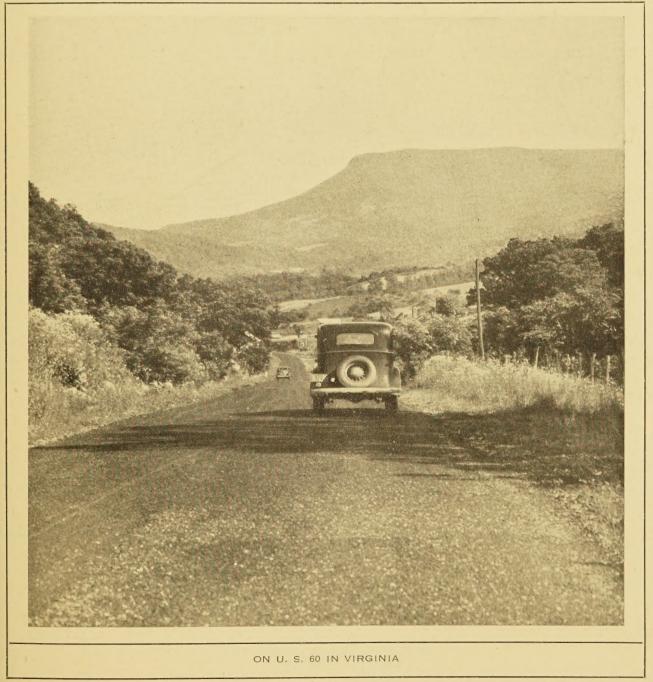


VOL. 16, NO. 6

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# PUBLIC ROADS A Journal of Highway Research

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The reports of research published in this magazine are necessarily qualified by the conditions of the tests from which the data are obtained. Whenever it is deemed possible to do so, generalizations are drawn from the results of the tests; and, unless this is done, the conclusions formulated must be considered as specifically pertinent only to described conditions.

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> CERTIFICATE: By direction of the Secretary of Agriculture, the matter contained herein is published as administrative information and is required for the proper transaction of the public business

# FURTHER STUDIES OF LIQUID ASPHALTIC ROAD MATERIALS

## BY THE DIVISION OF TESTS, U.S. BUREAU OF PUBLIC ROADS

Reported by R. H. LEWIS Associate Chemist, and W. O'B. HILLMAN, Assistant Highway Engineer

IN A REPORT recently published by the Bureau of Public Roads on A Study of Some Liquid Asphaltic Materials of the Slow-Curing Type,<sup>1</sup> it was shown that the action of sunlight, heat, and air on these materials when exposed in relatively thin films produced residues with physical and chemical characteristics differing greatly from those of the residues developed in the usual laboratory heat tests. It was also shown that when these materials were mixed with a standard sand, molded into cylinders by the Hubbard-Field method, and subjected to the same exposure conditions as the thin films, they developed stability, or bonding strength that could not be attributed entirely to the loss of volatile matter.

#### MATERIALS STUDIED TYPICAL OF ALL CLASSES OF LIQUID ASPHAL. TIC MATERIALS FROM PRINCIPAL PRODUCING AREAS

The materials used in the earlier investigation were slow-curing liquid asphalts. They were the products of 10 refineries located in the far and middle West. In further studies conducted in 1933, that are the subject of this report, 32 materials typical of the slow-, medium-, and rapid-curing types of liquid asphaltic materials were used. These samples are identified in table 1 and were the products of 25 refineries located in all sections of the country and probably were made from petroleums of widely different bases and by various refining processes.

Sample identifi- cation	Labo- ratory num- ber	Type of material	Pro- ducer	Refin- ery	Location of refinery	Remarks
1	$\begin{array}{c} 36626\\ 34354\\ 35130\\ 35481\\ 36425\\ 36045\\ 36945\\ 36994\\ 35195\\ 35334\\ 35200\\ 35394\\ 35200\\ 35351\\ 35180\\ 35181\\ 35103\\ \end{array}$	SC-2 SC-2 SC-3 SC-3 SC-3 SC-2 SC-2 SC-2 SC-2 SC-2 SC-2 SC-2 SC-2	1 2 2 3 3 4 5 6 7 8 9 9 2 2 2 2 10	$ \begin{array}{c} 1\\2\\3\\4\\5\\6\\7\\8\\9\\10\\11\\11\\12\\13\\14\\15\end{array} $	Oklahoma Missouri Illinois Arkansas do Oklahoma West Virginia Rhode Island Louisiana 	Included in 1932 ex-
18	$\begin{array}{c} 35367\\ 34322\\ 30089\\ 39161\\ 39162\\ 39068\\ 34285\\ 36762\\ 35345\\ 35352\\ 35352\\ 35352\\ 35247\\ 36771\\ 33607\\ 32916\\ 32622\\ \end{array}$	SC-2 SC-2 SC-2 SC-2 SC-2 SC-2 SC-2 SC-2	$ \begin{array}{c} 10\\ 11\\ 12\\ 13\\ 13\\ 7\\ 14\\ 3\\ 2\\ 2\\ 7\\ 15\\ 16\\ 2\\ 13\\ \end{array} $	$\begin{array}{c} 16\\ 17\\ 18\\ 19\\ 20\\ 21\\ 22\\ 12\\ 23\\ 24\\ 25\\ 13\\ 19\\ \end{array}$	Illinois do California do south Carolina New Jersey do Indiana Maryland do Wyoming California	posure. do. do. Steam-reduced Cali- fornia residual oil.

TABLE 1.—Products tested

2297-35-1

Furol viscosity was below specification limit.
 Furol viscosity was above specification limit.
 Penetration of distillation residue was below specification limit.

1 R. H. Lewis and W. O'B. Hillman, PUBLIC ROADS, June 1934, vol. 15, no. 4.

As shown in table 1, 23 of the materials were of the slow-curing type, of which 4 samples were tested in 1932 and were included in this study for comparative purposes. Four were of the medium-curing type and five were rapid-curing products. They met the provisional specifications, as given in table 2, of the Bureau of Public Roads and the Asphalt Institute, except as noted in table 1. Samples 21 and 22 were the only slow-curing products for which there was any definite information as to origin or method of manufacture. Both of these materials were steam-reduced California residuals without subsequent blending. All of the rapid-curing products were prepared from 85-100 penetration asphalt and solvent naphtha. The composition of sample 27 was unknown but the other medium-curing products, samples 30, 31, and 32, were, respectively, 110-120 penetration asphalt, 94+ asphaltic road oil, and 100-120 penetration asphalt fluxed with a heavy grade of kerosene. These three medium-curing materials, although subjected to all of the laboratory tests, were exposed only under special conditions that are described later in this report.

TABLE 2.—Specification requirements for grades of liquid asphaltic road materials investigated

	SC-1	SC-2	SC-3	MC-1	MC-2	RC-2
Flash point, °F Furol viscosity at 77° F., seconds	150+20-150	200+	200+	40-150	150+	80+
Furol viscosity at 122° F., seconds. Furol viscosity at 140° F., seconds. Total distillate to 437° F., percent		200-320	150-300		150-250	200-400
by volume		2-	2-	10-	2-	10+
Total distillate to 600° F., percent by volume		15-	10-	25+	10-20	20+
Total distillate to 680° F., percent by volume	50-	25-	20-	50-	27-	35-
Float at 122° F., seconds Penetration at 77° F.	50-	25+	25+	70-300	100-300	60-120
Ductility at 77° F., centimeters. Soluble in CS <sub>2</sub> , percent	99.0+	99.0+		60+	60+ 99.5+	60+ 99.5+
					l	

The test procedure followed that of the 1932 study except that the fixed-carbon test was omitted as the changes in inherent characteristics that occurred under laboratory and exposure conditions appeared to be more strikingly illustrated by the test for solubility in 86° B. naphtha. Two new tests were added. The Oliensis test  $^2$  for heterogeneity was made on the original materials and on all of the residues, except those from the 50-gram oven-loss test and the 10-week exposure test, and the original materials were examined micro-scopically. The results of the tests on the original materials are given in table 3 and a detailed analysis of the residues obtained in the routine laboratory tests is given in table 4.

<sup>2</sup> A qualitative test for determining the degree of heterogeneity of asphalts. G. L Oliensis, Proc. A. S. T. M., vol. 33, pp. 715-728.

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								-	TABLE 3	3.— <i>I</i>	Resu	lts d	of te	sts	on o	rigi	nal 1	nater	rials										
			E		CS2	CCI4		t 1							Di	stillat	ion						Loss	at 325°	F., 5	hour	S	Aspha	oltia
0П 77° F.			Fur visco		insoluble in C		B. naphtha	liensis tes	Micro-		D	volu		y	F. (by	weight)	ght)	Tes dist	ts on illate	22º F.	esidue at 5 sec.	50-gr	am sa	ample	20-gr	am sa	ample	resid	lue
Sample identification Specific gravity at 77	Flash point	Float at 77° F.	At 122° F.	At 140° F.	Organic matter inso	Organic matter insoluble in	Insoluble in 86° B.	Characteristic by Oliensis test	scopic smear test	Initial boiling point	At 374° F.	At 437° F.	At 600° F.	At 680° F.	Distillate to 680° weight)	Loss on cooling (by weight)	Total loss (by weight)	Specific gravity at $77^{\circ}$ F.	Index of refraction	Float of residue at 122° F.	Penetration of res 77° F., 100 g, 5	Loss	Float of residue at 122° F.	Penetration of residue at 77° F., 100 g, 5 sec.	Loss	Float of residue at 122° F.	Penetration of residue at 77° F., 100 g, 5 sec.	Time of reduction	Amount of residue
	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 52\\ 50\\ 63\\ 366\\ 36\\ 11\\ 1\\ 4\\ 4\\ 9\\ 9\\ 6\\ 20\\ 21\\ 1\\ 38\\ 33\\ 33\\ 33\\ 33\\ 33\\ 33\\ 33\\ 33\\ 33$	$\begin{array}{c} 303\\ 304\\ 274\\ 404\\ 331\\ 335\\ 181\\ 277\\ 197\\ 750\\ 228\\ 406\\ 278\\ 219\\ 217\\ 197\\ 105\\ 208\\ 219\\ 217\\ 50\\ 208\\ 219\\ 207\\ 300\\ 203\\ 200\\ 123\\ 351\\ 472\\ 2351\\ 472\\ 2351\\ 472\\ 351\\ 472\\ 431\\ 351\\ 431\\ 350\\ 350\\ 350\\ 350\\ 350\\ 350\\ 350\\ 350$	$\begin{array}{c} 123\\ 174\\ 160\\ 170\\ 104\\ 146\\ 87\\ 110\\ 195\\ 124\\ 100\\ 195\\ 124\\ 100\\ 195\\ 124\\ 100\\ 195\\ 124\\ 100\\ 137\\ 331\\ 133\\ 133\\ 133\\ 133\\ 133\\ 133$	0.19 .11 .13 .38  .09  .09	.10 .20 .09 .09 .09 .08 .11 2.17 .11	$\begin{array}{c} 9.\ 4\\ 10.\ 6\\ 6\\ 7\\ 11.\ 3\\ 9\\ 7.\ 6\\ 11.\ 3\\ 9\\ 7.\ 6\\ 11.\ 3\\ 17.\ 6\\ 11.\ 3\\ 9.\ 6\\ 11.\ 3\\ 17.\ 6\\ 11.\ 3\\ 17.\ 6\\ 10.\ 7\\ 11.\ 3\\ 17.\ 6\\ 10.\ 7\\ 10.\ 10.\ 10.\ 10.\ 10.\ 10.\ 10.\ 10.\$	ОНОНОННОВНИЯ В НИВ В ООННООНОВНИИ ООНОВНИИ ООНОВНИИ	do	$\begin{array}{c} 594\\ 515\\ 620\\ 520\\ 520\\ 525\\ 600\\ 525\\ 600\\ 525\\ 600\\ 536\\ 670\\ 536\\ 580\\ 670\\ 536\\ 580\\ 600\\ 530\\ 250\\ 332\\ 250\\ 332\\ 250\\ 370\\ 420\\ 420\\ 430\\ 450\\ \end{array}$		Pct.	1. 3 7. 0 1. 0 7. 0 7. 0 7. 0 7. 0 7. 0 7. 0 7. 0 7	$ \begin{array}{c} 8.0\\ 3.0\\ 2.0\\ 10.0\\ 2.0\\ 10.0\\ 2.0\\ 10.$	$ \begin{array}{c} 6.4\\ 2.4\\ 2.4\\ 1.7\\ 8.6\\ 2.5\\ 18.3\\ 1.6\\ 2.5\\ 18.3\\ 1.6\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 2.5\\ 10.2\\ 10.$	$\begin{array}{c} 2.8.6 (4) \\ 4.4.5 \\ 0.0 \\ 0.5.4 \\ 1.5.4 \\ 1.5.4 \\ 1.5.4 \\ 1.5.4 \\ 1.5.4 \\ 1.2.9 \\ 1.5.4 \\ 1.2.9$	$\begin{array}{c} 6.8\\ 5.7\\ 6\\ 7.4\\ 23.4\\ 4\\ 10.9\\ 22.3\\ 25.0\\ 17.9\\ 9.3\\ 25.0\\ 17.9\\ 9.3\\ 25.0\\ 17.9\\ 9.3\\ 25.0\\ 17.7\\ 9.9\\ 9.3\\ 23.4\\ 20.6\\ 8\\ 20.6\\ 20.2$	. 926 . 886 . 864 . 867 . 868 . 864 . 866 . 958 . 866 . 958 . 866 . 958 . 866 . 958 . 866 . 958 . 871 . 853 . 842 . 853 . 842 . 853 . 844 . 853 . 844 . 853 . 844 . 853 . 844 . 853 . 844 . 855 . 854 . 855 . 854 . 855 . 854 . 855 . 8555 . 8555 . 8555 . 8555 . 8555 . 8555 . 8555 . 8555 . 8555 . 8555	$\begin{array}{c} 1,5146\\ 1,5246\\ 1,5220\\ 1,4930\\ 1,4930\\ 1,4738\\ 1,4809\\ 1,4748\\ 1,4809\\ 1,5268\\ 1,4750\\ 1,5268\\ 1,4750\\ 1,5268\\ 1,4750\\$			$\begin{array}{c} 21.5 \\ 20.7 \\ 22.1 \\ 519.8 \\ 221.1 \\ 17.6 \\ 14.5 \end{array}$	35         30           309         39           36         17           18         25           29         42           20         10           18         30           300         31           24         26           422         26           424         26           425         22           100         3           100         3           1086         144		22. 4 22. 4 27. 8 20. 4	38 51 577 20 34 277 45 74 277 36 39 30 222 75 57 52 55 58 38 36 58 58 58 58 58 58 58 58 58 58 58 58 58		$\begin{array}{c} 400\\ 470\\ 800\\ 2366\\ 700\\ 2366\\ 700\\ 933\\ 121\\ 1100\\ 600\\ 113\\ 176\\ 833\\ 833\\ 833\\ 833\\ 833\\ 833\\ 833\\ 83$	$\begin{array}{c} Pct. \\ 71.9 \\ 74.7 \\ 74.7 \\ 74.7 \\ 74.7 \\ 62.4 \\ 64.8 \\ 65.3 \\ 69.0 \\ 65.3 \\ 26.5 \\ 84.5 \\ 84.5 \\ 77.2 \\ 75.9 \\ 77.2 \\ 78.4 \\ 84.5 \\ 78.4 \\ 84.5 \\ 78.5 \\ 84.4 \\ 78.5 \\ 84.4 \\ 78.5 \\ 84.4 \\ 78.5 \\ 78.9 \\ 75.8 \\ 93.75.8 \\ 75.8 \\ 93.75.8 \\ 75.8 \\ 93.75.8 \\ 75.8 \\ 93.75.8 \\ 75.8$

 ${}^{1}\mathbf{H} = \mathbf{H}$ eterogeneous;  $\mathbf{O} = \mathbf{H}$ omogeneous;  $\mathbf{SH} = \mathbf{Slightly}$  heterogeneous.

#### DISTILLATION CURVES CLEARLY DISTINGUISH SLOW-, MEDIUM-, AND RAPID-CURING MATERIALS

The distillation curves for the various materials, plotted in figure 1, serve to distinguish the three classes of materials according to their curing properties. It will be noted that the initial boiling points of the slowcuring products, samples 1 to 23, inclusive, were all above 450° F. and in only two instances were they less than 500° F. The medium-curing materials, samples 27, 30, 31, and 32, had initial boiling points as low as 360° F. with none above 450° F.; and the rapid-curing products, samples 24, 25, 26, 28, and 29, all had initial boiling points below 370° F.

With all three classes of materials, the percentage of distillate increased as the temperature increased but not at the same rate. For the slow-curing materials, the rate continued fairly uniform up to 680° F., but' with the medium- and rapid-curing products it decreased as the temperature approached 680° F. The decrease in rate was more pronounced for the rapidcuring products. The rate of distillation may be illustrated by expressing the amount of distillate off at any temperature in terms of the total distillate recovered at 680° F. For the medium-curing materials the amount at 600° F. was 65 to 85 percent of the total. For rapid-curing materials the amount was 0 to 50 percent at  $374^{\circ}F$ ., 41 to 85 percent at 437° F., and 88 to 94 percent at 600° F. The residues from the distillation of the slow-curing products were all fluid, while those of the medium- and rapid-curing materials were semisolid.

That portion of the total loss in the distillation test designated as loss on cooling depends upon the amount of material boiling off immediately above 680°F. As would be expected from the slope of the distillation curves, this loss was greater for the slow-curing products than for the medium- and rapid-curing products. It ranged from 1.2 percent to 5.8 percent with an average of 4.5 percent for the slow-curing products, from 2.9 percent to 4.2 percent with an average of 3.4 percent for the medium-curing products, and from 1.5 percent to 3.1 percent with an average of 2.2 percent for the rapidcuring products.

The total volatile matter in the slow-curing materials as determined by the distillation test, including both the distillate recovered and the loss on cooling, ranged from 2.8 percent to 25 percent with an average of 13.8 percent for the group. The average loss on cooling of 4.5 percent actually represented 36 percent of the total volatile matter in the average slow-curing material used in this study. While this loss on cooling may be unimportant in estimating the relative volatility of various liquid asphalts, it must be considered if the results of the distillation test are to be compared directly with the results of other laboratory and exposure tests.

