	642, 324 4, 344, 446	961, 919 621, 909
(AS PROVIDED			Estimated Total Cost	# 3.314.570 2.817.718 2.721.113	5,960,206 1,819,885 125,782	1,905,233	2.223.133	2.959.595 3.732.818 3.440.405	2.706.939 1,406.224 1.294.221	478.755 216.783	6,233,848 5,162,242	2,173,987 4,372,992 3,339,802	2,591,622	719,597 2,253,708	9,466,047 2,807.785 1,560,107	3,305,823 3,254,994	1.831.638	1,451,458	2,200,158 12,064,251 1,678,191	831,474 2,862,164 3,100,004	645,820 4,912,675 1,656,909	949,496
CUKKENI SIALUS (AS PROV			APPORTIONMENT	# 4,151,115 2,569,841 3,352,061	7.747.928 3.395.263	900,310 2.597,144	2,222,747 8,694,009	4,941,255 4,991,664 4,994,975	3.726.271 2.890.429 1.676.799	3.262.885	6,301,414 5,277,145	3,457,552 6,012,652 3,676,416	3.870.739 2.243.074	3,129,805 2,871,397	11,046,377 4,720,173 2,867,245	7,670,815 4,580,670 7,038,642	797.742.6	2,702,012 2,976,454	4,192,460 11,989,350 2.067,154	924,306 3,652,667 3,026,161	2.231.412 4.823.884 2.219.155	949,496 926,033
CUK			STATE	Alabama Arizona Arizona	California Colorado Connecticut	Delaware Florida Georgia	Idaho Illinois	lowa Kansas	kennicky Louisiana Maine	laryland lassachusetts	Michigan Minnesota	Mississippi Missouri Montana	Nebraska Nevada New Hamoshire	New Jersey New Mexico	New 1 ork North Carolina North Dakota	Ohio · · Ohiahoma Oklahoma	Pennsylvania Dhada Islaad	South Dakota	Tennessee Texas Utah	Vermont Virginia Washington	West Virginia Wisconsin Wyoming	District of Columbia Hawaii