The different classes of materials are also readily identified by the results of the volatilization and asphaltic-residue tests. The slow-curing products lost in the 50-gram volatilization test from 53 to 76 percent, the medium-curing products from 73 to 80 percent, and the rapid-curing products from 92 to 97

TABLE 4.—Results	of tests on	laboratory residues
------------------	-------------	---------------------

-				Dist	illatio	on								Lo	ss at 3	25° F	., 5 h	ours	3							Aspl	haltic	residu	le		
				Т	ests o	n resid	lue					50-	gram	sam	ple				20-gi	ram s	ample					,	Tests	on res	idue		
					Du	ietil-		14				Τe	ests o	n res	idue				Tes	ts on	resdue						Due			14	
cation	stillation test		Pene- tration		ity ce me per	at 5 nti- eters min- ite	: insoluble in $CS_2$	Organic matter insoluble in CCl <sub>4</sub>	° B. naphtha			trat	ne- tion		Duc ity cer met per 1 ut	ti- ters nin-			at 77° F., 100 g, 5 sec.	sr insoluble in S2	r insoluble in Jl4	B. naphtha		Pe: trat			per	at 5 nti- ters	insoluble in CS <sub>2</sub>	insoluble in CCl <sub>4</sub>	B, naphtha
Sample identification	Total loss in distillation test	Float at 122° F	At 77° F., 100 g, 5 sec. At 32° F., 200 g, 60 sec.	Softening point	At 77° F.	At 34°-35° F.	Organic matter insoluble in	Organic matter	Insoluble in 86°	Loss		At 77° F., 100 g, 5 sec.		Softening point	At 77° F.	At 34°-35° F.	Loss	Float at 122° F	Penetration at 5 s	Organic matter CS <sub>2</sub>	Organic matter i COl4	Insoluble in 86° B. naphtha	Loss	At 77° F., 100 g, 5 sec.	At 32° F., 200 g, 60 sec.	Softening point	At 77° F.	At 34°-35° F.	Organic matter insoluble in	Organic matter insoluble in	Insoluble in 86°
$\begin{array}{c}1\\2\\3\\4\\5\\6\\7\\8\\9\\10\\11\\12\\13\\14\\15\\16\\17\\18\\19\\0\\21\\22\\23\\24\\25\\26\\27\\8\\20\\30\\1\\32\end{array}$	$\begin{array}{c} Pct. \\ 2.8 \\ 11.0 \\ 6.8 \\ 5.7 \\ 13.6 \\ 6.4 \\ 23.4 \\ 10.9 \\ 9.3 \\ 5.1 \\ 15.4 \\ 6.8 \\ 20.6 \\ 6.17.7 \\ 9.9 \\ 13.9 \\ 13.9 \\ 8.23 \\ 4.8 \\ 23$	27 44 87 34 38 37 36 41 25 110 60 57 50 41 41 57 50 41 		      	Cm 	     		$\begin{array}{c} Pct. \\ 0, 10 \\ s1 \\ 0, 10 \\ s1 \\ 141 \\ .34 \\ .178 \\ .34 \\ .178 \\ .34 \\ .178 \\ .34 \\ .181 \\ .100 \\ .1$	$\begin{array}{c} 16.1\\ 10.9\\ 7.8\\ 24.1\\ 18.3\\ 10.3\\ 12.7\\ 11.7\\ 18.0\\ 0.1\\ 14.0\\ 10.7\\ 21.3\\ 5.2\\ 9.5\\ 6.1\\ 1\\ 6.7\\ 9.6\\ 6.23.0\\ 225.4\\ 21.0\\ 26.8\\ 21.0\\ 226.8\\ 29.0\\ 29.0\\ 0\end{array}$	$\begin{array}{c} 2.\ 6\\ 2.\ 6\\ 2.\ 6\\ 1\\ 3.\ 0\\ 8.\ 0\\ 2.\ 1\\ 4\\ 4\\ 12.\ 7\\ 8.\ 8\\ 6.\ 6\\ 3.\ 7\\ 2.\ 4\\ 4\\ 9\\ 8.\ 8\\ 8.\ 8\\ 10.\ 7\\ 4.\ 9\\ 9.\ 0.\ 7\\ 2.\ 1\\ 1\\ 10.\ 6\\ 10.\ 7\\ 10.\ 1\ 1\\ 10.\ 1\ 1\\ 10.\ 1\ 1\ 1\ 1\ 1\ 1\ 1\ 1\ 1\ 1\ 1\ 1\ 1\$	35 300 399 366 177 18 25 299 42 29 42 100 18 30 31 24 42 26 48 34 26 42 22  100  186 144	      	    35 33 300 -23	120	Cm 	.0	$\begin{array}{c} Pct. \\ 0.8 \\ 6.6 \\ 6.4 \\ .5 \\ .6 \\ .6 \\ .5 \\ .6 \\ .5 \\ .6 \\ .5 \\ .5$	27 57 38 51 57 200 34 27 45 74 20 27 36 39 30 24 112 55 52 39 52 39 52 39 52 39 52 39 52 52 53 51 57 74 77 74 57 74 77 74 57 75 79 90 00 24 57 77 74 57 77 74 57 77 74 57 77 74 57 77 79 77 77 79 52 39 95 52 39 57 77 77 77 77 77 77 77 77 77 77 77 77		1. 17 2. 26 . 28 . 66 	$\begin{array}{c} 3.51\\ 1.91\\14\\11\\15\\06\\ 5.31\\13\\14\\10\\16\\21\\21\\21\\21\\23\\22\\23\\20\\22\\23\\22\\23\\20\\27\\16\\27\\16\\27\\$	$\begin{array}{c} 11.9\\ 15.2\\ 21.7\\ 10.0\\ 21.7\\ 10.8\\ 8.8\\ 12.2\\ 21.7\\ 10.8\\ 8.8\\ 8.8\\ 31.4\\ 11.2\\ 22.3\\ 33.4\\ 4.1\\ 22.5\\ 5.2\\ 24.9\\ 22.5\\ 24.9\\ 24.5\\ 24.6\\ 24.5\\ 24.6\\ 24.5$	$\begin{array}{c} 25.3\\ 29.4\\ 37.6\\ 35.2\\ 34.7\\ 31.0\\ 34.1\\ 46.8\\ 38.2\\ 33.2\\$	$\begin{array}{c} 107\\ 1055\\ 99\\ 92\\ 100\\ 86\\ 106\\ 84\\ 97\\ 93\\ 3\\ 103\\ 109\\ 81\\ 103\\ 109\\ 81\\ 102\\ 110\\ 113\\ 98\\ 101\\ 113\\ 98\\ 101\\ 113\\ 98\\ 101\\ 104\\ 112\\ 12\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 12\\ 102\\ 104\\ 112\\ 102\\ 104\\ 112\\ 102\\ 104\\ 112\\ 102\\ 104\\ 102\\ 102\\ 102\\ 102\\ 102\\ 102\\ 102\\ 102$	$\begin{array}{c} 37\\ 15\\ 15\\ 8\\ 28\\ 41\\ 18\\ 28\\ 34\\ 41\\ 17\\ 21\\ 35\\ 36\\ 15\\ 32\\ 27\\ 11\\ 12\\ 11\\ 21\\ 18\\ 23\\ 36\\ 17\\ 122\\ 23\\ 32\\ 12\\ 12\\ 23\\ 30\\ 222\\ 16\\ 16\\ 16\\ 16\\ 16\\ 16\\ 16\\ 16\\ 16\\ 16$	$\begin{array}{c} 107\\ 108\\ 113\\ 116\\ 118\\ 114\\ 110\\ 117\\ 113\\ 119\\ 115\\ 117\\ 113\\ 119\\ 115\\ 110\\ 115\\ 117\\ 114\\ 109\\ 110\\ 115\\ 125\\ 121\\ 121\\ 109\\ 118\\ 120\\ 118\\ 120\\ 118\\ 115\\ 115\\ 115\\ 115\\ 115\\ 115\\ 115$	110+110+110+110+53 $113 98$ $110+110+110+110+110+110+110+110+110+110$	3.8 .0 6.0 .1	0.16 .16 .19  .55  .60 .24  .40	1.62 .12 .14 .19 .15	$\begin{array}{c} 36.2 \\ 32.5 \\ 52.5 \\ 24.2 \\ 22.6 \\ 22.7 \\ 22.1 \\ 22.6 \\ 22.7 \\ 22.1 \\ 22.7 \\ 22.6 \\ 22.9 \\ 22.2 \\ 22$

percent as much as they lost in the 20-gram volatilization test. The residues from both volatilization tests of the slow-curing products were fluid; those of the medium-curing products were fluid in the 50-gram volatilization test and semisolid in the 20-gram volatilization test; and those of the rapid-curing products were both semisolid. For the slow-curing products, the loss in neither volatilization test amounted to as much as that in the asphaltic residue test. For the medium-curing products, the loss in the 20-gram volatilization test approximated the loss in the asphaltic-residue test, while for the rapid-curing products the losses in all tests, including the distillation test, were approximately the same. The loss in the 20-gram volatilization test, while always greater than that in the 50-gram volatilization test, was always less than in the distillation test. The residue from the 20-gram volatilization test may, however, be harder or softer than the residue from distillation.

The slow-curing products were reduced to a residue of 100 penetration in from 30 to 420 minutes, with an average of 126 minutes. In producing residues of the same consistency medium-curing products took from 16 to 30 minutes, with an average of 23 minutes and the rapid-curing products took from 11 to 15 minutes, with an average of 13 minutes. When making the asphaltic-residue test on the rapid-curing products, difficulty was experienced in obtaining a residue that was

not too hard, since the high volatility of the solvent caused the cut-back materials to be reduced to 100 penetration before the temperature for making the test was reached.

#### MAJORITY OF MATERIALS SHOWED A CLEAR FIELD ON MICRO-SCOPIC EXAMINATION

The microscopic test, adopted by one State highway department for the detection of cracking-coil products, was made on all materials. The specification of the State did not set up an exact procedure. It merely stated that a freshly prepared smear of the asphaltic material diluted with carbon tetrachloride should show a clear field free from carbonaceous matter when subjected to a magnification of 200 diameters. In the work covered by this report, the test was standardized by using 2 parts by weight of carbon tetrachloride and 1 part by weight of asphaltic material in preparing the slides for observation. When prepared in this manner all of the materials but seven showed a clear field. In preparing the slides for the photomicrographs illustrated in figure 2, 6 parts by weight of carbon tetrachloride and 1 of the asphaltic material were used.

Because carbon tetrachloride has both solvent and flocculent properties, its use as a diluent was questioned. Therefore, slides were also prepared with the undiluted materials and it was found that only those materials that showed flecks when undiluted showed

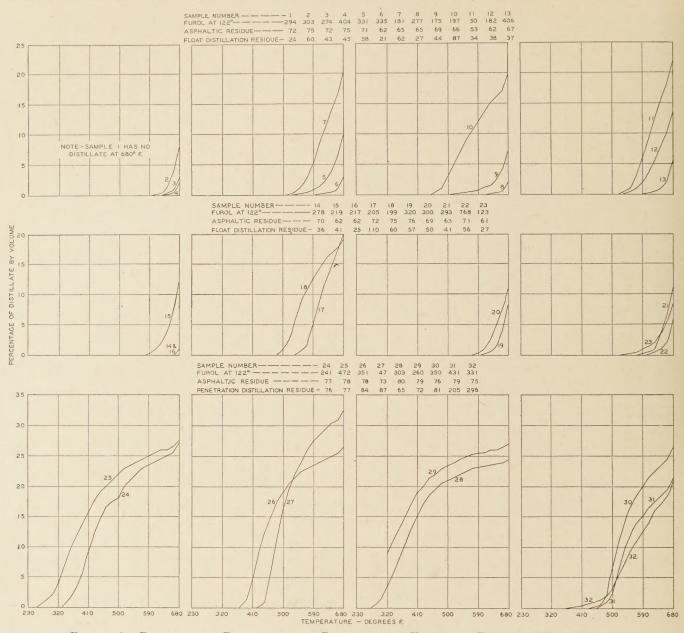


FIGURE 1.—RELATION OF PERCENTAGE OF DISTILLATE BY VOLUME TO DISTILLING TEMPERATURE.

them in the 2 to 1 and 6 to 1 dilutions. This indicates that the insoluble matter was already flocculated and that carbon tetrachloride, in the quantities used, did not precipitate carbonaceous flecks in those materials not containing them when undiluted. Recently a sample was tested that contained flecks when undiluted that disappeared on dilution with carbon tetrachloride; and another sample that was practically clear when undiluted contained flecks when diluted. In the first case, the poor solvent properties of the distillate used in the material were responsible for incomplete solution of the base asphalt, that immediately went into solution with the addition of carbon tetrachloride. In the second case it was quite evident that the carbon tetrachloride acted as a flocculent. However, since 6 of the 7 samples that contained carbonaceous material had relatively high percentages

probable that the flecks shown are particles of free carbon and carbenes.

In conducting the exposure tests three samples of each material were placed in seamless, flat-bottom tins having a diameter of  $5\frac{1}{2}$  inches and a depth of five-eighths inch. Fifty cubic centimeters of material were used to obtain a uniform film or layer thickness of about one-eighth inch. The samples were then placed in exposure boxes made of wood. A plateglass cover resting on strips of felt fastened to the edges of each box made a tight joint and excluded all dust and dirt. A current of air was passed through a wash bottle containing sulphuric acid to remove dust and eliminate moisture, and was admitted through the bottom of the boxes and escaped through slots in the sides, thus serving to carry off the vapors formed. The slots were protected from rain by thin boards of material insoluble in carbon tetrachloride, it is extending from the top of the box downward at an

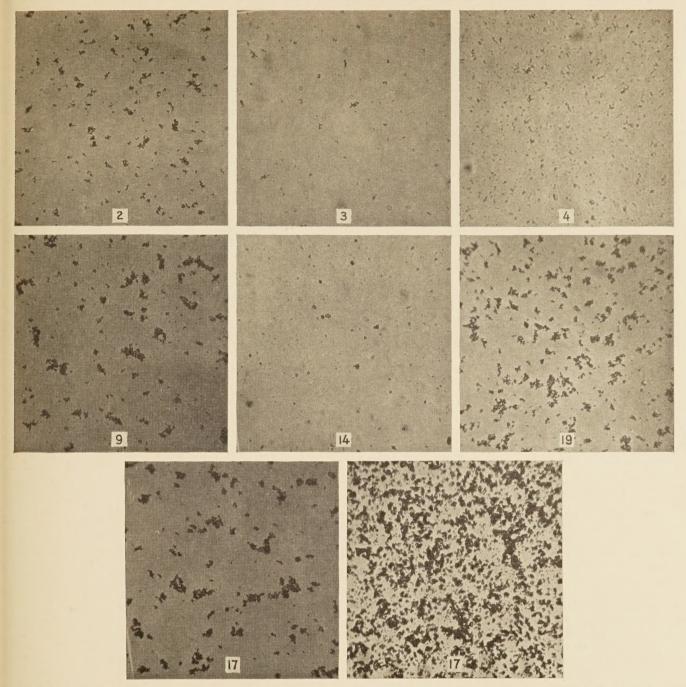


FIGURE 2.—PHOTOMICROGRAPHS OF MATERIALS CONTAINING CARBONACEOUS MATTER (MAGNIFIED 200 TIMES). THE LOWER Two Illustrations Show Results Obtained With Sample 17; in the One on the Left the Material Was Diluted WITH CARBON TETRACHLORIDE AND THE ONE ON THE RIGHT SHOWS UNDILUTED MATERIAL.

#### DIFFERENCES FOUND BETWEEN SLOW-CURING AND CUT-BACK PRODUCTS AFTER EXPOSURE

The samples were placed in the boxes on June 15, 1933, and were weighed periodically to determine the loss in weight. A complete set of samples was removed and tested at the end of 5 weeks, another set at 10 weeks, and the last set at 15 weeks. The temperature of the boxes was dependent entirely upon the radiant

angle of about 45°. Cotton batting inserted in the slots excluded dust. A thermometer in each box provided a means of determining the temperature. The assembly of the boxes is shown in figure 3. that of the air. During the period of exposure the average maximum daily air temperature was 85° F. The samples exposed for 5 weeks were subjected to 333 hours of sunlight and those exposed for 10 and 15 weeks were subjected to 611 and 866 hours of sunlight, respectively. The percentage of loss at different periods of exposure is given in table 5 and the results of tests on the residues are given in table 6. Photographs of typical surfaces at the end of 15 weeks are shown in figure 4.

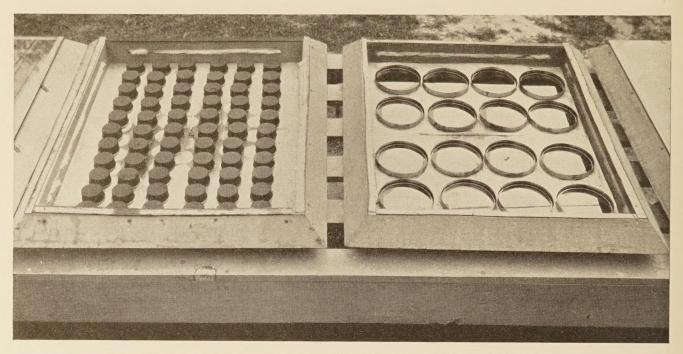


FIGURE 3.—STABILITY SPECIMENS AND THIN FILMS OF MATERIAL EXPOSED TO SUNLIGHT.

While the majority of the materials progressively lost weight during the period of exposure, a number of them actually gained at first, although they later lost more than the amounts gained. An exception to this was sample 1, which had gained 3.6 percent at the end of 8 days and at the end of 15 weeks still showed a slight gain. The samples exposed for 15 weeks were used in determining the loss at 2, 8, 15, 22, 50, and 105 days, while the percentage of loss given for the 35- and 70-day exposures was based on samples used for test at the end of 5 and 10 weeks, respectively. This was done to eliminate errors in calculating the results of subsequent tests made upon the respective residues and accounts for slight variations that may appear to indicate gains instead of losses.

As was expected, the cut-back products lost weight very rapidly. At the end of 2 days they had lost from 86 to 92 percent of their maximum loss with an average of 89 percent, and at the end of 35 days they had lost from 97 to 100 percent with an average of 99 percent. For the slow-curing products the rate of loss was much less, but was considerably more variable. In 2 days those samples that had undergone a loss had lost from 3 to 60 percent of their maximum loss with an average of 35 percent. In 15 days they had lost from 16 to 84 percent with an average of 50 percent, in 35 days from 63 to 100 percent with an average of 82 percent, and in 70 days from 74 to 100 percent with an average of 89 percent. Some idea of the relative speed of curing or volatility of the two types of material may be obtained by comparing their losses under these exposure conditions. The slow-curing materials took 70 days to lose an average of 89 percent of their volatile matter, while the cut-back materials underwent the same percentage loss in 2 days.

#### LOSS IN DISTILLATION TEST MOST NEARLY APPROXIMATED LOSS IN 15 WEEKS' EXPOSURE FOR ALL TYPES OF MATERIALS

Figure 5 shows the relation between the percentage

TABLE 5.—Loss in thin film expo
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			×					
Generale i den ti			L	oss on exj	posure for	r—		
Sample identi- fication	2 days	8 days	15 days	22 days	35 days	50 days	70 days	105 days
1           2           3           4           5           6           7           10           11           12           13           14           15           16           17           18           19           20           21           22           23           24           25           26           27           28	$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	$\begin{array}{c} Percent \\ -3.6 \\ 4.3 \\ 2.0 \\ 1.5 \\ 5.4 \\ 1.2 \\ 12.9 \\ -2.4 \\ 14.4 \\ 7 \\ 8.9 \\ 2.6 \\ -1.5 \\ -1.5 \\ 12.6 \\ 15.7 \\ 15.7 \\ 12.6 \\ 15.8 \\ 3.7 \\ 3.4 \\ 3.7 \\ 21.2 \\ 21.6 $	$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	Percent -3.4 7.6 4.7.6 4.7.6 4.1 1.7.7 5.3 15.7 18.2 11.8 6.1 3.4 8.4 1.8 1.5.9 17.4 8.6 9.8 8.2 6.7 6.6 22.0 22.1 22.5 20.0	$\begin{array}{c} Percent \\ -3.0 \\ 9.5 \\ 6.6 \\ 5.1 \\ 9.5 \\ 5.1 \\ 20.2 \\ 2.3 \\ 8.1 \\ 16.4 \\ 13.7 \\ 7.3 \\ 5.2 \\ 9.3 \\ 4.6 \\ 18.5 \\ 10.1 \\ 11.7 \\ 8.5 \\ 10.1 \\ 11.2 \\ 5.8 \\ 5.2 \\ 20.2 \\ 22.2 \\ 22.4 \\ 22.2 \\ 22.4 \\ 22.2 \\ 22.4 \\ 19.9 \\ 9.5 \\ 10.1 \\ $	$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	Percent -2.7 10.2 8 0 5.8 9.9 8.5 21.0 1.8 9.6 16.8 13.8 8.5 5 6.0 10.6 5.7 18.7 12.0 11.5 12.9 9.1 13.1 22.5 22.1 22.1 22.0 22.1 22.0 22.1 2.0 20.0 22.1 2.0 20.0 20	$\begin{array}{c} Percent \\ -0.9 \\ -0.9 \\ 12.4 \\ 9.4 \\ 5.8 \\ 11.5 \\ 2.0 \\ 10.7 \\ 17.3 \\ 15.2 \\ 9.8 \\ 8.1 \\ 13.3 \\ -6.7 \\ 19.5 \\ 19.0 \\ 13.8 \\ 13.6 \\ 14.3 \\ 10.5 \\ 22.2 \\ 22.2 \\ 22.2 \\ 22.0 \\ 11.2 \\ 22.2 \\ 22.0 \\ 12.2 \\ 22.0 \\ 12.2 \\ 22.0 \\ 12.1 \\ 12.1 \\ 21.1$
29	19.2	20.8	21, 3	21.4	21.6	41. 2	21.4	21. 4

loss in the two oven tests, and loss in the asphalticresidue test. Figure 5 indicates that for the slowcuring products the loss in 15 weeks' exposure was about  $2\frac{1}{2}$  times as great as that in the oven test on a 50-gram sample, about  $1\frac{1}{2}$  times as great as that in the oven test on a 20-gram sample, and about the same as the loss in the distillation test. No relationship was apparent between the loss in the exposure and asphalticresidue tests. The loss in the latter test was, however, invariably greater, ranging from  $1\frac{1}{3}$  to 14 times the loss occurring in 15 weeks' exposure with an average of  $2\frac{1}{2}$ times this loss. For the cut-back products the loss in all tests was approximately the same. In all the maof loss upon exposure and loss in the distillation test, | terials studied, both in 1932 and 1933, the total loss in

TABLE 6.—Resul	lts of tests on	exposure residues
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_				5-w	eek e	exposi	ıre							10-w	10-week exposure									15-w	eek e	xposu	ILO			
			Pen trati			Duct at 5 time per 1 ut	cen- ters nin-	ible in CS <sub>2</sub>	ible in CCI,	naphtha			Per trat			Duct at 5 time per 1 ut	cen- ters nin-	uble in CS <sub>2</sub>	ible in CCl4	naphtha			Per trat			Duct at 5 c time per r ut	ters	uble in CS <sub>2</sub>	ible in CCI4	naphtha
Sample identification	Loss	Float at 122° F.	At 77° F., 100 g, 5 sec.	At 32° F., 200 g, 60 sec.	Softening point	At 77° F.	At 34°-35° F.	Organic matter insoluble	Organic matter insoluble	Insoluble in 86° B. na	Loss	Float at 122 ° F.	At 77° F., 100 g, 5 sec.	At 32° F., 200 g, 60 sec.	Softening point	At 77° F.	At 34°-35° F.	Organic matter insoluble	Organic matter insoluble	Insoluble in 86° B. nar	Loss	Float at 122° F.	At 77° F., 100 g, 5 sec.	At 32° F., 200 g, 60 sec.	Softening point	At 77° F.	At 34"-35° F.	Organic matter insoluble	Organic matter insoluble	Insoluble in 86° B. nap
$\begin{array}{c}1\\2\\3\\4\\5\\6\\7\\1\\9\\2\\1\\1\\1\\2\\1\\3\\1\\4\\15\\16\\1\\7\\18\\19\\20\\21\\22\\23\\24\\25\\26\\27\\28\\29\end{array}$	$\begin{array}{c} Pct. \\ -2.6 \\ 9.5 \\ 6.6 \\ 6.1 \\ 1.9 \\ 5.6 \\ 20.2 \\ 2.3 \\ 8.1 \\ 1.6.4 \\ 13.7 \\ 7.3 \\ 3.4 \\ 6.6 \\ 18.5 \\ 8.2 \\ 10.1 \\ 1.1 \\ 7.3 \\ 3.4 \\ 6.6 \\ 18.2 \\ 2.2 \\ 2.2 \\ 2.2 \\ 2.2 \\ 4.2 \\ 2.2 \\ 4.1 \\ 1.1 \\ 7.6 \\ 1.1 \\ 1.1 \\ 1.1 \\ 7.3 \\ 1.2 \\ 5.5 \\ 1.2 \\ 1.$	$\begin{array}{c} 153\\ \hline 200\\ 34\\ 123\\ 163\\ 1,000+\\ \hline 131\\ 110\\ 129\\ 105\\ 150\\ 70\\ \hline 197\\ \hline 179\\ 106\\ 150\\ 116\\ \hline \end{array}$	86           152           300+           189              51           156              21           29           18           21           22           23	39 	95 106 96 110 106  136 93 93 98 95  136 155 151 138 151	95.0	Cm 1.5 4.0              	2. 93 1. 28  3. 22  1. 63  1. 81 4. 60       	8. 44 5. 95  25 7. 51  5. 69  7. 57 11. 69   15 5. 28 5.00 2.01 2.21 2.23	$\begin{array}{c} 28.0 \\ 25.6 \\ 25.9 \\ 15.3 \\ 21.0 \\ 22.5 \\ 9 \\ 21.0 \\ 22.5 \\ 9 \\ 21.0 \\ 22.5 \\ 9 \\ 21.8 \\ 22.5 \\ 9 \\ 21.8 \\ 22.5 \\ 9 \\ 21.8 \\ 22.5 \\ 9 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 21.4 \\ 22.5 \\ 9 \\ 22.8 \\ 22 \\ 22.8 \\ 22 \\ 22.8 \\ 22 \\ 22$	$\begin{array}{c} 8,0\\ 5,8\\ 9,9\\ 9,9\\ 9,9\\ 7,4\\ 21,0\\ 1,8\\ 8,6\\ 16,8\\ 21,6\\ 13,8\\ 8,5\\ 6,0\\ 13,8\\ 8,5\\ 7,7\\ 12,0\\ 11,5\\ 12,9\\ 9,1\\ 13,1\\ 22,5\\ 22,0\\ 22,1\\ 28,3\\ 20,0\\ \end{array}$	1,000+ 203 218 170	33         62           112         97           300+         95           73         89           126         105           36         152           44         136           157         27           12         25           20         12           16         18	$\begin{array}{c} 15\\ 22\\ 33\\ 37\\\\\\\\\\\\\\$	127 134 118 93 129 129 129 129 129 121 105 119 103 117 101 170 105 121 112 95	3.5 1.0 10.0 17.0	.0 .0 3.8 2.3 .5 3.5 1.5 3.5 2.5 3.5 3.5 3.5 3.5 3.5 3.5 3.5 3.5 3.5	4. 46 2. 46 5. 16 3. 08 3. 17	9. 66 16. 02  2. 91 . 47 . 43 . 41 2. 26	$\begin{array}{c} 31.3\\ 29.3\\ 30.5\\ 22.1\\ 17.0\\ 22.1\\ 22.5\\ 23.2\\ 29.4\\ 42.9\\ 33.2\\ 23.9\\ 23.9\\ 23.2\\ 33.5\\ 33.5\\ 23.7\\ 22.2\\ 33.2\\ 8.7\\ 22.2\\ 33.2\\ 8.7\\ 33.5\\ 33.9\\ 33.9\\ 33.5\\ 3$	$\begin{array}{c}9.4\\5.88\\11.5\\8.2\\2.1.5\\2.0.0\\10.7\\12.2.3\\22.3\\15.22.9\\8.1\\13.37\\6.7\\19.5\\19.0\\13.86\\11.2\\22.5\\22.2\\22.0\\22.6\\22.2\\22.0\\27.8\\22.6\\2.7,8\\22.6\\2.7,8\\2.0,1\\1\end{array}$	71	$\begin{array}{c} 197\\ 20\\ 49\\ 59\\ 52\\ 195\\ 300+\\ 199\\ 76\\ 51\\ 175\\ 72\\ 80\\ 64\\ 245\\ 29\\ 84\\ 166\\ 112\\ 49\\ 32\\ 49\\ 13\\ 24\\ 22\\ 112\\ 14\\ 17\\ 17\\ \end{array}$	$\begin{array}{c} 74\\ 9\\ 9\\ 15\\ 18\\ 23\\ 766\\ 104\\ 600\\ 47\\ 22\\ 84\\ 33\\ 30\\ 99\\ 92\\ 12\\ 28\\ 84\\ 8\\ 8\\ 28\\ 28\\ 48\\ 8\\ 28\\ 13\\ 13\\ 13\\ 5\\ 5\\ 8\\ 11\\ \end{array}$	$\begin{array}{c} 154\\ 144\\ 153\\ 135\\ 135\\ 135\\ 130\\ 115\\ 126\\ 112\\ 133\\ 109\\ 168\\ 120\\ 163\\ 111\\ 129\\ 160\\ 163\\ 161\\ 153\\ 168\\ 168\\ \end{array}$	$\begin{array}{c} Cm \\ 20.0 \\ 3.5 \\ 521.0 \\ 3.55 \\ .75 \\ 9.0 \\ 0.5 \\ .75 \\ 9.0 \\ 41.0 \\ 16.0 \\ 97.0 \\ 41.0 \\ 16.0 \\ 97.0 \\ 100 \\ 41.0 \\ 41.0 \\ 41.0 \\ 5.0 \\ 3.8 \\ 3.8 \\ 4.0 \\ \end{array}$	$ \begin{array}{c} 0 \\ 0 \\ 0 \\ 0 \\ 5 \\ 6.5 \\ 2.0 \\ 0 \\ 3.0 \\ 4.5 \\ 3.0 \\ -5 \\ 5.0 \\ 0 \\ 4.3 \\ \end{array} $	3. 36 . 45 5. 48 3. 07	$\begin{array}{c} 12.08\\ 8.40\\ .17\\ .13\\ .12\\ .14\\ .12\\ .24\\ .12\\ .24\\ .13\\ .38\\ .61\\ .23\\ .38\\ .08\\ .9.88\\ .10\\ .18.15\\ .09\\ .09\\ .81\\ .09\\ .81\\ .73\\ .70\\ .51\\ .52\\ \end{array}$	$\begin{array}{c} Pct,\\ 23,0\\ 40,6\\ 34,6\\ 35,0\\ 25,23,25\\ 31,2\\ 224,7\\ 29,6\\ 33,2\\ 224,7\\ 33,2\\ 224,7\\ 33,2\\ 224,7\\ 33,2\\ 224,7\\ 33,2\\ 224,7\\ 33,2\\ 224,7\\ 33,2\\ 224,2\\ 33,4\\ 33,2\\ 223,2\\ 31,4\\ 34,2\\ 33,4\\ 231,4\\ 34,2\\ 34,2\\ 3$

<sup>1</sup> The 10- and 15-week exposure residues of this sample were nonhomogeneous.

the distillation test most nearly approximated the loss in 15 weeks' exposure.

At the end of 15 weeks' exposure the surfaces of the samples varied greatly in appearance, as shown by the typical photographs in figure 4. The appearance of sample 14 was typical of samples 1, 2, 3, 4, 5, 6, 7, 9, 12, 18, 19, 20, 21, and 22. However, there were some variations in actual appearance that could not be shown in a photograph. Samples 1, 2, and 3 had mottled, slightly greasy surfaces. Samples 4, 9, and 19 had mottled, slightly greasy surfaces that were slightly irides-cent. Samples 14, 21, and 22 had mottled, iridescent surfaces that were not greasy. Sample 5 had a uniformly mottled appearance neither iridescent nor greasy. Samples 6, 7, 12, 18, and 20 were smooth and glossy.

Sample 23 was mottled and slightly greasy, like samples 1, 2, and 3, but had some dull areas as shown. Sample 13, while smooth and glossy, had some dull areas as shown; sample 16 was similar although it was slightly mottled.

Samples 10, 11, and 15 were checked over most of the area of their surfaces; their condition is shown by the photograph of sample 15. The unchecked areas in samples 10 and 11 were glossy but in sample 15 the surface was neither dull nor glossy. Samples 8 and 17 had very rough surfaces, shrunken and pitted as shown by their photographs. The material in the bottom of the cracks was soft but the outer surface was very hard. The cut-back products were all very rough and wrinkled,

<sup>3</sup> None of the residues after exposure of this sample was homogeneous.

The material in the cracks was glossy while the outer surface was dull.

At the end of 15 weeks all of the cut-back products were exceedingly hard. As they were originally combinations of semisolid asphalt and volatile fluxes and became semisolid in the laboratory tests, it would be expected that a semisolid residue would rapidly develop upon exposure. It has been believed generally that the asphalt base used in cut-back materials is relatively resistant to changes caused by weathering and that after the cutting medium is removed the character of the material changes but little. The six cut-back materials used in this study, while losing but little weight after 2 days, were much harder and had much less ductility at the end of 5 weeks' exposure than the asphalt used as base material.

The slow-curing products varied as greatly in consistency as in the rate and amount of volatile matter given off during exposure. At the end of 5 weeks all of them were harder than the residues from the distillation and oven tests, but only 5 had a penetration at  $77^{\circ}$ F. of less than 200. The consistency varied from a float of 34 seconds at  $122^{\circ}$  F. to a penetration of 51 at 77° F. At the end of 15 weeks only 3 materials had a penetration at 77° F. of over 200 and one of these materials had disintegrated to such an extent that no true penetration test was possible. The consistency ranged from a float of 71 seconds at  $122^{\circ}$  F. to a penetration of 20 at 77° F. In the previous work, the as shown by the photographs of samples 26, 27, and 28. slow-curing materials all developed residues after 15

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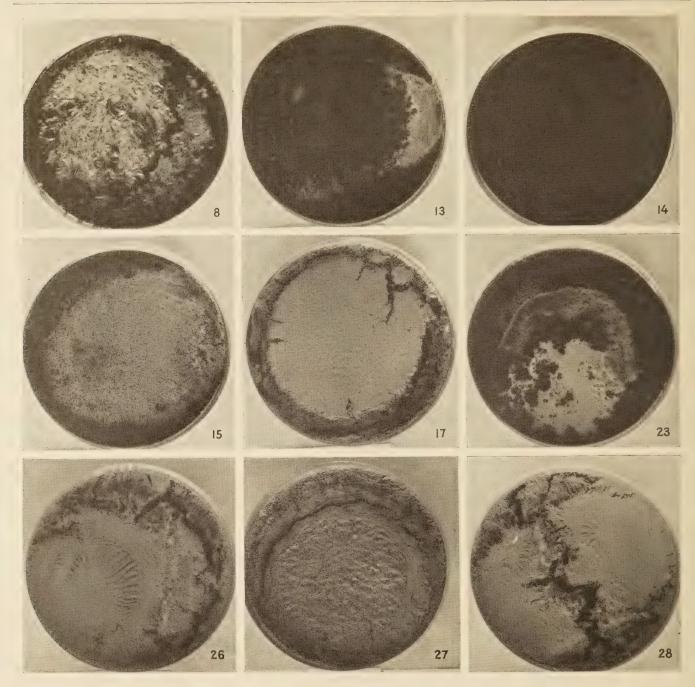


FIGURE 4.-CONDITION OF SURFACES AFTER 15 WEEKS OF EXPOSURE.

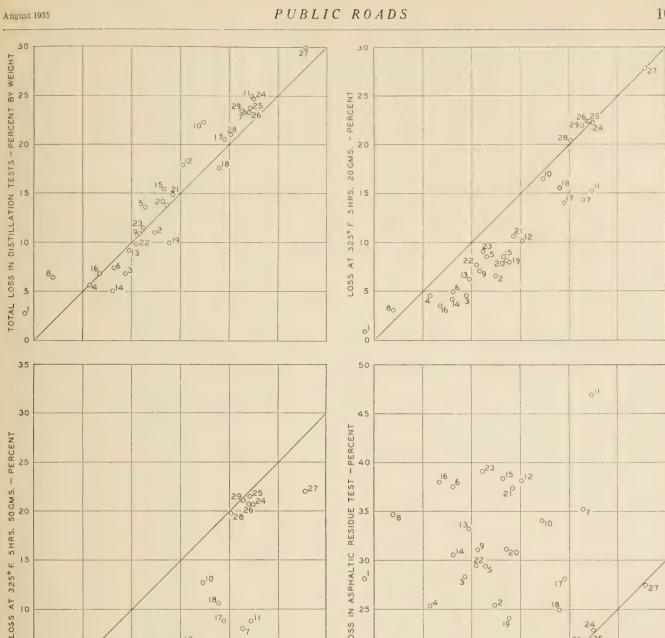
weeks that were as hard as or harder than their asphaltic residues of 100 penetration. It seems that it is impossible to predict, from the results of any of the laboratory tests, the consistency of the residues after exposure. Generally, however, those materials that took a long time to be reduced to 100 penetration in the asphalticresidue test and those whose residues from the distillation and oven tests had a low float-test value were the softest or most fluid at the end of 15 weeks.

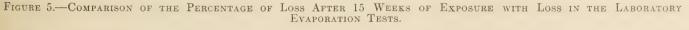
#### ASPHALTIC RESIDUES DIFFERED IN CERTAIN RESPECTS FROM RESIDUES AFTER EXPOSURE OF ABOUT THE SAME PENETRATION

A comparison of the residues after exposure and asphaltic residues is of interest. All but 5 of the asphaltic residues had ductilities at  $77^{\circ}$  F. of more than 110 cen-

timeters and 15 had ductilities at  $34^{\circ}-35^{\circ}$  F. of  $3\frac{1}{2}$  centimeters or more. After 15 weeks' exposure only 7 products had ductilities at  $77^{\circ}$  F. over 50 and only 4 over 110. Only 6 had ductilities at  $34^{\circ}-35^{\circ}$  F. over  $3\frac{1}{2}$ . These differences in ductilities may, in a number of instances, have been caused by differences in the consistencies of the residues after exposure and asphaltic residues.

However, in some cases one of the residues from exposure had approximately the same penetration as the asphaltic residue and the laboratory residues and residues after exposure may be compared directly. Table 7 shows the results of tests on both such residues and indicates that the two residues from the same material, although having the same consistency, varied considerably in other respects.





30

IN ASPHALTIC

LOSS

30

0

25

20

0

EXPOSED FOR 15 WEEKS - PERCENT

5

was always lower for the residue from exposure. F The ductility at 77° F. of the residue from exposure was less than that of the asphaltic residue in all but two cases, and for these the ductility of both residues was 110+. In five cases the ductility at  $34^{\circ}-35^{\circ}$  F. of the residue from exposure was greater than that of the asphaltic residue. In every case except two the percentage of material insoluble in naphtha was greater for the residue from exposure than for the asphaltic residue. In every case where there was an appreciable

15

13 60094 °16

10

5

010

18

011

25

IN FILM

07

20

LOSS

The ratio of the penetration at  $77^{\circ}$  F. to that at  $32^{\circ}$  and carbon disulphide, the percentages were much greater for the residues from exposure.

15

19

22 0

17

18

28

20

25

27

30

30

10

Figure 6 shows the development of free carbon, carbenes, and asphaltenes in laboratory and exposure tests for the samples that originally contained or finally developed carbenes in appreciable amounts. Samples 8, 23, and 27 developed only relatively small amounts of carbenes and the development of the insoluble constituents is shown only for sample 27. In figure 6 the volatile matter and the material insoluble in carbon disulphide, carbon tetrachloride, and 86° B. naphtha amount of material insoluble in carbon tetrachloride are plotted for the original materials and their residues.

15

10

5

0

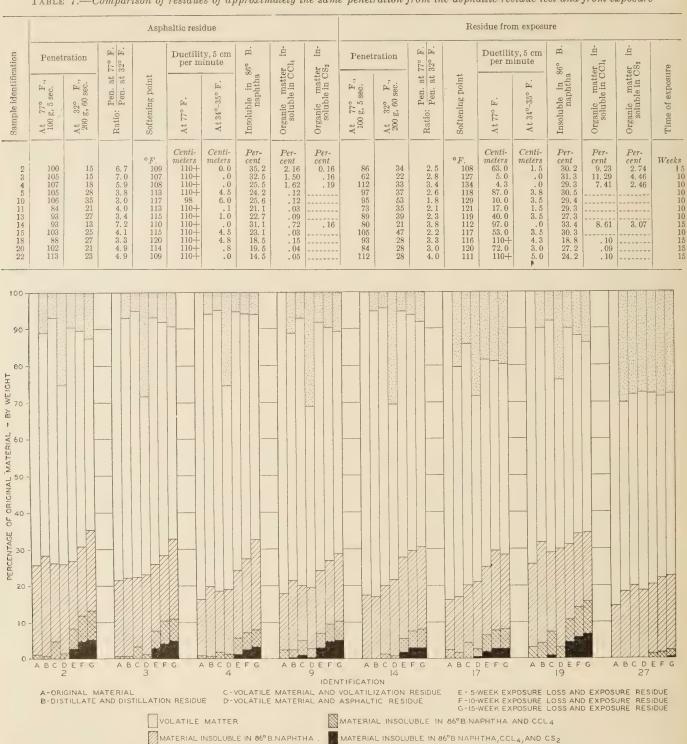


TABLE 7 .- Comparison of residues of approximately the same penetration from the asphaltic-residue test and from exposure

FIGURE 6.—COMPOSITION OF SELECTED MATERIALS AND THEIR RESIDUES AS DETERMINED BY SOLUBILITY TESTS.

All percentages are expressed in terms of the weight of insoluble in 86° B. naphtha. The remainder of the column represents material soluble in 86° B. naphtha

The solid portion in each vertical column represents free carbon or material insoluble in carbon disulphide. The remainder represents bitumen or material soluble in carbon disulphide. The double cross-hatched portion represents carbenes or bitumen insoluble in carbon tetrachloride, and the single and double cross-hatched portions together represent asphaltenes or bitumen insoluble in 86° B. naphtha. The remainder of the column represents material soluble in 86° B. naphtha that, in the original material, includes the more volatile hydrocarbons vaporized and lost under test conditions, as shown by the dotted area. The materials soluble in 86° B. naphtha have been termed "malthenes" by Richardson.<sup>3</sup> This designation, however, has not been generally accepted in the United States.

<sup>3</sup> Clifford Richardson, The Modern Asphalt Pavement, p. 544 (2d ed.).

As shown by figure 6, the material insoluble in naphtha included carbenes and free carbon; in fact, in the residue from 15 weeks' exposure of sample 19 the material insoluble in naphtha contained 19 percent free carbon and 27 percent carbones. The term asphaltenes also includes carbenes. Nellenstevn <sup>4</sup> has stated that asphaltenes, carbenes, and free carbon all consist of the same matter, that is, dispersed carbon in decreasing states of protection. The so-called "protective bodies" he designates as "micelles." He states that when extracted asphaltenes are heated at a temperature of 527° F. they will be changed largely to free carbon, but a normal asphaltic material can be heated at 662° F. for a long period with little production of free carbon. The reason for this is that in normal asphalts the amount of protective bodies, or micelles, is such that the decomposition of part of them influences their protective qualities only slightly. Marcusson <sup>5</sup> states that the material soluble in 86° B. naphtha (malthenes) is composed essentially of oily constituents and asphaltic resins. As the time of exposure increased, the percentage of asphaltenes, carbenes, and free carbon increased while the percentage of malthenes decreased.

In the 1932 investigation only those materials that had high specific gravities and initially contained some material insoluble in carbon disulphide with appreciable amounts insoluble in carbon tetrachloride developed carbenes either in laboratory or exposure tests. In the present study some of the materials such as samples 14 and 27, that originally had high solubilities in carbon disulphide and carbon tetrachloride, developed carbenes even during some of the laboratory tests. In those materials in which carbenes and free carbon were developed, it may be considered that the amount or the protective quality of the micelles was insufficient to prevent carbonization.