Ś			an anter ted	EVENDS AVAU. FUNDS AVAU. ABLE FOR NEW. PROJECTS	\$ 191,928 42.210	36,415	208,681 491,121 880,575	27.739	3,528,864	374.885 78,221 12 246	49.572	538,936	414,782 284.859	569.323	925.700 37.459	120.021	101.371	56.413	77.422	689.706 1,410	608.279	376,274	1,038,080	1.51,7,894	15,006 582,188	217.874	405,024 39,140	691,881	3.752	150,308	14,000	18,711,844
PROJECTS				Grade Crossings Protect: ed by Signals of Other-	91	30			140	60	~					- 0	2	17	nm			-	10		142	22	163	10 #	~	\$		397
OJE			NUMBER	Grade Crassing Struc- taret Re- construct- ed								4	2	Q			-				-	-	2	4	Ħ							23
PR		UCTION	4	Grade Crossings Eliminated by Separa- tion or Relocation	-	-	9	~~~	10	м	~	4	9	~	•	LC.	- 1	N		-	m	NRCE	m	- 01	m	00		5	7	0		119
CROSSING 0F 1935)		APPROVED FOR CONSTRUCTION		Works Prestam Funds	\$ 6.220 95.405	195,448	73,800	260.500	457,310	13,339	216,070	628,328	943,290	615.536	249,991	178.870	253,300	274.271	5.050 8.214	125,990	681.000	312,170 312,170	342,033	1,212,148	162* 164	1424, 384 All All	184,143	6,200 439,325	7.908	210.602		13,381,125
ADE A		APPRO		Estimated Total Cost	\$ 6,220 95,405	195.718	73,800	260,500	457,310	13,339 808,000	229,877	628,328	988,020	638,005	249,991	178.870	367,252	274.271	5.050	125,990	681,000	312,170	342.033	21,842	200.047	468, 200	111. 484	6,200 1145,114	7,908	203.012		14.079.179
GR				Grade Crossings Protect- ed by Signals or Other- wise			-			6	25.	-		19	,	7		-	-		0.0 0	201	-	6	m	- t o	11	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~		-		345
M			NUMBER	Grade Crossing Struc- ture- coastruct- ed		2		-	.#	- 90	101	- ~	∾ -		21	-		~		5	đ,	nmu	23.	- 60	2.0	ma	J	20	m	nm		138
RA		NO	Z	Grade Crossings Eliminated by Separa- tion or Relocation	12	50	0 10 10	10	17	522	37	ño	19	- ~	11	50	54.0	14	- 5	50	28	22	5	45 8	53	30	12 42	ωĔ	5	2010	- m	797
KS PROGRAM GR	30,1937	UNDER CONSTRUCTION		Works Program	# 1.254.847	1.749.539	2,242,680 939,940 758,630	R26.198	852,182	386,236 5,161,247 7 hzh h22	2,612,629	2, 450, 516	1,855,395	518.504	2,144,993 3,454,277	2.522.519	4.998.457	1.339.912	395.100	2,542,324 670,400	9.571.965	2,129,192 2,019,557 5 201 800	1,857,765	6,993,014	61,626	1.578.528 2 166 770	4,437,420	1,046.340	1,158,884	2,886,659 651,775	226,161	95,689,928
<b>O STATES WORKS</b> P THE EMERGENCY RELIEF	OF APRIL			Estimated Total Cost	\$1,254,847	1.752.024	2,310,757 939,941 779,941	RP6. 541	852,182	386.349 5.264.447 7.421.022	2,693,205	5,010,989 1.740.249	1.855.395	518.504	2,454,277	2.581.680	5,194,702	1.339.912	211,658 395.145	2,553,369	9.828.011	2,022,837 5,022,837 5,551,208	1.857.765	7,501,793	61,626	2 1.578.528	1, 111, 1488 14, 1411, 1488 966, 695	273,801	1.194.299	2.977.911	351.976	98,004,266
STATES	AS			Grade Crossings Protect- ed by Signals or Other- wite	9	2				4		ŧ	0	5	l t	27	-	9				0	4	N	m	21	- 20-	00 1	2			1111
ST/			NUMBER	Grade Crossing Struc. taret Re- construct ed	-	4	00	Ľ	N		~	2	_	m	01 <del>2</del> 1 0	9 4		- 1	nη	~~~	12	N	- 101	∩ ∞	- 7	~	- 21-	un 100	σ	7		171
ED Y TJ				Grade Crossings Eliminated by Separa- tion or Relocation	36	35	21	11		140 140	68	ری ۲۱	13	m	74-7	56	345	5	- 11	5	12	19	3	26 8	14	25	- 29 ci	30 7	18	52	ົດເບ	987
IF UNITED		COMPLETED		Works Program Funds	\$2,581,622 1.047.303	1.592.659	4,961,201 1,200,507 73,479	1 657 852	57,593	900,019 4,259,716	2.722.408	1,054,606	718.213	358.388	3,273,461	2.159.413 1.025 605	789,024 2.494,225	1,885,845	341.748	625,806 1,053,475	2.715.945	1,009,4440 499,472 282 212	1.766.832	867,119	623.059 821.976	847.204 615 005	5,529,395 224,927	1,596,742	1.924.497	1.985.716 540.513	170,643	68,217,103
ATUS OF (AS PRO				Estimated Total Cost	\$ 2,581,622 1.066.867	1.597.621	5,126,776 1,234,005 73,479	130,000	59.085	906,484 4,260,678 1 707 167	2.779.978	1,054,606	719.316	358, 388	3, 339, 1443	1.035.753	789,025	1,895,865	341.748	625,806	2.720.770	1,009,4440 1499,551	1.769.995	1,964,345	624,125 830,842	847,205 615 105	5.533.373	1,667,608	1.931.431	2.017.945 540.621	170.643	69.214.326
CURRENT STATUS				APPORTIONMENT	\$ 4,034,617 1.256.099	3.574,060	7.486.362 2.631.567	418,239	4,895,949	1.674,479 10.307,184 5.111,096	5,600,679	3.672.387	3,213,467	2.061.751	6.765.197	7.241.176	6, 142, 153 2, 722, 327	3,556,441	822.484	3,983,826	13.577.189	3,207,473 3,207,473 8 hzo 207	5,004.711	11,483,613	3.059.956	7 off 070	10,855,982	729.857 3.774.287	3.095.041	5,022,683 1,360,841	110,80H	196,000,000
CURI				STATE	Alabama Arizono	Arkansas	California Colorado Connecticut	Delaware Florida	Georgia	Idaho Illinois Indiana	lowa	kentucky	Louisiana	Maryland	Massachusetts Michigan Minneedta		Missouri Missouri Montana	Nebraska	New Hampshire	New Jersey New Mexico	New York	North Carolina North Dakota Ohio	Oklahoma	Oregon Pennsylvania	Rhode Island South Carolina South Debote	RIONR/T ITTOOC	Tennessee Texas Utah	Vermont	W ashington	West Virginia Wisconsin Wyoming	Dist of Columbia Hawaii	TOTALS

May 1937

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## SEPARATE REPRINT FROM THE YEARBOOK

No. 1036Y. Road Work on Farm Outlets Needs Skill and Right Equipment.