All of the rapid-curing products at the end of 15 weeks had developed residues containing about 0.5 percent of material insoluble in carbon disulphide and carbon tetrachloride. For all materials the solubility

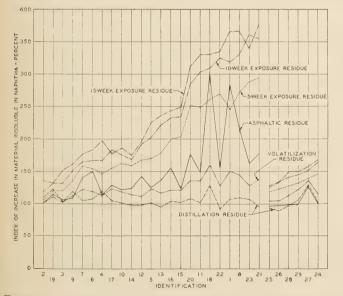


FIGURE 7.—INDEX OF INCREASE IN MATERIAL INSOLUBLE IN NAPHTHA IN THE VARIOUS RESIDUES.

<sup>4</sup> Report by F. J. Nellensteyn and R. Loman, Sixth Congress, Permanent International Association of Road Congresses, first section, second question, paper 2-0. <sup>4</sup> See Asphalts and Allied Substances by Herbert Abraham, third edition, p. 755.

in carbon disulphide was almost the same as the solubility in carbon tetrachloride, indicating the almost complete absence of carbones.

#### MATERIAL INSOLUBLE IN NAPHTHA DEVELOPED MOST UNDER EXPOSURE AND LEAST IN THE DISTILLATION TEST

In figure 6, where all the percentages are expressed in terms of the weight of the original material, if the percentage of material insoluble in naphtha in any residue is divided by the percentage of material insoluble in naphtha in the original material and multiplied by 100, the result is the index of increase in material insoluble in naphtha. An index of 100 therefore indicates no change in the amount of material insoluble in naphtha. This index for all the samples is plotted in figure 7 and shows that generally there was an increase in material insoluble in naphtha during the various tests. Generally, this index was least for the distillation test and greatest for the 15-week exposure test. In the distillation test it ranged from 88 to 129.

While inaccuracies in testing may account for indexes of less than 100 in the distillation residues, it is possible for indexes actually to be below 100. Several samples of very viscous, slow-curing asphaltic material, really semisolid asphalts, were recently subjected to the distillation test. The materials did not yield any distillate but there was considerable loss on cooling. Nevertheless, the residues were softer than the original materials and contained less material insoluble in naphtha.

In the oven-loss test on 20-gram samples the index varied from 101 to 161 and in the asphaltic residue test it varied from 96 to 302. In the case of the asphaltic residue the index was low when the time of reduction was low, and for the cut-back materials the asphaltic residue test was the least severe of all tests, although there is not much difference between this and the distillation test. The index was high when the time of reduction was high, as indicated by sample 18 which took 420 minutes to come to 100 penetration and had an index of 302, and by sample 1 which took 280 minutes and had an index of 284. The index varied from 105 to 294 for 5 weeks' exposure, from 119 to 361 for 10 weeks' exposure, and from 133 to 376 for 15 weeks' exposure. Except for four samples, the index of increase was greater for 5 weeks' exposure than for any of the laboratory tests. In the exposure tests, products originally containing the highest percentages of material insoluble in naphtha generally had the smallest indexes of increase.

### OLIENSIS TEST INVESTIGATED

An interesting development in the study of asphaltic materials is the test for determining heterogeneity. This test has been called the Oliensis or spot test. The method of making the test and the interpretation of the results were outlined in a paper read before the 1933 meeting of the American Society for Testing Materials. In making this test on the original material, one part by volume of the asphaltic material was treated with 5.1 parts by volume of a special naphtha at such a temperature that solution or dispersion was complete in 6 to 8 minutes. After cooling to room temperature and adding fresh naphtha to replace any losses, a drop of the mixture was allowed to drop on a filter paper (J. H. Munktells, No. OO). The appearance of the resulting stain varied from a uniformly colored spot, indicating complete dispersion, to stains in which the center was black and rough and surrounded by a lightercolored ring. When an entirely uniform stain was obtained another test was made after the mixture had stood for 24 hours. The appearance of the central spot was taken as an indication of the degree of heterogeneity, and those materials that gave a uniform stain after standing 24 hours were considered as homogeneous.

The test is presumed to give an insight into the conditions of manufacture. The types of asphaltic products that should be expected to appear homogeneous are as follows:

1. Steam-refined residuals known to have been refined without serious cracking.

2. The bitumen of certain native asphalts.

3. Some types of slightly oxidized residuals from asphaltic-base crude oils.

The materials that should be expected to appear heterogeneous are:

1. Steam-refined residuals that have been overheated during the refining process.

2. Cracking-coil residuals.

3. Highly blown residuals

The test was initially developed and used to determine whether or not petroleum asphalts had been subjected to higher temperatures than usually occur in steam refining. It has been used by some States as an identification test for the control of liquid asphaltic materials. Recently the test procedure has been standardized by a group of Middle-Western States. In applying the test to slow-curing materials they state that if the material has less than 15 percent of distillate by volume at  $680^{\circ}$  F., the test may be made on the original material. For all other materials of the slow-, medium-, or rapidcuring classes the test shall be made on the distillation residue. The presence of the volatile distillate in these types of materials was thought to interfere with the sensitivity of the test. It will be noted that in the case of the slow-curing materials investigated the test may be run on 18 samples as received, since in only 5 cases was the percentage of distillate by volume at 680° F. more than 15 percent.

The identification of the character of the manufacturing process was the main object of this test, and for this purpose it should be made only on the finished products as they leave the refinery and not after they have been subjected to various laboratory heat tests. However, in order to determine if the character of the materials underwent a change during the laboratory tests and under exposure, the residue from the distillation test, the 20-gram oven-loss test, and the asphalticresidue test, as well as the residues from the 5-week and 15-week exposures were subjected to the Oliensis test.

For the original materials 5.1 parts of naphtha by volume were mixed with 1 part of asphaltic material. For the various residues the volume of naphtha was kept constant and the weight of a unit volume of the original material minus the weight of volatile matter that occurred in producing the residue was used. Table 8 shows the ratio of naphtha to asphaltic material used for each sample. Only in the case of the asphaltic residue of sample 11 did the ratio of naphtha to asphaltic material exceed 7 to 1 by weight, the proportion that was being used by one State at the time these tests were made. It is thought that the variations in proportions of naphtha used in this work were not sufficiently wide to affect the character of the stains

#### TABLE 8.—Ratio of naphtha to asphaltic material in Oliensis test

	Origi- nal ma-		lation due	20-gram loss	Asphal- tic resi-	Residu exposu	
Sample identification	terial <sup>1</sup> (by weight)	By weight	By volume	residue (by weight)	due (by weight)	5 weeks (by weight)	15 <mark>weeks</mark> (by weight)
1	$\begin{array}{c} 4.10\\ 3.74\\ 3.74\\ 3.79\\ 4.06\\ 4.11\\ 4.13\\ 3.83\\ 4.10\\ 3.91\\ 3.410\\ 3.91$	$\begin{array}{c} 4.\ 21\\ 4.\ 20\\ 4.\ 01\\ 4.\ 01\\ 4.\ 58\\ 5.\ 37\\ 4.\ 58\\ 5.\ 28\\ 5.\ 28\\ 5.\ 28\\ 5.\ 28\\ 5.\ 28\\ 5.\ 28\\ 5.\ 28\\ 5.\ 28\\ 4.\ 99\\ 4.\ 31\\ 3.\ 92\\ 4.\ 77\\ 4.\ 32\\ 5.\ 02\\ 4.\ 78\\ 4.\ 78\\ 4.\ 78\\ 4.\ 78\\ 4.\ 48\\ 4.\ 41\\ 5.\ 40\\ 5.\ 48\\ 5.\ 48\\ 5.\ 48\\ 5.\ 48\\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 48\\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\$	5. 83 5. 51 6. 04 6. 90 5. 86 6. 84 6. 94 6. 38 5. 68 6. 18 6. 18 6. 42 5. 76 6. 05 6. 05 5. 70 5. 86 7. 40 7. 24 7. 24	$\begin{array}{c} 4.12\\ 4.01\\ 3.92\\ 3.96\\ 4.33\\ 4.26\\ 4.30\\ 4.27\\ 4.61\\ 4.92\\ 4.61\\ 4.57\\ 4.17\\ 9.3\\ 4.92\\ 4.52\\ 4.52\\ 4.52\\ 4.53\\ 4.53\\ 4.53\\ 4.53\\ 4.38\\ 4.22\\ 5.38\\ 5.36\end{array}$	$\begin{array}{c} 5.\ 69\\ 5.\ 00\\ 5.\ 22\\ 5.\ 61\\ 6.\ 49\\ 6.\ 32\\ 5.\ 55\\ 6.\ 22\\ 7.\ 68\\ 5.\ 86\\ 5.\ 86\\ 5.\ 53\\ 6.\ 518\\ 5.\ 52\\ 4.\ 73\\ 5.\ 95\\ 6.\ 518\\ 5.\ 76\\ 6.\ 427\\ 5.\ 34\\ 1.\\ 5.\ 5.\ 5.\\ 5.\ 5.\ 5.\\ 5.\ 5.\ 5.\\ 5.\ 5.\ 5.\\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\$	$\begin{array}{c} 4.10\\ 4.13\\ 4.00\\ 3.99\\ 4.37\\ 4.35\\ 5.15\\ 4.20\\ 4.16\\ 4.90\\ 4.91\\ 4.75\\ 4.21\\ 3.4\\ 4.90\\ 4.91\\ 4.70\\ 5.07\\ 4.06\\ 4.65\\ 4.65\\ 4.65\\ 4.42\\ 4.22\\ 5.36\\ 5.222\\ 5.36\\ 5.37\end{array}$	$\begin{array}{c} 4.\ 10\\ 4.\ 27\\ 4.\ 13\\ 4.\ 02\\ 4.\ 42\\ 5.\ 24\\ 4.\ 22\\ 4.\ 22\\ 4.\ 22\\ 4.\ 23\\ 4.\ 96\\ 5.\ 33\\ 4.\ 96\\ 5.\ 10\\ 4.\ 23\\ 4.\ 76\\ 4.\ 76\\ 4.\ 5.\ 36\\ 5.\ 5.\ 5.\\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\ 5.\$
27. 28. 29.	3. 99 4. 14 4. 11	5. 72 5. 24 5. 38	7. 98 6. 90 7. 29	5. 54 5. 20 5. 24	5. 50 5. 15 5. 24	5. 52 5. 17 5. 24	5. 54 5. 18 5. 24

<sup>1</sup> All original material 5.1 naphtha to 1 of sample by volume.

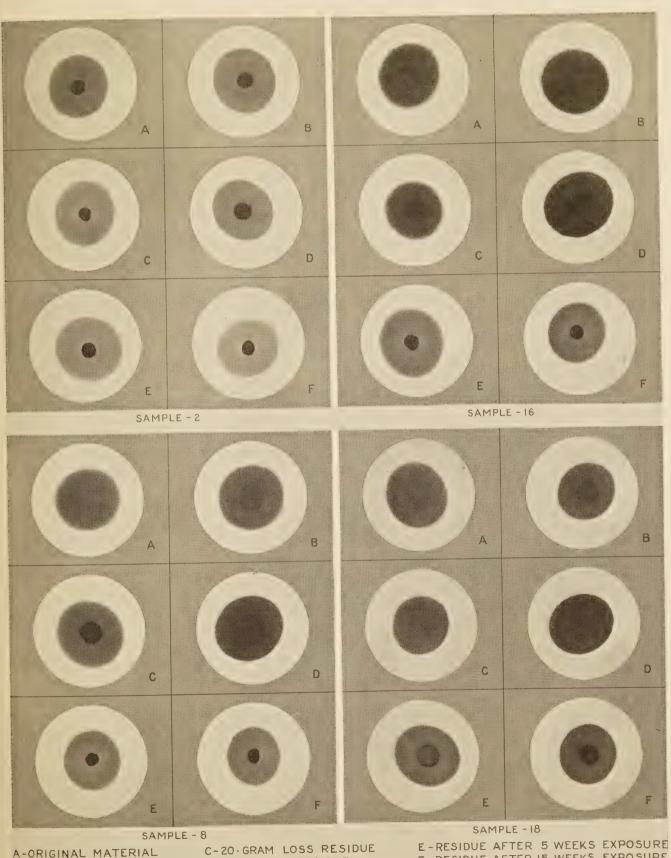
TABLE 9.—Character of original materials and residues as determined by the Oliensis test <sup>1</sup>

Sample identi- fication	Original material	Distilla- tion residue	20-gram loss residue	Asphaltic residue	5-week exposure residue	15-week exposure residue							
$\begin{array}{c} 1 \\ 2 \\ 2 \\ 3 \\ 3 \\ 3 \\ 5 \\ 5 \\ 5 \\ 5 \\ 6 \\ 7 \\ 8 \\ 8 \\ 9 \\ 10 \\ 10 \\ 10 \\ 11 \\ 12 \\ 13 \\ 14 \\ 15 \\ 16 \\ 17 \\ 18 \\ 19 \\ 20 \\ 20 \\ 21 \\ 22 \\ 22 \\ 23 \\ 24 \\ 25 \\ 26 \\ 27 \\ 28 \\ 29 \\ 29 \\ 29 \\ 20 \\ 21 \\ 22 \\ 23 \\ 24 \\ 25 \\ 26 \\ 27 \\ 28 \\ 29 \\ 29 \\ 20 \\ 21 \\ 21 \\ 22 \\ 21 \\ 22 \\ 22 \\ 23 \\ 24 \\ 23 \\ 29 \\ 29 \\ 29 \\ 20 \\ 21 \\ 21 \\ 21 \\ 21 \\ 21 \\ 21 \\ 21$	ННННО SHO Н O Н O Н H O SH H H H H O SHO H O H O H O H O H	НННННО ОННИНИИ SEH О ОННО ОННО ОННО ОННО ОННО ОННО ОННО	ННННН OOHHHH SHHHHHSHHOOOOHHHO SHHHOHOOOOHHO SHHOHOO HHOH	НННННО ОННО НИЗИНИ ОНО ООННО ОНО НО Н	ннннннннннннннннннннннннн зороосоосоосоосоосоосоосоосоосоосоосоосоо	нннннннннннннннннннннннннн залаан зан							

<sup>1</sup> H=Heterogeneous; O=Homogeneous; SH=Slightly heterogeneous.

obtained. Table 9 gives a classification of the stains and figures 8, 9, and 10 show stains typical of those obtained with the various samples and their residues.

Since the results of the tests were based upon the appearance of the stain as interpreted by the observer, it is difficult if not impossible to distinguish between border-line materials or to express clearly the apparent degree of heterogeneity that may be indicated by the varying degrees of nonuniformity in the stain. The classification given in table 8 should be understood to mean that, in the judgment of the observers, the materials and their residues gave stains that appeared either entirely uniform throughout or were only slightly non-



A-ORIGINAL MATERIAL B-DISTILLATION RESIDUE

FIGURE 8.-TYPICAL OLIENSIS STAINS.

D-ASPHALTIC RESIDUE

F-RESIDUE AFTER 15 WEEKS EXPOSURE

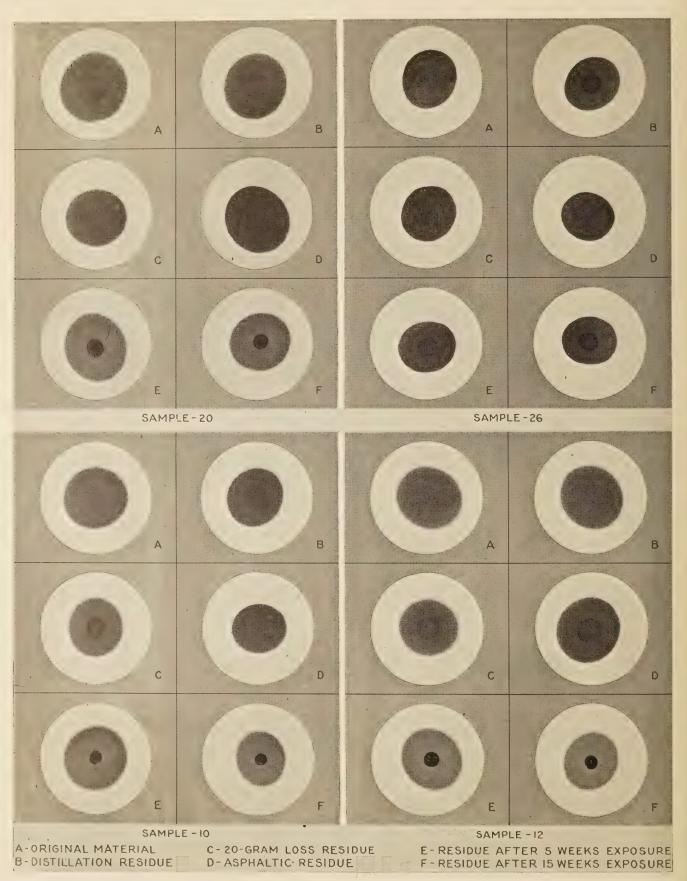


FIGURE 9.-TYPICAL OLIENSIS STAINS.

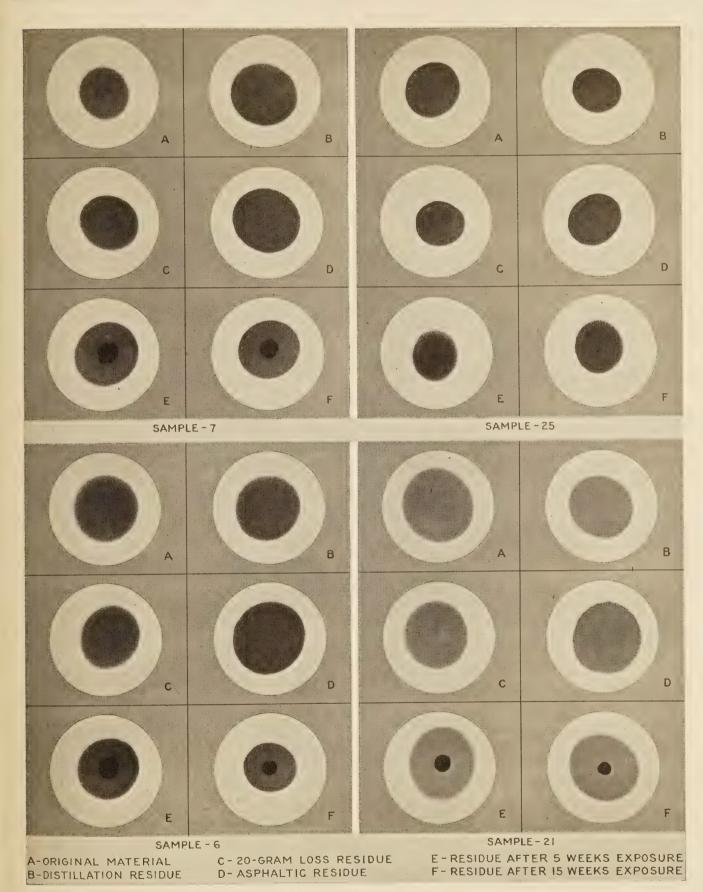


FIGURE 10.-TYPICAL OLIENSIS STAINS.

uniform, having a slightly darker, more-pronounced center, or else they had a definite dark to black center surrounded by a uniformly lighter-colored stain and were consequently classified as homogeneous, slightly heterogeneous, and heterogeneous, respectively. No attempt was made to indicate the extent or degree of heterogeneity other than to classify as slightly heterogeneous those materials and residues giving stains only slightly nonuniform.

#### ALL EXPOSURE RESIDUES FOUND TO BE HETEROGENEOUS

A study of the results of the Oliensis test shows that of the 29 materials, 10 were homogeneous, 4 slightly heterogeneous, and 15 definitely heterogeneous in their original state. All the residues from exposure were heterogeneous or slightly heterogeneous. Of the 10 materials appearing homogeneous in their original state, 5 developed homogeneous residues in all 3 laboratory loss tests, 2 developed slightly heterogeneous residues in all 3 tests, 2 developed slightly heterogeneous residues in the oven test and distillation test but not in the asphaltic-residue test, and 1 developed slightly heterogeneous residues in the oven test, and asphaltic-residue tests but not in the distillation test.

Of the 4 materials appearing slightly heterogeneous in their original state, 1 developed homogeneous residues in all 3 laboratory tests, 2 developed homogeneous residues in the asphaltic-residue and loss tests but remained heterogeneous in the distillation test, and one material remained heterogeneous in all tests. The 15 materials that appeared heterogeneous in their original state all developed heterogeneous residues.

A comparison of these materials according to the character of the stains produced in the laboratory and their behavior under laboratory and exposure conditions may be of interest. Nine of the materials, samples 2, 3, 4, 9, 14, 17, 19, 23, and 27, were heterogeneous originally and had heterogeneous residues. The photograph of sample 2 in figure 8 is typical. All of the above samples, except 23 and 27, showed micropscopic flecks as shown in figure 2 when examined under the microscope. All except samples 14 23, and 27 had appreciable amounts of material insoluble in carbon tetrachloride in the original sample. Samples 14, 23, and 27 developed carbenes at the end of 15 weeks of exposure and the other samples showed an increase in carbenes and free carbon.

All of the nine samples, except sample 23, had high percentages of material insoluble in naphtha. All had specific gravities greater than 1.01, except sample 27. This material has a very high specific gravity for a fluid cut-back and its behavior and characteristics place it in this group having high specific gravities. The asphaltic residues of the 9 samples showed the effect of changes in temperature, having low penetration at 32° F. and no ductility at 34°-35° F., although all had ductilities of over 110 at 77° F.

After 15 weeks' exposure all the materials were very hard, except sample 9, which, at the end of 5 weeks, had separated into two parts, one hard and brittle, the other soft and oily. It was impossible to flux these two parts so that, while having a float at  $122^{\circ}$  F. of over 1,000 seconds at the end of 5 weeks, it was impossible to get a penetration until the end of 15 weeks. When running the softening point test on the residues of this sample, after exposure, the material did not flow slowly to the bottom but dropped immediately at the temperature

sample but the results are unimportant as no true ductility value was obtained. The residues after 15 weeks' exposure of samples 2, 3, 4, 17, 23, and 27 had very low ductility at  $77^{\circ}$  F. but samples 14 and 19 had ductilities at  $77^{\circ}$  F. of 97 and 110+, respectively. None of the residues had any ductility at 34°-35° F

Three of the materials, 5, 11, and 13, gave Oliensis stains identical with those of the preceding samples. They did not, however, have as high specific gravities, were clear in the microscopic test, and did not, even at the end of 15 weeks, develop any carbenes. Their asphaltic residues had some ductility at  $34^{\circ}-35^{\circ}$  F. and, at the end of 15 weeks, while having low ductility at  $34^{\circ}-35^{\circ}$  F. at 77° F., samples 5 and 13 had some ductility at  $34^{\circ}-35^{\circ}$  F.

The remaining 17 materials all had low specific gravities, were clear in the microscopic test, did not have carbenes, and, with 1 exception, did not develop them. They showed various types of stains in the Oliensis test although all were heterogeneous at the end of 5 weeks. Their asphaltic residues varied in ductility and effect produced by changes in temperature as did their residues after exposure.

The stains of samples 1, 24, and 29, while resulting in photographs similar to those of the high-gravity materials, did not have the nuclei as raised or as rough as the stains of the materials with high specific gravities. The producers of the cut-back asphalt designated as sample 24 stated that their plant had no cracking equipment. This material, therefore, had evidently become heterogeneous in the refining process because of overheating, since the residue obtained in laboratory tests had good ductility, indicating that it had not been overblown. Sample 29 likewise was a cut-back asphalt with a ductile base. Sample 16 was slightly heterogeneous originally, produced slightly heterogeneous residues by distillation and volatilization, and produced a heterogeneous asphaltic residue. The asphaltic residues of these four samples were not especially affected by temperature change, having some ductility at 34°-35° F., although the asphaltic residues of samples 1 and 16 had comparatively low ductilities at  $77^{\circ}$  F. The residues of samples 1 and 16 after exposure had low ductility at  $77^{\circ}$  F. Those of samples 24 and 29 (cutback asphalts) were very hard and consequently had very low ductility.

#### OLIENSIS TEST MORE SENSITIVE THAN MICROSCOPIC TEST IN DETECTING OVERHEATED MATERIALS

Samples 8 and 15 were homogeneous but all of their residues from laboratory tests were slightly heterogeneous. The asphaltic residue of sample 8 had relatively low ductility at 77° F. but that of sample 15 had good ductility. Both asphaltic residues had good ductilities at  $34^{\circ}$ - $35^{\circ}$  F. The residue of sample 15 after exposure had low ductility at 77° F. Sample 8, the only material of low specific gravity to develop carbenes, separted in the exposure test in the same manner as sample 9.

Samples 18 and 20 were both slightly heterogeneous but their asphaltic and loss-test residues were homogeneous. Their asphaltic residues and their residues after exposure had good ductility at 77° F. Although the asphaltic residue of sample 20 had a ductility of only three-fourths centimeter at 34°-35° F., the ductility of its residue after exposure, as well as the ductility of the two residues of sample 18, was good at 34°-35° F. reported in table 5. Ductility tests were made on this Sample 18 developed a residue at the end of 5 weeks'

exposure that was the least heterogeneous of any of the residues from the slow-curing materials.

Samples 10 and 26 and their asphaltic residues were homogeneous. The asphaltic residues of both samples had good ductility at 77° F. and  $34^{\circ}-35^{\circ}$  F. At the end of 15 weeks, sample 10 had low ductility at 77° F. but good ductility at  $34^{\circ}-35^{\circ}$  F., while sample 26, a cut-back asphalt, was extremely hard and nonductile.

Sample 12 was homogeneous as was its residue after distillation. Its asphaltic residue had good ductility at 77° F. and  $34^{\circ}$ - $35^{\circ}$  F. and the residue after exposure had fair ductility at 77° F. and good ductility at  $34^{\circ}$ - $35^{\circ}$  F.

Sample 7 was slightly heterogeneous but all of its residues from laboratory tests were homogeneous. Its asphaltic residue was ductile at 77° F. but only slightly so at  $34^{\circ}-35^{\circ}$  F. and its residue after 15 weeks of exposure had good ductility at 77° F. and at  $34^{\circ}-35^{\circ}$  F.

Samples 6, 21, 22, 25, and 28 were homogeneous with homogeneous residues from laboratory tests. All of their asphaltic residues had good ductility at 77° F. and all except those of samples 21 and 22, the California residuals, had good ductility at  $34^{\circ}-35^{\circ}$  F. At the end of 15 weeks, sample 6 was still fluid, while the cutback asphalt samples 25 and 28 were very hard and nonductile. Samples 21 and 22 had good ductility at 77° F. and at  $34^{\circ}-35^{\circ}$  F. although their asphaltic residues were nonductile at  $34^{\circ}-35^{\circ}$  F.

It is readily apparent that the laboratory tests did not produce residues that gave stains in the Oliensis test radically different from the stains of the original ma-terials. The behavior of the residues from exposure showed, as did the other tests, that outdoor exposure alters asphaltic materials far more than any of the laboratory heat tests. This was strikingly shown by the decidedly heterogeneous stains obtained with the residues from exposure, especially in the case of the materials originally homogeneous. It is not believed, however, that it is possible to predict the physical and chemical characteristics of the material after exposure from the results of the Oliensis test, whether made on the original material, the residues from laboratory tests or both. Residues having what are believed to be desirable qualities were obtained from both homogeneous and heterogeneous materials, although heterogeneous materials undoubtedly have a more pronounced tendency to carbonize and their slow-curing products generally develop a less-ductile residue.

For detection of materials that have been inadvertently or intentionally subjected to too high a temperature during the refining process, the Oliensis test seems to be more sensitive than the microscopic test. All of the materials that had the characteristics of overheated or cracked materials were heterogeneous in the Oliensis test but only seven of them showed microscopic flecks.

#### HUBBARD-FIELD STABILITY TEST USED TO MEASURE BONDING STRENGTH AND DEVELOPMENT OF BONDING STRENGTH UPON EXPOSURE

Cylinders were made according to the Hubbard-Field method and tested to determine the adhesiveness or bonding strength of the original material, the residue after distillation and the asphaltic residue, and the development of bonding strength by the original materials after exposure. The first series, for the determination of bonding strength, consisted of 3 sets of 3 cylinders each for each material. The first and second

sets contained 16.6 percent by volume of the original material and distillation residues respectively mixed with 83.4 percent of a standard sand. The third set contained the same percentage of asphaltic residue by weight as was contained in the cylinders made with the original materials that gave an almost constant percentage of bitumen by volume in the cylinders of this set. All cylinders of series 1 were tested immediately for stability at 77° F.

The second series of cylinders, for determination of the development of bonding strength, likewise consisted of 3 sets of 3 cylinders using the same aggregate used in the first series and the same percentage of the original materials by volume. These three sets were placed in the exposure boxes and subjected to the same exposure conditions as the thin films. One set was removed at the end of 5, 10, and 15 weeks. The cylinders were weighed before and after exposure and the loss in weight was expressed as a percentage of the bituminous material present in the cylinder as made. After weighing, the cylinders were tested for stability at 77° F.

For comparative purposes two additional sets of cylinders were made, using as a binder the amounts of distillation residue and asphaltic residue that would have been obtained if the bitumen in the cylinders containing the original material had been subjected to the distillation or asphaltic-residue test. The aggregate used was a Potomac River sand that had been separated on standard sieves and recombined to give the following grading:

I 61	rcent
Passing no. 10, retained on no. 20	3.7
Passing no. 20, retained on no. 30 1	0.3
Passing no. 30, retained on no. 40 1	8.1
Passing no. 40. retained on no. 50 2	1.3
Passing no. 50, retained on no. 80	6.6
Passing no. 80, retained on no. 100	6. L
Passing no. 100, retained on no. 200	3.2
Passing no. 200	. 7

This sand had a specific gravity of 2.666 and the voids in the mineral aggregate, determined on the compacted cylinders of both series, were 38 percent for the cylinders made with the original materials, 37.4 percent for the cylinders made with the distillation residue, and 36.9 percent for the cylinders made with the asphaltic residue.

The method of mixing and molding the cylinders was the same as that used in 1932. The results of the tests on the cylinders of series 1 and 2 are given in tables 10 and 11, respectively. All results are the averages of three tests.

The results of tests on the cylinders of series 1 are shown graphically in figure 11. The stability of the cylinders at 77° F. was plotted against the Furol viscosity at 122° F., and the results of the float test at 77° F. for the cylinders made with the original materials and against the float test results at 122° F. and the penetration at 77° F. for the cylinders made with the distillation residue. Since the asphaltic residues are all of approximately the same consistency, the stabilities were plotted for each sample independently.

Figure 11 shows that although the stability of the mixtures was roughly proportional to the consistency of the contained bitumen, materials having the same consistency as measured by viscosity at 122° F., float test at 77° F. and 122° F., and penetration at 77° F. had different stabilities. This was especially noticeable

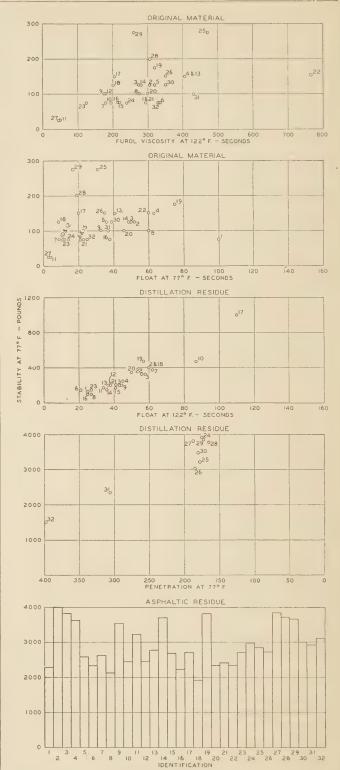
	Orig	rinal mat	erial	Disti	llation re	sidue	Asphalticresidue						
Sample iden- tification	Stabil- ity at 77° F.	Float at 77° F.	Furol viscos- ity at 122° F.	Stabil- ity at 77° F.	Float at 122° F.	Pene- tration at 77° F.	Stabil- ity at 77° F.	Pene- tration at 77° F.					
_		Seconds					Pounds						
1	75	100	294	125	24		2, 275	96					
2	125	52	303	400	60		3,975	100					
3	125	50	274	200	43		3,775	105					
4	150	63	404	250	45		3,600	107					
5	125 75	36 21	331 335	325 150	58 21		2, 575 2, 275	105 99					
6	75	9	181	375	62		2, 275	99					
8	100	60	277	100	27		2, 100	100					
9	100	33	175	200	44		3, 500	86					
10	75	11	197	475	87		2,425	106					
11	25	4	50	175	34		3, 175	84					
12	100	13	182	200	38		2,450	97					
13	150	41	406	200	37		2,775	93					
14	125	49	278	150	36		3,675	93					
15	75	21	219	175	41		2,675	103					
16	75	38	217	100	25		2, 225	109					
17	150	20	205	1,000	110		2,700	102					
18	125	9	199	400	60		1,900	100					
19	175	75	320	475	57		3,775	102					
20	100	46	300	350	50		2,325	104					
21	75	23	293	200	41		2,375	110					
22	150	60 12	768	325	56		2,325	113					
23	75 75	12	123 241	$150 \\ 3,925$	27		2,700	98					
25	275	31	472	3, 925		$\frac{76}{77}$	2,950	101					
26	150	35	351	3, 225		84	2,825	94 95					
27	25	3	47	3,800		87	3,850	90					
28	200	19	303	3,775		65	3, 850	90					
29	275	17	260	3, 850		72	3, 650	100					
30	125	39	350	3, 475		81	3,000	100					
31	100	37	431	2, 325		205	2,950	112					
32	75	25	331	1,500		298	3,075	101					
	10	20	001	2,000		200	0,010	101					

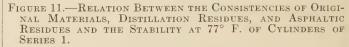
TABLE 10.—Results of tests on series 1 cylinders

TABLE 11.—Results of tests on series 2 stability cylinders

ication	Cyli	nders n	nade wi	th the c	made	lation	Cylir made asph resid	with altic			
dentif	S	tability	at 77° I	<u>?</u> .	Loss	of bitu	ımen	Sta-	Theo- reti-	Sta-	Theo- reti-
Sample identification	When made			In 15 weeks	In 5 weeks	In 10 weeks	In 15 weeks	bility at 77° F.	cal loss of bitu- men	bility at 77° F.	cal loss of bitu- men
$\begin{array}{c}1\\2\\3\\4\\5\\6\\7\\8\\9\\9\\10\\11\\12\\13\\14\\15\\16\\17\\18\\20\\21\\22\\23\\24\\25\\26\\27\\28\\8\\29\\30\\31\\32\end{array}$	Lbs. 75 75 125 125 125 75 100 125 75 100 125 75 100 125 75 150 125 75 150 125 75 150 125 75 150 125 125 100 125 75 100 125 125 100 125 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 100 125 125 100 125 100 125 155 100 125 155 100 125 155 100 125 155 150 125 155 155 155 155 155 155 155	Lbs. 3000 900 750 600 675 225 425 250 775 425 350 650 650 650 950 825 525 1, 660 725 525 1, 660 725 4, 275 3, 700 3, 600 4, 650 4, 650 5, 705 5,	$\begin{array}{c} Lbs.\\ 300\\ 300\\ 300\\ 300\\ 300\\ 300\\ 300\\ 30$	$ \begin{array}{c} Lbs. \\ 425\\ 1, 300\\ 850\\ 0 \\ 1, 150\\ 850\\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ $	Per- cent 0 7 6 3 6 4 4 5 3 7 7 14 18 8 4 4 15 16 6 4 4 8 4 4 15 16 6 7 7 8 8 8 5 7 7 20 0 17 14 15 15 15 16 17 14 15 15 16 16 17 16 17 16 16 17 16 17 16 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 16 17 17 17 16 17 17 17 17 17 17 17 17 17 17 17 17 17	Per- cent 3 3 1 10 9 9 9 9 20 5 12 18 8 21 14 14 10 0 8 8 19 9 20 20 12 12 18 8 11 11 12 3 3 9 9 11 11 22 4 30 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	Per- cent 1 1 11 6 11 11 6 12 12 18 8 222 15 15 10 0 8 8 13 3 9 21 20 20 20 20 20 20 20 20 20 20 20 20 20	$\begin{matrix} Lbs. \\ 125 \\ 425 \\ 5250 \\ 300 \\ 475 \\ 175 \\ 325 \\ 550 \\ 275 \\ 275 \\ 275 \\ 275 \\ 275 \\ 275 \\ 275 \\ 350 \\ 300 \\ 300 \\ 625 \\ 350 \\ 300 \\ 300 \\ 200 \\ 300 \\ 2800 \\ 2, 800 \\ 2, 800 \\ 2, 800 \\ 2, 800 \\ 2, 800 \\ 3, 400 \\ 3, 250 \\ 3, 400 \\ 3, 275 \\ 1, 425 \\ 1, 425 \end{matrix}$	$\begin{array}{c} Per-\\cent\\ 3\\ 11\\ 7\\ 7\\ 6\\ 6\\ 14\\ 22\\ 25\\ 25\\ 15\\ 18\\ 9\\ 9\\ 5\\ 15\\ 7\\ 21\\ 18\\ 10\\ 14\\ 15\\ 10\\ 12\\ 22\\ 22\\ 23\\ 30\\ 21\\ 24\\ 26\\ 20\\ 22\\ 22\\ \end{array}$	$\begin{array}{c} Lbs.\\ 1, 975\\ 3, 575\\ 3, 075\\ 3, 150\\ 2, 400\\ 1, 975\\ 2, 075\\ 2, 075\\ 2, 075\\ 2, 075\\ 2, 175\\ 2, 175\\ 2, 450\\ 2, 350\\ 2, 325\\ 1, 750\\ 2, 325\\ 1, 950\\ 2, 325\\ 1, 950\\ 2, 825\\ 2, 325\\ 1, 950\\ 2, 100\\ 2, 175\\ 2, 800\\ 2, 100\\ 2, 175\\ 3, 600\\ 3, 475\\ 3, 400\\ 2, 775\\ 3, 400\\ 2, 775\\ 2, 775\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 400\\ 2, 675\\ 2, 775\\ 2, 775\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 2, 775\\ 3, 600\\ 3, 475\\ 3, 400\\ 2, 675\\ 2, 775\\ 3, 75\\ 3, 400\\ 2, 775\\ 3, 400\\ 2, 775\\ 3, 400\\ 3, 400\\ 3, 400\\ 3,$	Per- cent 28 25 28 25 35 35 35 31 34 47 38 33 30 38 38 38 38 28 25 24 31 31 37 29 39 23 22 22 22 22 27 20 20 21 22 22 22 22 22 22 22 22 22 22 22 22

in the results with the cylinders made with the asphaltic residue. Although all of these residues had approximately the same penetration, the stability of the cylinders varied from 1,900 pounds for sample 18 to 3,975 pounds for sample 2.





It is seen that the cylinders made with the asphaltic residues of samples 2, 3, 4, 9, 11, 14, and 19 all had stabilities of over 3,000 pounds. All of these materials were heterogeneous originally, all were materials of high specific gravity, and all except sample 11 contained or developed carbenes and free carbon. Sample 11 had a relatively high specific gravity but did not develop

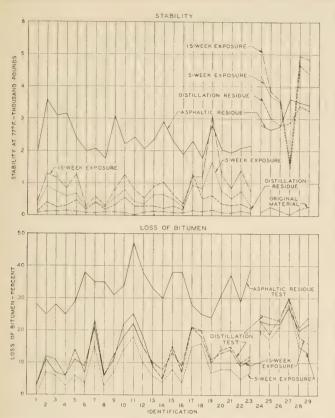


FIGURE 12.—COMPARISON OF LOSS OF BITUMEN AND STABILITY OF SERIES 2 HUBBARD-FIELD CYLINDERS.

carbenes. Samples 5, 7, 13, 15, 17, and 23 had stabilities between 2,500 and 3,000 pounds. Samples 17 and 23 were heterogeneous materials of high specific gravity that contained or developed carbenes. Samples 5 and 13 were heterogeneous materials of fairly high specific gravity but they did not develop carbenes, and samples 7 and 15 were materials of low specific gravity. All of the cut-back products had asphaltic residues giving stabilities of 2,500 pounds or over and five gave stabilities of 3,000 pounds or over.

Figure 12 shows the results of stability tests on the cylinders of series 2. The loss of bitumen in 5 and 15 weeks of exposure and the theoretical loss of bitumen in the cylinders made with the distillation and asphaltic residues were plotted for each sample. The stabilities at 77° F. for each sample were also plotted.

In this figure it is seen that, in the case of the slowcuring materials, although the loss of bitumen in the exposed cylinders was approximately the same as the loss in the distillation test, the exposed cylinders had greater stability than the cylinders made with the distillation residue except in the case of sample 17. This sample, even in 15 weeks, did not attain as high a stability as the cylinders made with the distillation residue. It is also seen that the loss in 15 weeks' exposure did not approach the loss in the asphaltic residue test and that the stability of the exposed cylinders did not approach the stability of the cylinders made with the asphaltic residue, except in the case of sample 19.

For the cut-back materials the indicated losses were probably in error due to unavoidable loss of volatile matter while mixing and molding the cylinders. The losses in 5 and 15 weeks of exposure probably should have been about the same as the losses in the distillation and asphaltic residue tests. The stabilities at 5 weeks

were higher than the stabilities of cylinders made with the asphaltic and distillation residues except in the case of sample 27.