### TRANSPORTATION SURVEY REPORTS

- Report of a Survey of Transportation on the State Highway System of Ohio (1927).
- Report of a Survey of Transportation on the State Highways of Vermont (1927).
- Report of a Survey of Transportation on the State Highways of New Hampshire (1927).
- Report of a Plan of Highway Improvement in the Regional Area of Cleveland, Ohio (1928).
- Report of a Survey of Transportation on the State Highways of Pennsylvania (1928).
- Report of a Survey of Traffic on the Federal-Aid Highway Systems of Eleven Western States (1930).

# UNIFORM VEHICLE CODE

- Act I.—Uniform Motor Vehicle Administration, Registration, Certificate of Title, and Antitheft Act.
- Act II.—Uniform Motor Vehicle Operators' and Chauffeurs' License Act.
- Act III.-Uniform Motor Vehicle Civil Liability Act.
- Act IV .- Uniform Motor Vehicle Safety Responsibility Act.
- Act V.-Uniform Act Regulating Traffic on Highways.

Model Traffic Ordinances.

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BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS 12.020 2.469 1,765 13,190 ,007,216 38,049 83,408 113,629 32.984 12.789 339.947 164.249 17.002 74.516 32.621 7,111 15.773 96,071 14,074 52,811 57,902 217,215 50,501 7,160 172.239 18,484 128,369 38,582 52,159 81,941 20,967 139,697 832 339 7.677 35.907 53.036 57.314 26,866 72,061 11,025 23,011 7,845 38, 347 1935 Public Work Funds 3,562,428 1934 Public Works Funds 12,895
2,711
6,508 47.854 359.365 8.910 92.786 16.203 4 25.575 64,940 8,420 88,911 46,322 87.669 41.035 92.035 14.463 80,316 43,866 6,000 25,266 27,052 7,278 2,141 4,061 26,581 50.631 1,913 43,668 2,700 53,681 3,825 84,460 32,906 13,250 284 4,888 1.574.707 AS PROVIDED BY SECTION 204 OF THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS) Mileage 1.5 .1.8 117.1 10.6 €. 7.5 6.9 33.0 APPROVED FOR CONSTRUCTION 1935 Public Works Funds 1,886 7,100 61,727 285,822 810 4,200 5,000 388,620 45,000 4,863 133,980 20,500 800 299,044 91,9444 40,000 216,950 9.212 31,465 12,194 318,502 17.808 10,802 66,425 635 2,565,252 5,400 23.834 46.727 14,861 1,650 CURRENT STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION 1934 Public Works Funds 542.688 \$ 5.574 11,054 3,960 43,050 55,000 146,380 140,000 51,340 6.735 131.952 5,521 323.0 Mileage 21.5 9. 4.1 5.8 5.0 21.4 2.7 3.5 6.9 ۰. 2.9 13.3 1.5 7.2 16.4 11.1 7 8 7.9 5.7 29.1 1935 Public Works Funds \$ 115,872 24,073 49,922 114.732 233,464 693,523 83,807 885,007 98,761 196,600 107,484 67,504 215,459 26,540 68,343 96.788 292.626 488,701 121,765 56,546 21,979 21,979 4,169 663,430 9,681 525,499 261,141 116,121 245,354 196,489 64,223 356,300 2,478 198,079 91,231 240,138 46,596 279.519 9.383.613 119.88 535,213 JNDER CONSTRUCTION 1934 Public Works Funds 58,110 55.923 80.459 59.618 56.300 327.443 129,245 15,000 147,960 16,043 215.511 291.910 137.686 90.378 148,147 33.941 54.892 10.550 2.797 2,458,814 199.835 192.543 AS OF APRIL 30, 1937 Cost 114,732 59,618 209.734 135.923 67.504 215,480 43,100 68,343 650,119 1,125,699 56,546 9,681 9,681 884,792 171.795 24.073 130.731 289,764 85,967 1,141,513 227,844 96,788 372,262 201,659 21,979 14,169 553.051 253.807 392.271 196,489 64,223 569,624 2,478 348,311 96,671 115.830 111.675 357,457 552.762 12,810,860 763.9 633.2 74.0 770.8 542.9 619.9 130.0 307.3 