#### SATISFACTORY CHECKS OBTAINED WITH RESULTS OF 1932 TESTS

After the exposure tests had been started, a question was raised concerning the use of plate glass covers for the exposure cabinets because it prevented the active ultra-violet rays from acting on the materials. Fused quartz glass not being available, Vita glass, which, after a short stabilization period, is guaranteed to permanently transmit an effective volume and combination of wave lengths of active ultra-violet light, was used to determine the effect of the passage of more active light. Duplicate sets of 10 of the slow-curing materials, 2 of the rapid-curing materials, and sample 27 and 3 new medium-curing materials, samples 30, 31, and 32, were exposed under both Vita and plate glass for 5- and 10-week periods. The materials were exposed in thin films and also admixed with the standard sand in the form of Hubbard-Field cylinders.

This exposure was started August 28, 1933. During the first 5-week period the average maximum air temperature was  $80^{\circ}$  F. and the number of sunlight hours was 266. During the 10-week period the average maximum air temperature was  $73^{\circ}$  F. and the number of sunlight hours was 512. The results of the tests on the thin films are given in table 12, and those on the Hubbard-Field cylinders in table 13.

As shown by tables 6 and 12, the materials did not lose as much nor get as hard in 10 weeks as they did in the original 5 weeks of exposure. This was due to the lower air temperature and also to the fact that the sun's rays striking at an oblique angle did not cause the material to get as hot as earlier in the summer. There was little if any difference between the materials exposed under the different types of glass. The samples exposed under Vita glass had a little more material insoluble in carbon disulphide and carbon tetrachloride and generally had a little more material insoluble in naphtha. Each solubility reported was the average of 3 or more tests. The results of the stability tests do not show that there was any difference between the two types of glass. The comparative study of the effectiveness of Vita glass and plate glass as cover for the exposure of the samples did not produce differences great enough to indicate their relative efficiency for this purpose.

As stated previously, four of the samples tested in 1932 were included in the 1933 work and the results obtained, as shown in table 14, were in remarkably close agreement with the previous tests. In the case of samples 17 and 19 the residues from the 1933 exposure tests had greater? percentages of free carbon and carbenes than did residues from the 1932 exposure tests.

The results of the two sets of stability tests were not in such close agreement because the aggregate used in the 1933 tests was somewhat coarser than that used in 1932. In 1932, cylinders made with sample 17 were the only ones that, after 15 weeks' exposure, had a stability about the same as those made with the distillation residue. In 1933 the cylinders made with sample 17, after 15 weeks' exposure, had less stability than the cylinders made with the distillation residue. Cylinders of sample 19 that in 1932, after 15 weeks' exposure, had a stability approaching that of the asphaltic-residue cylinders, in 1933, after 15 weeks' exposure, had a stability higher than that of the asphaltic-residue specimens.

#### TABLE 12.—Results of tests on plate and Vita glass exposure residues

5 WEEKS' EXPOSURE

					Plate	glass					Vita glass									
ntification	-	° F.	Penet	ration	point		ity at 5 minute	30	atter in- CCl 4	n 86° B. ha		۲. ۲.	Penet	ration	oint	Ductil cm per	ity at 5 minute	OS	matter in- in CCl 4	n 86° B. ha
Sample identification	Loss	Float at 122°	At 77° F., 100 g, 5 sec.		Softening p	At 77° F.	At 34°- 35° F.	Organic mat soluble in	Organic matter in soluble in CCl <sub>4</sub>	Insoluble in 86° naphtha	Loss	Float at 122°	At 77° F., 100 g, 5 sec.	At 32° F., 200 g, 60 sec.	Softening point	At 77° F.	At 34°- 35 °F.	Organic matt soluble in (	Organic me soluble in	Insoluble in 86° B naphtha
I2 13	4.5 2.2 12.4 16.0 6.4	Sec- onds 50, 113 61 53 140 105 103 53 52		18 14 17 16 18 24			Centi- meters	Per- cent 0.82 .29 1.08 1.68 	Per- cent 6.86 2.68 4.09 9.63 	Per- cent 12, 7 29, 9 18, 4 24, 8 26, 8 12, 2 32, 6 17, 2 14, 1 9, 2 27, 3 27, 7 33, 3 32, 7 29, 7 19, 8	Per- cent -3.3 4.5 2.2 1.6 6 12.4 14.9 5.6 6.5 5.7 5.6 21.5 26.2 18.8 22.1 17.3 17.5	Sec- onds - 46 92 54 47 156 104 47 156 104 49 98 96 49 50	33 39 59 40 52 52 123		°F.      	Centi- meters	Centi- meters	Per- cent 1.22 .47 1.28 2.00 	Per- cent 6.85 2.73 4.64 10.22 	Per- cent 13. 4 29. 2 17. 8 23. 4 28. 3 12. 9 33. 6 17. 0 13. 7 19. 2 27. 6 28. 0 28. 0 31. 7 33. 5 30. 5 20. 8
								10 WE	EKS' I	EXPOS	URE									
1         2           13         14           17         18           19         20           23         23           26         27           29         30           31         32	$\begin{array}{c} -2.5\\ 7.1\\ 5.1\\ 2.2\\ 12.9\\ 16.4\\ 7.6\\ 8.3\\ 8.7\\ 6.1\\ 21.4\\ 26.4\\ 19.3\\ 22.6\\ 17.8\\ 18.1 \end{array}$	54 152 85 64 178 144 145 135 700 71 71		17 13 17 14 14 14 22				1. 42 . 50 1. 03 2. 92 . 42 . 25	$\begin{array}{c} 0.\ 14\\ 7.\ 77\\ .\ 04\\ 3.\ 16\\ .\ 12\\ 10.\ 42\\ .\ 07\\ .\ 09\\ .\ 10\\ .\ 36\\ .\ 22\\ .\ 23\\ .\ 13\\ .\ 17\\ \end{array}$	$\begin{array}{c} 14.\ 1\\ 30.\ 2\\ 19.\ 6\\ 23.\ 9\\ 28.\ 1\\ 12.\ 9\\ 34.\ 3\\ 19.\ 4\\ 15.\ 0\\ 20.\ 4\\ 28.\ 0\\ 29.\ 9\\ 33.\ 4\\ 33.\ 9\\ 33.\ 7\\ 22.\ 2\end{array}$	$\begin{array}{c} -3.3 \\ 4.1 \\ 3.2 \\ 2.3 \\ 12.6 \\ 16.1 \\ 6.9 \\ 8.0 \\ 8.7 \\ 6.1 \\ 21.5 \\ 26.7 \\ 18.9 \\ 22.3 \\ 17.3 \\ 18.0 \end{array}$	56 141 71 65 297 154 133 125 74 84	30 26 52 34 44 73	     	149 130 136 131 117	   6 7 16 11 32 65	     	1. 73 . 55 1. 24 2. 99 	$\begin{array}{c} 0.\ 22\\ 8.\ 13\\ .05\\ 3.\ 45\\ 5.\ 39\\ .16\\ 10.\ 61\\ .11\\ .12\\ .08\\ .25\\ .59\\ .24\\ .20\\ .14\\ .15\\ \end{array}$	$\begin{array}{c} 14.\ 6\\ 30.\ 5\\ 20.\ 2\\ 24.\ 1\\ 29.\ 1\\ 13.\ 9\\ 33.\ 7\\ 19.\ 0\\ 15.\ 9\\ 21.\ 3\\ 28.\ 7\\ 31.\ 1\\ 33.\ 35.\ 3\\ 35.\ 3\\ 33.\ 0\\ 23.\ 7\end{array}$

#### CONCLUSIONS

The results of this investigation substantiate most of the conclusions arrived at in the 1932 investigation, modify two of the conclusions, and indicate some new conclusions. The conclusions substantiated are:

1. Materials of high specific gravity and their residues are, in general, more susceptible to changes in temperature than materials of low specific gravity and their residues.

2. Hardening due to causes other than loss of volatile matter, and changes in inherent characteristics that may be attributed to oxidation, polymerization, and carbonization, occur to the greatest extent upon exposure and least during distillation.

3. The development of a ductile residue either in the asphaltic-residue test or, in the case of cut-back materials, in the distillation test, does not indicate that the material will develop a ductile residue upon exposure.

4. The bonding strength of the original materials and their residues is roughly proportional to their consistencies, but materials having the same consistency as measured by the present tests do not always give the same stability. The reasons for these differences in stability cannot be determined under the present methods of testing.

The following conclusions are somewhat modified from 1932:

TABLE 13.—Results of stability tests on cylinders exposed under plate and Vita glass

		Loss of	bitumen			Stability	at 77° F		
Sample identi- fication	In 5	weeks	In 10	weeks	In 5 v	weeks	In 10 weeks		
	Plate Vita		Plate	Vita	Plate	Vita	Plate	Vita	
	Percent	Percent	Percent	Percent	Pounds	Pounds	Pounds	Pound	
1	2	1	2	1	225	250	275	37	
2	8	5	8	7	850	700	850	80	
13	7	4	6	5	450	450	500	42	
14	4	2	4	3	475	. 400	500	45	
17	13	12	15	13	650	700	650	70	
18	18	16	18	17	500	500	525	50	
19	8	6	8	8	1,075	1,000	1,250	1, 27	
20	8	7	8	8	550	500	500	52	
21	8	7	8	8	325	350	350	35	
23	5	4	6	5	225	200	250	25	
26	21	21	24	23	2,775	2,650	3, 550	3, 50	
27	28	28	30	30	1, 225	1, 225	1,575	1, 35	
29	12	12	14	13	3,775	3, 725	4,600	4, 57	
30	23	23	24	24	2, 575	2,550	2,850	3, 02	
31	18	19	20	20	2, 325	2, 375	2,725	3, 12	
32	18	17	19	20	1,675	1,725	2,200	2,67	

5. The relative rates of volatilization of the various materials can be anticipated most readily from the distillation curves. The different classes of material may be differentiated in the loss and asphaltic-residue tests, especially if the time of reduction to 100 penetration is considered. However, sharp distinctions in initial curing properties, that may be of importance in some types of construction, can be determined only from the distillation curves. 6. Carbonization generally occurs in materials that originally contain some material insoluble in carbon disulphide and carbon tetrachloride, but some materials with exceptionally high solubility in these solvents show a tendency to carbonize under both laboratory and exposure conditions.

The following conclusions are developed upon the basis of the data collected in 1933 only:

7. The Oliensis test is more sensitive than the microscopic test in the detection of materials that have been subjected to excessively high temperatures in manufacture. However, neither test seems definitely to distinguish products that will weather badly.

8. The use of Vita glass in place of plate glass for the cover of the exposure boxes did not materially change the results. However, because of the lateness in the year when these tests were started the results are considered inconclusive.

9. If like periods of the year are used for exposure, satisfactory check tests can be obtained with the exposure assembly used in these investigations.

Many of the laboratory heat tests have been criticized as producing conditions dissimilar to and more severe than service conditions. These investigations have shown that the physical and chemical characteristics generally believed to belong to unsatisfactory materials are developed upon exposure in many products that satisfactorily withstand laboratory testing. While it is possible, by the utilization of identification tests, to restrict materials to a limited number of sources or manufacturing processes, it is impossible to predict, with any degree of accuracy, the weather-resisting properties of the material thus obtained. It is believed that efforts should be directed to the modification of some of the present laboratory heat tests so that differences in the tendency of various materials to develop unsatisfactory residues may be recognized.

TABLE	14.— <i>Comp</i>	oarison of	1932 and	l 1933 exposure	e tests
	TESTS ON	RESIDUE	S AFTER	EXPOSURE	

Sample identification														
Test characteristic	1	.7	1	8	1	9	2	0						
	1932	1933	1932	1933	1932	1933	1932	1933						
Loss in 10 weeks, percent Penetration at 77° F Penetration at 32° F. Softening point, °F. Ductility at 34°-35° F., centimeters. Insoluble in CS4, percent Insoluble in CS4, percent Insoluble in CS4, percent Percent Loss in 15 weeks, percent Penetration at 32° F. Softening point, °F. Ductility at 34°-35° F., centi- meters Insoluble in CS2, percent Insoluble in CS2, percent Insoluble in CS2, percent Insoluble in CS4, percent	18.3 41 17 163 1.0 .0 1.06 8.27 34.8 19.8 19 12 181 0.3 .0 2.50 9.11 37.2	18.7 36 14 170 0.5 .0 3.17 9.66 36.2 19.5 29 12 168 0.5 .0 3.30 9.88 35.4	18.5 132 37 106 110+ 1.5 .10 17.6 18.7 93 29 113 86.0 4.0 .17	17. 7 152 43 105 110+ 7. 5  17. 3 19. 0 93 28 116 115. 0 4. 5 10	$\begin{array}{c} 13.3\\ 40\\ 9\\ 121\\ 110+\\ .0\\ 5.96\\ 14.94\\ 41.1\\ 13.7\\ 28\\ 8\\ 129\\ 110+\\ .0\\ 6.17\\ 15.61\\ 38.8\\ \end{array}$	$12.0 \\ 44 \\ 10 \\ 121 \\ 110+ \\ .0 \\ 6.50 \\ 16.02 \\ 38.7 \\ 13.8 \\ 29 \\ 7 \\ 128 \\ 110+ \\ .0 \\ 7.50 \\ 18.15 \\ 39.9 \\ 100+ \\ .0 \\ 18.15 \\ 39.9 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ 100+ \\ .0 \\ .0 \\ .0 \\ .0 \\ .0 \\ .0 \\ .0 \\$	3.3 .05 26.8 13.7 67 24 121 42.0 1.3 .14	11. 5 136 41 112 90. 0 3. 8 						
-	37.2 35.4 19.6 18.8 38.8 39.9 28.5 27 N HUBBARD-FIELD CYLINDERS													
Loss of bitumen: In 5 weeks, percent In 10 weeks, percent Stability at 77° F.: Original cylinders, pounds. After 5 weeks' exposure, pounds After 10 weeks' exposure, pounds After 15 weeks' exposure, pounds Series 2, distillation residue cylinders, pounds Series 2, asphaltic residue cylinders, pounds	15 18 21 275 1, 100 1, 575 1, 550 1, 475	15 19 21 150 950 1,000 1,200 1,275	18 19 21 200 800 1, 125 1, 650 750	16 20 20 125 825 950 1, 200 500	8 10 14 325 1, 375 3, 175 4, 050 800	7 11 175 1, 650 2, 525 3, 075 625	9 12 13 225 775 1, 525 1, 975 500 3, 325	8 12 13 100 725 1,000 1,250 350 2,050						

<sup>1</sup> Differences in stability probably are caused by differences in grading of the sand.

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	Percentage change in	sumption from pre-	vious year .		13.0 4.3 2.1		25.5 25.5 7.8 7.8	20.0 10.1 9.4	2.3867 7.3867 7.3867 7.3	23.2 33.2 33.2	11.2 28.5 16.8 3.9	12.9 1.3 11.2 17.6	8.77 6.75 8.77	** 15.6 8.7 8.7	11.1 14.9 14.9 12.6	10.1 15.8 1.5 25.5		7.5
		other uses	Classes of use		HN		AV		HN S	ΗN	В	HN	HN	HN	1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k 1 k			
	By use 5	For oth	Amount	1,000 gallons	<sup>9</sup> 5, 896		253		4, 870	12, 763	112	6, 111	120, 943	9, 752	1         1         1         1           3         4         4         1           5         4         4         1           6         5         4         1         1           8         4         5         4         1         1           8         5         4         1         1         1           8         5         4         1         1         1           9         6         6         1         1         1           9         6         6         1         1         1           9         6         6         8         1         1           9         6         6         8         1         1		011 001	162, 446
Classification of taxed motor fuel		- - -	ror nign- way use 5	1,000 gallons	60, 565 119, 680 1, 198, 655	$\begin{array}{c} 143,290\\ 248,658\\ 39,514\end{array}$	57, 300 970, 874 438, 743	283, 876	110, 924 13 195, 663 566, 735 698, 681	361, 512 112, 666 478, 764 73, 271	22, 355 68, 641 567, 727	$\begin{array}{c} 51,134\\ 1,464,242\\ 273,686\\ 75,390\end{array}$	910, 214 270, 432 144, 917	102, 834 89, 245	791,005	$\begin{array}{c} 239,187\\ 142,393\\ _{e^{-e}}384,981\end{array}$	~ 103, 129	12, 485, 421
on of taxed		es 4	Classes of use		(8)	(4)	AV	(11)	U U AV	ΗN	В	ΗN	ΗN	ΗN				
Olassificati	of tax	At reduced rates <sup>4</sup>	Rate per gallon	Cents	6, 5, 4, 2	4	2/2	4	$\frac{1}{3}$	1	2	1	1	2				
	By rate of tax	Ati	Amount	1,000 gallons	17, 162	68, 909	253	23	$\begin{array}{c} 4,870\\ 1,702\\ 1,149\end{array}$	12, 763	112	6, 111	120, 943	9, 752		k         J         J         k           I         J         J         J         K           I         J         J         J         K           I         J         J         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K           I         J         K         K         K		244, 346
		11.2.4.4	rate	1,000 gallons 154,977	60, 565 108, 414 1, 198, 655	$\begin{array}{c} 74,381\\ 248,658\\ 39,514\\ 235,698\end{array}$	$\begin{array}{c} 239,435\\57,300\\970,874\\438,743\end{array}$	374,998 283,876 184,369 178,434	$\begin{array}{c} 110,924\\ 193,961\\ 566,735\\ 698,681\end{array}$	361, 512 112, 666 478, 764 73, 271	214, 257 22, 355 68, 641 567, 727	$1, \begin{array}{c} 51, 134 \\ 1, 464, 242 \\ 273, 686 \\ 75, 390 \end{array}$	$\begin{array}{c} 910,214\\ 270,432\\ 144,917\\ 1,113,629\end{array}$	$\begin{array}{c} 102,834\\ 128,646\\ 89,245\\ 201,627\end{array}$	$\begin{array}{c} 791,005\\ 62,858\\ 48,550\\ 249,540\end{array}$	239, 187 142, 393 384, 981 44, 111	103, 129	15, 210, 135
	Amount on which	tax was earned		1,000 gallons 154, 977	$\left  \begin{array}{c} 60, 565 \\ 125, 576 \\ 1, 198, 655 \\ \end{array} \right $	$\begin{smallmatrix} 143, 290\\ 248, 658\\ 39, 514\\ 235, 698 \end{smallmatrix}$	239, 435 57, 553 970, 874 438, 743	274, 998 283, 876 184, 369 178, 457	115, 794 195, 663 566, 735 699, 830	$\begin{array}{c} 361, 512\\ 125, 429\\ 478, 764\\ 73, 271 \end{array}$	214, 275 22, 355 68, 641 567, 839	$\begin{array}{c} 51,134\\ 1,464,242\\ 279,797\\ 75,390\end{array}$	$1,031,157\\270,432\\145,514\\1,113,629$	$\begin{array}{c} 102,834\\ 128,646\\ 98,997\\ 201,627\end{array}$	$\begin{array}{c} 791,005\\ 62,858\\ 48,550\\ 249,540\end{array}$	239, 187 142, 393 384, 981 44, 111	103, 129	15, 454, 481
	o refund re tax		Classes of use		NH, E NH, C NH, C	NH, D, E NH NH	HNN	HN	12 NH F, NH NH	HN HN	E, D NH NH	HN HN NH	D, R NH, E	NH, E F D, R	HN	HNN	ΗN	
	Subject to refund of entire tax		Amount	1,000 gallons	10, 280 3, 953 122, 030	24, 013 6, 275 2, 042	5,961 54,877 26,895	28, 805	$\begin{array}{c} 10,616\\ 21,093\\ 35,763\end{array}$	$\begin{array}{c} 43,500\\11,743\\12,211\end{array}$	2,414 2,352 2,011	5, 043 37, 221 21, 485	673 20, 464	6, 030 1, 960 7	84, 133 14, 562	21, 591 5, 217 35, 744	169	681, 656
	Gross amount	assessed for taxation		1,000 gallons 154, 977	$\begin{array}{c} 70,845\\ 129,529\\ 1,320,685\end{array}$	$167, 303 \\ 254, 933 \\ 41, 556 \\ 235, 698 \\ \\ 235, 698 \\ \\ \end{array}$	$239, 435 \\ 63, 514 \\ 1, 025, 751 \\ 465, 638$	403, 803 283, 876 184, 369 178, 458	$\begin{array}{c} 115, 794 \\ 206, 279 \\ 587, 828 \\ 735, 593 \end{array}$	405, 012 125, 429 490, 507 85, 482	216, 671 24, 707 70, 652 567, 839	$ \begin{bmatrix} 56, 177 \\ 1, 501, 463 \\ 279, 797 \\ 96, 875 \end{bmatrix} $	$\begin{smallmatrix} 1, \ 031, \ 830 \\ 270, \ 432 \\ 165, \ 978 \\ 1, \ 113, \ 629 \end{smallmatrix}$	$\begin{array}{c} 108, 864 \\ 130, 606 \\ 98, 997 \\ 201, 634 \end{array}$	$\begin{array}{c} 875,138\\ 62,858\\ 48,550\\ 264,102\end{array}$	260, 778 147, 610 420, 725 44, 111	103, 820	16, 136, 137
;	Shrinkage allow- ance, discounts, etc. <sup>3</sup>		Amount	1,000 gallons	$\begin{array}{c} 713\\ 1, 276\\ 13, 340\end{array}$	3, 353 2, 549 3, 125	2, 185	$\begin{array}{c} 12,  717 \\ 7,  575 \\ 5,  519 \end{array}$		$12,534\\2,547\\15,170\\2,523$	6, 726	15, 167 1, 877	$\begin{array}{c} 31,912\\7,985\\ \end{array}$	$\frac{4}{2}, \frac{125}{075}$	8, 840 1, 978	10, 788	1,601	201, 812
	Shrinka ance, di		Per- cent- age			2 1 (10)	(01)	m 10 m		100100	en	115	P3 00 00	4	3 1	21/2	67	
	Exempted from pay- ment of tax <sup>2</sup>		Classes of use		F, F	ғ, ғ ғ, ғ	F.	F, NH, IC	F. F.	(13) F, E	F, E, IC H F, P, NH, E	$\substack{\mathrm{F,E,IC}\\\mathrm{F,P}\\\mathrm{F,AV}\end{array}$	F, E, IC F, IC, NH	F, F	Æ		F, P	
	Exempte		Amount	1,000 gallons	3, 944 3, 444 22, 361	2, 016 50, 757 7, 564	2, 203 2, 314	7, 366 87, 330	1, 200 1, 373 2, 797	$\begin{array}{c} 1,908\\ 2,180\\ 2,562\end{array}$	2, 497 164, 922	$ \begin{array}{c} 3,922\\52,511\\2,540\end{array} $	239, 900 33, 748	124, 303 4, 148	9, 824		42, 186	882, 618
	Gross	reported by State <sup>1</sup>		1,000 gallons 154,977	$^{*75, 502}_{*134, 249}$ 1, 356, 386	*172, 672 *308, 239 41, 556 246, 387	$\begin{smallmatrix} 243,823\\65,828\\1,025,751\\465,638\end{smallmatrix}$	*378, 781 *378, 781 184, 369 183, 977	$116,994 \\ 207,652 \\ 207,652 \\ 735,593$	419,454 130,156 505,677 *90,567	$^{*224}_{70, 652}$	$^{*60, 997}_{1, 569, 141}$ $^{284, 214}_{264, 214}$	*1, 303, 642 *312, 165 *165, 978 1, 136, 343	*233, 167 130, 606 103, 122 207, 857	893, 802 64, 836 264, 102 264, 102	260, 778 147, 610 431, 513 44, 111	147, 607	17, 220, 567
		State		Alabama	Arizona . Arkansas. California	Colorado Connecticut Delaware Florida.	Georgia Idaho Illinois. Indiana	Iowa Kansas Kentucky Louisiana	Maine Maryland. Massachusetts. Michigan.	Minnesota. Mississippi Missouri. Montana.	Nebraska Nevada. New Hampshire. New Jersey.	New Mexico New York. North Carolina. North Dakota.	Ohio <sup>15</sup> Oklahoma. Oregon. Pennsylvania <sup>16</sup>	Rhode Island South Carolina. South Dakota. Tennessee.	Texas. Utah. Vermont. Virginia.	Washington . West Virginia Wisconsin .	District of Columbia	Total

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F-Sales to Federal Government. P-(Publie) sales to State, country, or municipal governments. E-Fuel exported to other States or countries. IC-Fuel moving through the State in interstate commerce.

NH-Uses other than for propelling motor vehicles on the highways. C-Use in public construction. A V-Aviation use. U-Use in vehicles licensed to operate exclusively in cities.

B-Motor-boat use. D-Fuel destroyed by fire, acts of God, etc. R-Routine refunds (overpayment, etc.).

<sup>1</sup> In this column is given the total amount of motor fuel reported, prior to deduction of exempted fuel, allow-mee for sintilaxes, etc., and amount subject to retund. Wherever possible, thel exempted because of export or interstate movement, or to avoid duplication of ax payment, has been eliminated, in order that the total may represent a closely as possible the total consumption of max partner, has been eliminated, in order that the total may and interstate movement, or to avoid duplication of ax payment, has been eliminated, in order that the total about represent a closely as possible the total consumption of max partner are included in the total abour. <sup>4</sup> A number of States failed to report exempted tuel. Symbols are given only where amounts are reported. <sup>4</sup> A number of States failed to report exempted the index the percentage is computed on the grous a kilowanes for loss by leakage, versporation, etc., and percentage fistes the percentage is fred, in others reported, in others on the maximum allowable. <sup>4</sup> In the case of Arkanass and Colorado, where the rate was changed during the year, the amounts taxed at the lower rates, land 4 cents, respectively, are shown under this heading. <sup>5</sup> The purpose of this classification is to distinguish heading. <sup>6</sup> The purpose of the classification is to distinguish heading. <sup>6</sup> These percentages are based on the amount taxed <sup>7</sup> the purpose of this distinction, the distification is an ittle. <sup>8</sup> These percentages are based on the amount taxed <sup>8</sup> the maximum allowable. <sup>9</sup> These percentages are based on the amount taxed <sup>9</sup> the rates of this distinction, the dissification is antited. <sup>9</sup> These percentages are based on the amount taxed <sup>9</sup> there is no classification by use excerting is based in the total amounts taxed <sup>9</sup> there is no classification of the amount taxed <sup>9</sup> the rate of 10 which the Mation-wide percentage is based on the total amounts taxed <sup>9</sup> there is no classification of the amount taxed <sup>9</sup> there is no the stotal on which the Mation-wide percenta

<sup>7</sup> Refunds on nonhighway use not allowed after Feb. 12, 1934.
 <sup>7</sup> Aft2000 gallons at 6 cents prior to Feb. 13, 1934. Tared at reduced rates at State borders: At 5 cents, 138,000 e Blons at 6 cents prior to Feb. 13, 1934. Tared at reduced rates at State borders: At 5 cents, 138,000 e Blons at 6 cents prior to Feb. 13, 1934. Tared at reduced rates at State borders: At 5 cents, 138,000 e Blons; at 2 cents, 935,000 gallons.
 <sup>9</sup> Estimated by State
 <sup>9</sup> Estimated by State
 <sup>9</sup> Actual allowance reported: no fixed percentage.
 <sup>9</sup> Additional.
 <sup>10</sup> A state of 4 cents per gallon applies to any gases enclored the field of the field field field.
 <sup>10</sup> A 3-cent tax is imposed on motor-vehicle fuel, and a 1-cent tax on all fiquid fuels, including fuel oil and fuel reported was 1,305,642,000 gallons; the gross amount of motor-vehicle fuel reported was 1,234,786,000 gallons.
 <sup>10</sup> Tax is imposed on all liquid fuels, including fuel oil and kerosene.

# STATE MOTOR-FUEL TAX EARNINGS, 1934

[Compiled for calendar year from reports of State authorities 1]

	Tax	rate					Cl	assificatio	on of tax e	arnings <sup>1</sup>		Oti	her earn	ings in a	connect	ion	
	per g	allon		Gross	Re-	Net earn- ings	B	y rate of :	tax	By use o	of fuel <sup>2</sup>		with n	iotor-fue	el tax <sup>3</sup>		Grand
State	On Jan. 1	On Dec. 31	Date of rate change	tax as- sessed 4	funds earned or paid 5	on all motor- fuel taxed 4	At full rate	At reduc Rate per gallon	ed rates 6	For high- way use <sup>2</sup>	For other uses	Dis- tribu- tors' li- censes	Deal- ers' li- censes	In- spec- tion fees	Other fees, etc.	Total	total earn- ings
Alabama Arizona Arkansas	Cents 6 5 6	Cents 6 5 6 <sup>1</sup> /2	Feb. 3	1,000 dollars 9,299 3,542 8,118 20,621	1,000 dollars 514 254	1,000 dollars 9,299 3,028 7,864 35,960	1,000 dollars 9,299 3,028 7,047	Cents 6, 5, 4, 2	1,000 dollars <sup>7</sup> 817	1,000 dollars 3,028 * 7,481 35,960	1.000 dollars 8 383	1.000 dellars	1,000 dollars	1,000 dollars	1,000 dollars 1	1,000 dollars 1	1,000 dollars 9, 29 3,029 7, 864 35, 970
California	3	3	Feb. 1 and	39, 621 7, 591	3,661 1,116	6,475	35, 960 3, 719	64	2,756	6, 475		10				10	6, 475
Connecticut Delaware Florida	2 3 7	$\begin{array}{c}2\\3\\7\end{array}$	Sept. 1.	5,099 1,246 16,499	126 61	4, 973 1, 185 16, 499	$\begin{array}{c} 4,973 \\ 1,185 \\ 16,499 \end{array}$			4,973 1,185			(10) <sup>(1)</sup>			35	$\begin{array}{r} 4,973 \\ 1,185 \\ 16,534 \end{array}$
Georgia Idaho Illinois Indiana	6 5 3 4	6 5 3 4		14,366 3,169 30,772 18,626	298 1,646 1,076	$14,366 \\ 2,871 \\ 29,126 \\ 17,550$	14,3662,86529,12617,550	21/2	6	2, 865 29, 126 17, 550	6				1 5 20	1 $5$ $20$	$14, 366 \\ 2, 872 \\ 29, 131 \\ 17, 570$
Iowa Kansas Kentucky Louisiana	3 3 5 5	3 3 5 5		12, 114 8, 516 9, 218 8, 923	864	$11, 250 \\ 8, 516 \\ 9, 218 \\ 8, 923$	11, 250 8, 516 9, 218 8, 922			11, 250 8, 516		1 5		76	4 37 3	5 118 3	$     \begin{array}{r}       11, 255 \\       8, 634 \\       9, 221 \\       8, 923     \end{array} $
Maine Maryland Massachusetts Michigan	4 4 3 3	4 4 3 3		4,632 8,251 17,635 22,068	$     \begin{array}{r}       146 \\       442 \\       633 \\       1,090     \end{array} $	4, 486 7, 809 17, 002 20, 978	4, 437 7, 758 17, 002 20, 961	1 3 1½	49 51 17	$\begin{array}{r} 4,437 \\ 12,7,809 \\ 17,002 \\ 20,961 \end{array}$	49				2	6	4, 486 7, 809 17, 002 20, 984
Minnesota Mississippi Missouri Montana	3 6 2 5	3 6 2 5		$12,150 \\ 7,519 \\ 9,810 \\ 4,274$	1,305 631 235 610	10,845 6,888 9,575 3,664	10,8456,7609,5753,664	1	128	$ \begin{array}{c} 10,845\\6,760\\9,575\\3,664 \end{array} $	128		1	97 98	<sup>13</sup> 137 8	98 137 106	$ \begin{array}{c} 10, 943 \\ 7, 025 \\ 9, 681 \\ 3, 664 \end{array} $
Nebraska Nevada New Hampshire New Jersey	4 4 4 3	4 4 4 3		8, 667 988 2, 826 17, 035	97 94 80 1	8.570 894 2,746 17,034	8, 570 894 2, 746 17, 032	2	2	894 2,746 17,032	2		25	85 	1	86 64	8,656 894 2,746 17,098
New Mexico New York North Carolina North Dakota	5 3 6 3	5 3 6 3		2,809 45,044 16,788 2,906	$252 \\ 1, 117 \\ 306 \\ 644$	2,557 43,927 16,482 2,262	2,557 43,927 16,421 2,262	1	61	2,557 43,927 16,421 2,262	61	9	9 58	14 704	5	18 58 709	2,575 43,985 17,191 2,262
Ohio Oklahoma Oregon Pennsylvania	4 4 5 3	4 4 5 3		38, 982 10, 817 8, 299 33, 409	1,364 1,047	37,618 10,817 7,252 33,409	36,409 10,817 7,246 33,409	1	1, 209	36, 409 10, 817 7, 246	1,209				4	4	37, 618 10, 821 7, 252 33, 413
Rhode Island South Carolina South Dakota Tennessee				$2, 177 \\7, 836 \\3, 960 \\14, 114$	120 117 195	2,057 7,719 3,765 14,114	2,057 7,719 3,570 14,114	2	195	2, 057 3, 570	195		3			3	2,060 7,719 3,765 14,114
Texas Utah Vermont Virginia.	4 4 4 5	4 4 4 5		35,005 2,514 1,942 13,205	3, 365 	31,6402,5141,94212,477	31, 640 2, 514 1, 942 12, 477			31, 640			1		111	111 1	31,7512,5151,94212,477
Washington West Virginia Wisconsin Wyoming	54444	5 4 4 4		$13,039 \\ 5,905 \\ 16,829 \\ 1,764$	$1,080 \\ 209 \\ 1,430$	$ \begin{array}{r} 11,959\\5,696\\15,399\\1,764\end{array} $	11,959 5,696 15,399 1,764			11,9595,69615,399		2	5		1	8	$     \begin{array}{r}       11,959 \\       5,704 \\       15,399 \\       1.768     \end{array} $
Dist. of Columbia Detailed totals <sup>16</sup> .	2	2		2, 077	14	2,063	2,063			2,063	9.056						2,063
Grand totals		1	3. 66	591, 995	26, 968	565, 027	559,729		5, 298	434, 634	2,056	30	141	1,099	345	1,615	566, 642
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<sup>1</sup> See preceding table for gross gallons of motor fuel reported, exemptions, allowances, etc., gross gallons taxed, gallons subject to refund, net gallons taxed, and information regarding classes of use exempted, subject to refund, or taxed at lower rates.
<sup>2</sup> The purpose of this classification is to distinguish between the tax earnings on motor fuel sold for use in motor vehicles on the highways and tax earnings on motor fuel sold for other uses. In the case of those States that do not make this distinction, the classification is omitted.
<sup>3</sup> Amounts less than \$500 not tabulated.
<sup>4</sup> In the great majority of cases the assessments or earnings of the calendar year were reported. A few States reported the actual collections of the year, rather than refunds claimed on motor fuel purchased during the year. The refunds tabulated include both refunds of the entire tax and partial refunds.
<sup>8</sup> In the case of Arkansas and Colorado, where the rate was changed during the year, the tax earnings at the lower rates, 6 and 4 cents, respectively, are shown under this heading.

<sup>7</sup> Includes \$445,000 on 6-cent tax prior to Feb. 13, 1934, and amounts at reduced rates at State borders, as follows: At 5 cents, \$7,000; at 4 cents, \$347,000; at 2 cents, \$18,000.
<sup>8</sup> Estimated by State.
<sup>9</sup> Rate was 5 cents from Feb. 1 to Aug. 31, 1934.
<sup>10</sup> Retail gasoline station licenses, \$45,000, included in report on motorvehicle receipts.
<sup>11</sup> Includes distributors' licenses.
<sup>12</sup> Refunds are made on all nonhighway uses with the exception of fuel used in commercial motor boats. Earnings on motor-boat fuel (amount not reported) are included.

in commercial motor boats. Earnings on motor-boat fuel (amount not reported) are included. <sup>13</sup> Includes \$138,560, earnings on special gasoline tax collected in Gulf Coast counties (Hancock, Harrison, and Jackson) for seawall protection, and \$1,559 in penalties, less \$2,629, refunds for notary fees. <sup>14</sup> Inspection fees on gasoline and kerosene; bulk of receipts on gasoline. <sup>15</sup> Includes dealers' licenses. <sup>16</sup> Classification by use includes 36 States and the District of Columbia. <sup>17</sup> Weighted average rate.

			S AVAILABLE	1935 Public Works Funds	\$ 430,180 112,202	74.974 12.768 52.774	4, 589 33, C68 741, 845	249,207 73,045	254,000 112,710	239.706 1.652 36.389	587.503 131.272 25.971	265,571 17,839	2,178 7,528 24,000	350,605 169,549	518,146 677,556 593,305	146,499 69,449 196,623	10, 200 396, 775 124, 290	109,304 434.738 135,271	6,333 87,067 9,884	320,389 33,644 35,444	315,362	8,233,430
(SUNU			BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS	1934 Public Works Funds	\$ 6,795 10,772 17,126	1,360 3,063	40 23,507 139,193	7,986 11,330 62,579	8,878 55,786	8,173 14,056 90,213	63 55,067 16,930	60,337 80,043 51,554	2,963 20,566	34,100 237,906	321,703 41,167 13,669	4,272 45,196 58,508	35,811 46,881	67.756 7.787	27,273	2,093 6,403 4,502	18,793	1,733,986
(1935 FU				Milcage	11.8 6.2	22. K	2.8	.1 .12.51	÷. •	1.3	14.3 28.5	30.3	a 20 ***	0.	21.1 82.3	6-1	1.9	2.0 1.1	1.6	3.1	1.4	373.5
. <mark>ON</mark> JNE 18, 1934	ES		APPROVED FOR CONSTRUCTION	1935 Public Works Funds	\$ 195,768 229,602	906,903	205.345 185,822	1,670 177,386 238,513	13,700 26,242 14,820		313,975 141,226	601,627 776,607 140,318	75,122 13,205	10,167 18,900	322,565 151,500 600	143,167	51,578 96,982	183,4445 694,311 17,000	3,866 157,378 130,598	47,222 69,796 43,808	183,992	6,585,135
STATUS OF UNITED ŞTATES PUBLIC WORKS ROAD CONSTRUCTION THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS) ROJECTS ON THE FEDERAL-AID HIGHWAY SYSTEM OUTSIDE OF MUNICIPALITIES	NICIPALITI		APPROVED	1934 Public Works Funds	\$ 20,073 15,039 39,563		23,968 6,394	3,000 77,951		51,291	60,000	23,588 245,950 63,005	17.734	11,623	91,438 143,824	6,016 2,126	18,520	2,195	4,042 92.935	27,869	20+973	1.139.370
	OF MUI			Mileage	109.8 32.22 76.2	55-5 30-5 17-1	3.2 30.1 86.1	39-7 67.4 149.6	103.1 125.1 63.9	27.3 11.6 19.3	20.2 145.9 110.1	1444.1 78.2 58.7	96.4 147.2 7.5	14.5 52.7 133.4	127.4 178.6 62.1	70.7	9.54 19.61 192.4	53.2 338.2 28.4	14.1 69.7 15.1	14.8 67.4 193.5	20.3	3,693.4
	A OUTSIDE		RUCTION	1935 Public Works Funds	\$ 1,364.071 775.339 742.291	2,122,000 516,692 554,726	140,992 606,649 859,437	517.743 2.086.143 2.330.037	1,555,141 1,531,325 977,685	1,047,066 461,258 129,670	2,637,386 637,386 626,499	1,276,205 1,881,117 791,682	1,688,363 626,302 240,113	560, 360 629,496 3,027,452	665, 944 324, 435 2,692, 721	1,436,273 1,103,852 3,266,673	295,834 275,918 809,360	1, 235, 361 4, 226, 173 341, 550	331,049 1,043,049 938,045	413,465 1,354,392 927,921	99,424	55,080.948
BLIC WO ACT (1934 F	AY SYSTEP	35	UNDER CONSTRUCTION	1934 Public Works Funds	\$ \$60,140 700,927	122,227 77.176 295,028	8,477 66,825 922,864	131.759 2.002.682 635.108	49,000 31,530 287,374	234,221 11.351 900,555	52,687 782,950 217,577	723,191 501.643 4,129	21,644 204,818	1,113,074	663,506 74,036 227,251	348, 296 102, 555 172, 976	79.740 315.630 426.644	176,547 247,933 37,000	198,838 227,922	82,643 273,603 141,021	1,126,854	16,605,7 <sup>1µµ</sup>
ATES PUI	WH9IH QL	JULY 31, 1935		Estimated Total Cost	\$2,147,304 902,109 1,662,770	3,163,930 625,115 1,137,322	149,545 711,020 1,865,325	681,542 4,088,825 3,018,635	2,002,083 1,630,648 1,398,252	1,434,550 472,624 1,030,226	1,099,880 3,441,400 846,799	2,523,432 2,541,590 897,888	2,054,603 884,019 243,690	1,770,707 631,786 7,064,830	1,435,381 536,450 3,255,174	2,000,652 1,292,234 3,548,329	391,864 511,548 1,285,130	1,508,309 4,673,720 480,185	1,447.575 1,286,948	496,108 1,707,503 1,095,104	1,430,996	80,941,786
ED ŞT <i>i</i> trial f	ERAL-A	AS OF JU		Milcage	358.2 362.8 184.5	370.4 262.3 20.8	45.6 125.7 338.9	205.7 40.3 123.3	331.3 637.9 283.3	78.3 52.7 18.6	37.4 242.3 960.4	292.9 217.8 577.3	378.1 332.7 15.8	38.1 374.7 219.2	645.2 1,165.4 203.0	330.6 193.9 149.9	223.4 625.0	201.7 1.161.1 274.7	53.2 183.1 114.9	86.5 238.6 539.9	19.3	13,962.3
OF UNIT	N THE FEL	A	ED	1935 Public Works Funds	# 139,902 563,373 629,904	609,765 1,895,044	316.118 271.538 769.641	363,290 145,249 47,038	394.520 796.564 196.985	93,648 319,285 123,550	143,650	688,779 232,942 1,764,369	216,519 703,321 201,292	29,746 1,047,273 457,330	423,710 315,993 252,630	616,650 279,440 1,086,377	168,738 214,683 493,190	577, 343 1,503, 032 572, 525	124.795 632,668 1474,277	359,092 361,138 679,195		24,036,147
	ROJECTS OI		COMPLETED	1934 Public Works Funds	\$ ,460,745 3,829,744 2,576,551	7,789,341 3,357,026 1,109,185	868,470 2,355,070 3,977,140	2,027,114 2,391,815 4,243,283	4,978,530 5,004,394 3,408,445	2,399,450 1,541,605 791,495	1,048,966 5,213,496 4,266,504	2,682,221 4,409,896 4,345,161	3,859,875 2,666,269 692,118	2,025,846 2,835,025 9,094,974	3,684,500 2,643,198 7,036,839	4,249,815 2,905,698 6,457,583	899,627 2,359,621 2,532,214	3,999,811 11,305,035 2,322,418	912,376 3,412,159 2,830,012	1,900,800 4,392,512 2,105,140	526,724	165.756,136
	CLASS 1.—PF			Total Cost	\$6,527,856 5,076,955 3,828,359	11,199,998 5,475,319 1,135,531	1,204,411 3,437,615 4,954,069	2,462,814 2,583,089 4,415,382	5,508,127 5,933,599 3,856,210	2,864,278 1,890,999 916,280	1,473,517 5,670,800 6,200,552	5,647,003 5,254,610 6,580,382	5,154,215 3,442,296 937,865	2,247,203 4,093,367 11,942,977	5,168,217 3,428,060 7,683,377	5,005,045 3,584,581 7,956,871	1,136,743 2,638,860 3,514,733	5,226,315 13,416,341 3,170,271	1,135,574 4,340,679 3,329,879	2,301,844 4,971,601 2,947,719	829,029	213.731.417
CURRENT AS PROVIDED BY SECTION 204 OF	0		IMENTS	Act of June 18, 1934 (1935 Fund)	# 2,129,921 1,738,712 1,714,000	3,713,643 2,424,504 607,500	461,697 1,116,600 2,556,745	1,131,910 2,408,778 2,688,633	2,217,361 2,354,131 1,302,209	1,380,419 782,195 289,609	1,582,874 3,226,284 2,533,733	2,832,182 2,890,666 2,714,208	1,982,182 1,350,356 1,65,404	951.379 1.676.769 3.673.231	1,930,365 1,469,484 3.539,255	2,342,590 1,452,741 4,554,082	474,772 940,954 1,523,821	2,105,454 6,858,253 1,066,345	466,042 1,916,178 1,553,206	1,140,167 1,818,970 1,686,368	598.778	93,935,660
AS PROVID			APPORTIONMENTS	Sec. 204 of the Act of June 16, 1933 (1934 Fund)	\$ 3.947.753 3.855.555 3.334.167	7,912,928 5,437,265 1,404,213	877,566 2,469.370 5,045,592	2,166,858 4,406,858 5,018,921	5,027,830 5,044,802 3,751,605	2,693,135 1,567,012 1,782,263	1,101,716 6,051,533 4,561,011	3,489,337 5,237,532 4,463,849	3,914,481 2,909,387 692,118	3,173,019 2,846,648 10,465,672	4,761,147 2,902,224 7.277,758	4,608,399 3,053, <sup>048</sup> 6,691,194	979.367 2.729.583 3.005.739	4,246,309 11,588,643 2,367,205	928,184 3.731.207 3.057.934	2,013,405 4,697,518 2,250,663	1,693,344	185, 235, 236
				STATE	Alabama Arizona Arkansas	California Colorado Connecticut	Delaware Florida Georgia	Idaho Illinois Indiana	Iowa Kansas Kentucky	Louisiana Maine Maryland	Massachusetts Michigan Minnesota	Mississippi Missouri Montana	Nebraska Nevada New Hampshire	New Jersey New Mexico New York	North Carolina North Dakota Oliio	Oklahoma Oregon Pennsylvania	Rhode Island South Carolina South Dakota	Tennessee Texas Utah	Vermont Virginia Washington	West Virginia Wisconsin Wyoming	District of Columbia Hawaii	TOTALS