763.8 501.4 724.3 482.7 .132.3 768.1 724.1 .441.2 .058.4 .035.5 758.8 78.3 806.6 168.0 89.2 622.0 492.8 .780.7 590.9 141.0 621.0 302.7 213.3 619.7 51.8 35.125.2 258.9 193.5 152.2 85.9 749.9 822.6 1.345.2 2.116.1 792.1 Milcage 3,813,459 2,605,032 3,371,459 7.735.533 3.465.039 2,414,688 3,278,773 2,153,743 7,945,886 4,814,845 4,920,929 5,009,852 3,719,296 2.694.989 1.671.457 1.102.724 3,094,281 6,298,778 4,841,459 3,000.758 4,998,938 3,655,874 3.746.127 2.271.616 949.520 2,452,166 2,910,398 10,637,410 4.516.899 2.217.011 7.524.157 ,430,360 ,949,337 1,012,094 2,472,056 2,918,908 ,159,606 ,873,156 ,132,691 940.847 5,342,209 5,026,509 1,806,022 4,888,205 2,235,553 968,979 184,488,707 1935 Public Works Funds COMPLETED 4,477,339 17,348,739 9,892,557 10,055,660 10,089,600 7,480,730 7.812.918 1934 Public Works Funds \$ 8,301,315 5,203,675 6,627,424 15.607.354 6.874.530 2.758.269 1,818,504 5,175,534 9,272,425 5,763,651 3,346,497 3,475,616 6,550,778 12,736,227 10,510,789 6,785,720 2,088,271 7,425,285 6,050,212 5,749,069 22,118,746 .205.116 9.214.658 6.047.835 18.525.500 ,998,708 ,220,386 ,911,486 8,490,706 24,197,559 4,192,008 1,867,452 7,322,399 6,112,042 .334,882 .724,881 .457,871 1.916,469 389,423,791 12.970.090 7.068.575 2.999.744 30,631,871 11,269,838 4,516,912 2.782.833 9.024.260 13.427.069 7,077,041 26,465,811 15,554,741 15,581,748 15,426,822 12,187,003 9,135,275 5,245,971 5,676,194 10,401,682 20,717,272 16,197,923 12.773.693 18.185.313 11.764.611 9.101,212 8,898,727 39,972,260 14.951.036 8.592.708 24.759.236 14.745.689 9.960.993 28.996.588 3,150,270 8,025,120 9,407,360 6.479.960 5.457.551 6.931.484 2.887.584 631.103.599 15,671,604 9,005,933 10,969,067 .692.251 .981.565 3,166,359 11,588,998 9,397,793 Cost Total Act of June 18, 1934 (1935 Fund) 7,932,206 5,486,006 1,454,868 5,118,361 5,117,675 5,818,311 .963.932 .711.586 ,452,558 ,425,558 3.540.227 6.173.740 3.769.734 3,964,364 2,302,356 969,462 3.220.879 2.941.700 11.327.921 4,840,941 2,938,967 7,865,012 4,685,180 5,097,814 9,590,788 4.259.842 2.641.935 3.428.049 923,395 2,661,343 ,921,486 ,921,401 2.770.954 3.047.643 2,280,335 1,941,837 973,842 4,302,991 12,291,253 2,132,691 948,007 3,765,387 3,106,412 200,000,000 APPORTIONMENTS Sec. 204 of the Act of June 16, 1933 (1934 Fund) 8.370.133 5.211.960 6.748.335 15.607.354 6.874.530 2.865.740 1,819,088 5,231,834 10,091,185 4,486,249 17,570,770 10,037,843 10,055,660 10,089,604 7,517,359 6.597,100 12.736.227 10.656.569 6.978.675 12.180.306 7.439.748 7,828,961 4,545,917 1,909,839 6, 346, 039 5, 792, 935 22, 330, 101 5,828,591 3,369,917 3,564,527 9,522,293 5,804,448 9.216.798 6.106.896 18.891.004 1.998,708 5.459,165 6.011,479 1,918,469 8,492,619 24.244.024 1,867,573 7,416,757 6,115,867 4.474.234 9.724.881 4.501.327 394,000,000 District of Columbia. Hawaii Nebraska Nevada New Hampshire Rhode Island South Carolina South Dakota North Carolina North Dakota Ohio Massachusetts Michigan Minnesota STATE West Virginia Wisconsin Wyoming Oklahoma Oregon Pennsylvania New Jersey New Mexico. New York TOTALS. Vermont Virginia Washington Mississippi. Missouri Montana Tennessee Texas Utah California... Colorado.... Connecticu Iowa Kansas Kentucky Louisiana Maine Maryland Delaware Florida Georgia Alabama. Arizona... Arkansas. Idaho Illinois Indiana