	·		OJECTS	1935 Public Works Funds	朱 500,041 61,883 75,931	107,734 20,589 140,004	138,809 94,969 693,715	69,560 516,050 282,342	262,865 213,325	134,271 6,907 452,515	579.317 86.667 415.538	100,163 40,949	146,743 756 14,877	264,787 96,014 393,850	57,611 363,480 471,186	2448,847 92,418 633,092	1444,000 197,422 427,962	246,675 365,123 2,306	51,170 24,730	270,035 20,389 12,500		9,540,167
(DS)			BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS	1934 Public Works Funds	本 24,965 4,857 22,919	7,422 35,346	127 17.554 215.987	16,349 12,435 27,544	39 31,436	7,200 5,038 2444,658	28,646 20,453 65,722	48,163 150,471 32,749	41.657 58.515	124,515 77.794 162,814	49,790 22,388 4,020	43,921 63,976	60,634 196,716	246,459 550	15,036 7,002 1,845	61 29,375 6,474		2,263,622
[1935 FUI	'IES		APPROVED FOR CONSTRUCTION B	Mileage	3.1 5.7	5.5 1.3	2.2 1.2	0.4.4 0.4.6	3.5 3.5	2.7 7.5	5°0	2.5 1.0 1.0	±.6	0.1 ci.	2.1 16.4	۰. و،، و.	.1 7.4	*7 8*7	4°-9	.6 4.2		104.7
<b>DN</b> NE 18, 1934 (	UNICIPALIT			1935 Public Works Funds	\$ 122,470 311,035	194,200 134,613	139,600 31,607	35,000 620,357 329,789	137.700 10.980 145.906	203,681 158,253	130,975 54,188	18,487 665,320 7.553	11,402	631,441 183,963 219,150	91,264 190,563 80,000	126,468 80,000 288,881	15,41 <b>2</b> 125,639	115,946 398,296	39, 244 155, 506. 7, 370	47,791 215,499		6+505+549
STRUCTI	STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS) EXTENSIONS OF THE FEDERAL-AID HIGHWAY SYSTEM INTO AND THROUGH MUNICIPALITIES		APPROVED I	1934 Public Works Funds	\$ 29,878 101,076			47,260 148,115	48,919 30,000	10,666	19,400 4,220	36,943 2,474	26,150 13,044	1,010	91,217 87,346	518	8,705 40,456	16,465	13.975	28,642 57,664		<ul> <li>864.143</li> </ul>
D CON	AND TF			Mileage	23.4 8.5	8.0 .8 1.6	1.1 1.4 18.2	-5 11.1 29.0	23.1 11.3 9.9	14.0 0.50 0.0	5.0 17.2 17.1	23.0 12.4	4.5 .5	5.9 1.7 22.9	13.4 9.9 13.5	9.0 9.3 11.8	, 9 12.8 9.2	7.6 21.3 13.5	2.7 10.3 5.0	6.8 3.4	CJ.	453.2
RKS ROA UNDS) AND	TEM INTO		RUCTION	1935 Public Works Funds	\$ 323,075 156,833 259,979	1,070,344 142,521	18,165 163,804 104,136	189,485 1.085,092 1.532,231	808,800 1,044,466 528,141	259,141 271,842	169,924 1,262,400 575,719	120,417 228,978 15,352	296.070 53.741	804,667 69,432 2,964,090	426,700 88,185 1,374,172	575,152 544,705 749,944	105, 759 251, 547 164, 405	494,651 814,902 433,800	105,082 396,439 359,190	224,149 629,047 14,132		22,570,806
3LIC WOI	ES PUBLIC WORKS OVERY ACT (1934 FUND: OVERY ACT (1934 FUND: AllD HIGHWAY SYSTEM	935	UNDER CONSTRUCTION	1934 Public Works Funds	\$ 4463,060 129,322 248,967	356,840 11,229	2,324 275,754	1,036,345 748,415	508,465 389,559	896, 730 83, 155 262, 457	2,876,372 342,450 513,160	626,097 867,762 64,410		182,690 875,017	103,576 99,844 93,352	192,161 66,635 511,111	156,274 77,024	201,632 671,837 129,130	26,801 246, <del>141</del> 2 53,369	270,594 147,667	250,164	15,058,226
TES PUF	T-AID HIG	JULY 31.1		Estimated Total Cost	# 786,135 323,930 510,198	2,680,138 11,229 193,692	18,165 166,128 703,177	190,673 2,121,438 2,230,824	1,408,378 1.059,769 945,972	1,197,913 360,761 1,075,978	3,071,272 1,642,750 1,183,363	817.550 1.142.578 94.702	296,070 53,951	1,225,905 69,432 3,994,541	548,276 188,029 1,611,417	786,181 619,349 1,354,210	105,759 420,872 241,429	696,283 1,662,930 650,283	137,402 702,604 412,559	494,743 629,047 162,381	250,164	41,300,530
ED ST <i>A</i> TRIAL R	FEDERA	OF		Mileage	19.1 15.1 18.3	65.8 40.4	8.1 21.4 74.8	20.9 66.5 70.4	58.9 45.6 35.8	23.5 17.5 3.6	14.9 40.2 113.9	37.0 52.9 36.6	145.3 10.8 18.7	22.9 40.2 63.4	96.8 54.5 69.0	49.2 32.8 72.7	8.0 38.2 111.5	28.3 134.6 20.8	15.5 24.1 12.5	18.6 62.2 22.5	6*5	2,021+2
OF UNIT	IS OF THE	AS	ED	1935 Public Works Funds	\$ 119,375 86,475 210,081	847,032 169,411 9,362	73,875 102,827 148,915	27,081 8,850 104,496	101,635 377,503 71,226	147,468 47,377	98,359 133,100 376,049	114,956 24,854 49,238	548,278 57,842 173,847	108,606 180,097 384,600	634,661 92,515 1434,145	220,827 150,853 725,785	36.001 23,620 143,905	264, 518 216, 679 97,066	96,285 352,905 385,313	28,109 514,579 2,784	181,051	9,404,416
STATUS	EXTENSION		COMPLETED	1934 Public Works Funds	\$1,872,024 622,804 1,591,572	3,849,724 1,672,058 802,407	1,450,282 1,439,771 2,232,880	1,151,479 6,285,871 3,362,976	2.105,968 2.473,482 1.476,834	793,980 872,233 384,017	2,102,182 3,118,335 3,136,040	1,033,466 3,001,268 1,016,328	1,915,583 473,901 668,776	2,810,715 1,595,354 7,217,830	2,135,990 1,241,533 4,238,314	2,112,039 1,415,650 4,279,869	518,991 1,199,812 1,188,674	1,921,523 5,708,102 649,146	<sup>458,671</sup> 1,681,361 1,922,046	1,042,973 2,509,104 971,191	696,281	97,431,410
CURRENT	OF			Total Cost	\$2,000,380 738,883 1,902,827	5,358,835 1,937,435 838,549	542,664 1,839,561 2,414,917	1,222,156 6,593,153 3,615,513	2,303,093 3,007,698 1,607,817	948,832 925,109 390,021	2, 249, 980 3, 392, 363 3, 575, 293	1,166,485 3,101,773 1,093,057	2,498,414 539,562 845,980	3,080,011 1,778,171 8,085,610	2.791,849 1.341,415 5.155,741	2,415,796 1,617,923 5,208,661	755,890 1,226,620 1,232,982	2,209,866 6,049,343 812,144	636,840 2,266,435 2,330,254	1,108,178 3,102,009 977,168	877,332	111,510,588
SEO	CLASS 2.—PROJECTS		NMENTS	Act of June 18, 1934 (1935 Fund)	\$1,064,961 305,191 857,025	2,219,360 190,000 1426,500	230.849 501.200 1.278.373	321,126 2,230,350 2,248,858	1,311,000 1,432,949 958,599	744,550 484,379 452,515	847,600 1,613,142 1,421,494	354,022 919,152 113,092	991,091 100,000 242,465	1,809,500 529,506 3,961,690	1,210,236 734,741 2,359,504	1,171,295 867,977 2,397,703	285,760 488,000 761,911	1,121,789 1,795,000 533,173	240,611 956,021 776,603	570,085 1,379,513 29,416	181,051	48,020,938
AS PROVIDED BY	CI		APPORTIONMENTS	Sec. 204 of the Act of June 16, 1933 (1934 Fund)	\$ 2, 389, 928 756, 982 1, 964, 534	4,213,986 1,718,633 802,407	460,409 1,459,648 2,724,620	1,197,829 7,381,910 4,287,050	2,614,472 2,522,401 1,927,828	1,708,577 960,426 891,132	5,007,199 3,500,637 3,719,143	1,744,669 4,019,501 1,115,962	1,957,240 500,051 740,335	3,117,921 1,674,158 8,255,661	2,380,573 1,451,112 4,335,686	2,304,200 1,526,724 4,854,988	579,625 1,364,791 1,502,870	2,123,155 6,642,863 778,826	500,509 1,948,780 1,977,260	1,342,270 2,596,143 1,125,332	946,445	115,617,401
				STATE	Alabama Arizona Arkansas	California Colorado Connecticut	Delaware Florida Georgia	Idaho Illinois Indiana	Iowa Kansas Kentucky	Louisiana Maine Maryland	Massach <b>n</b> setts Michigan Minnesota	Mississippi Missouri Montana	Nebraska Nevada New Hampshire	New Jersey New Mexico New York	North Carolina North Dakota Ohio	Oklahoma Oregon Pennsylvania	Rhode Island South Carolina South Dakota	T ennessee T exas Utah	Vermont Virginia Washington	West Virginia Wisconsin Wyoming	District of Columbia Hawaii	TOTALS

			VDS AVAILABLE PROJECTS	1935 Public Works Funds	\$ 104, 426 77,009	155.951 185.099	2,615 22,261 785,690	179.044 334.400 42.347	40,596	18,305 3,102 425,574	82,567 15,866	95.132	49,531 54,195 12,181	182,063 12,218 80,562	15,304 325,960 450,380	174,863 705	4,663 66,495 91,350	221,404 27,169 1,000	57,607 1,264	175,768 115,064 21,077	135,097 189,347	5,039,424
UNDS)			BALANCE OF FUNDS AVAILABLE FOR NEW PROJECTS	1934 Public Works Funds	# 41.549 23,461 53.114	1.700	23,060 109,711	27,032 19,058 35,227	707 19,513	374 36 30	16,445 40,290 51,133	35, 263 1, 716 37, 031	. 2,683		43,584 28,938 70,269	12, 317 143, 080	14,613 2,661	63,978 25,209	3,019	5.338 21,472 2,590	295	901,168
(1935 FU	(1935 FUN		CTION	Milcage	10.6 27.1	<sup>4</sup> *6	2.7	7.8	2.5 2.5	9.3 6.3	4.6 8.0 3.2	16.5 29.0 11.7	12.2 15.2	1.1 7.4	9.6 93.0 6.2	4.2	1.4	5.1 10.6	15.0	7.3 *3	9°#	393.0
I <b>ON</b> JNE 18, 1934			APPROVED FOR CONSTRUCTION	1935 Public Works Funds	\$ 128,940 229,946	.262.527	33, 738	473,156	16,200 29,555	129,772	313,670 42,275 112,651	156,539 224,335 55,435	37,414 169,589	170,412	134,315 305,995 27,200	27,349	36,815 140,098	121,139 50,500	193,855	47,548 123,338 40,927	75.395	11°1410,854
STRUCTI s ACT OF JU			APPROVED	1934 Public Works Funds					# 43.499	130,522 9,500		82,163	ε,Ψι3	65.732	100,391			17,894	54.644			513,088
LD CON	DS			Milcage	52.6 45.8 68.9	56.4 39.2 4.5	14.2	27.0 286.5 50.2	246.0 77.8 167.3	38.7 3.4	15.9 84.9 89.1	40.9 359.5 31.9	58.5 28.0 6.4	1.7 41.4 257.0	101.6 54.6 89.8	56.1 19.0 198.5	6.7 167.5 159.9	37.6 170.8 46.3	5.3 69.3 22.7	28.0 56.7 115.2	2.5	3.734.3
RKS ROA UNDS) ANI	CLASS 3PROJECTS ON SECONDARY OR FEEDER ROADS		RUCTION	1935 Public Works Funds	\$ 634,885 563,989 501,385	1,364,725 320,140 222,880	ы1,690 656,064 Ц39,574	247,061 3.332,895 103,626	1,141,950 1,181,237 1,211,964	589,699 51,608 359,120	1,467,417 701,239	85.852 1.707.303 311.474	405,493 180,528 174,655	107,525 351,820 3,038,790	862,777 50,713 1,316,823	898,864 328,814 2,160,362	212,563 1,148,229 399,219	2,866,130 2,866,130 255,000	74,410 531,717 507,750	346,766 1,196,317 374,548	393,065	36,564,187
STATUS OF UNITED STATES PUBLIC WORKS ROAD CONSTRUCTION THE NATIONAL INDUSTRIAL RECOVERY ACT (1934 FUNDS) AND BY THE ACT OF JUNE 18, 1934 (1935 FUNDS)	ARY OR FE	35	UNDER CONSTRUCTION	1934 Public Works' Funds	\$ 187,785 145,000 147,095	323,897 110,000	215,448 620,788	2,265,472 274,972	119.252 305.453 46.363	209,970	205,227 149,444	460,584 235,551	12,000 29,000	36,931 515,500	160,265 151,236 49,860	251,646 19,526 1,093,809	263,518 174,741	269,714 34,517 94,022	112,886	318,088 258,460 75,284		9.859.427
	N SECOND	JULY 31, 1935		Estimated Total Cost	# 822,670 648,037 650,656	1,833,841 550,809 222,880	265,716 704,065 1,060,362	247,061 5.648,367 378,598	1,555,581 1,486,690 1,341,513	799,670 59,505 375,243	1,672,663 927,633	546,437 1,998,538 311,474	405,493 199,878 206,460	107,525 388,751 4,153,510	1,023.042 201,949 1,418,066	1,299,063 402,753 3,353,144	212,563 1,450,097 573,960	837,013 2,911,390 407,861	80,229 653,915 515,250	664,854 1,661,331 450,373	393,065	48,603,438
ED ST.	JECTS O	AS OF JI		Mileage	143.0 88.5 176.3	180.1 272.6 14.9	57.1 88.5 125.3	196.8 205.0	1445.7 234.7 241.3	50.5 103.1 68.5	15.2 207.7 318.0	143.1 673.2 283.4	430.2 189.2 28.3	-5 253.0 119-5	319.4 387.3 322.6	273.4 154.9 555.1	33.1 130.1 436.9	148.9 872.7 217.4	47.9 216.3 102.1	43.5 179.7 156.8	10.4 4.9	10,021.3
OF UNIT NAL INDUS	SS 3PRO	A	reD	1935 Public Works Funds	# 196.710 357.034 125.693	216,000 551,362 12,889	186,544 331,480 53,105	395.344 91.822 5.500	431.850 149.358 275.369	101,176 390,302 60,271	103,450 640,518	16.500 431.784 575.525	498,603 1417,688 74,757	371,386 573,549	687,943 46,074 169,850	70,199 447,577 341,938	127.275 131.244	165,906 694,201 277,173	166,042 105,009 267,589	308.635 135.376	189.234	11,995,937
	CLA		COMPLETED	1934 Public Works Funds	<pre># 1,803,119 530,962 1,249,425</pre>	3,154,843 1,608,632 659,120	265,665 1,279,756 1,590,474	1,094,530 3,495,503 421,674	2,293,399 2,173,449 1,772,049	1,086,013 842,404 865,179	469,741 2,937,940 2,175,738	1,248,797 2,636,906 1,740,693	1,957,240 1,113,353 1,443,386	55,099 1,235,198 3,027,536	2,176,724 1,170,548 3,751,018	2,052,553 1,494,881 6,207,932	439,716 1,036,655 1,325,467	1,771,569 5,952,792 954,655	435,862 1.551,099 1.080,673	795,133 2,151,283 1,047,457	971.729 177.718	81.877,442
				Total Cost	♣ 2,016,508 985,299 1,382,268	4,050,207 2,601,784 697,300	1,626,676 1,668,552	1,670,559 3,633,423 469,232	2,799,645 2,325,666 2,175,313	1,198,264 1,346,314 973,930	477,470 3,275,590 2,951,675	1,265,297 3,194,579 2,324,901	2,472,234 1,586,093 570,258	56,528 1,606,584 4,111,171	2,868,726 1,217,254 4,172,803	2,304,496 2,115,745 6,718,399	1,214,563 1,458,798	2,004,529 7,114,265 1,487,378	694,760 1,716,601 1,376,024	839,957 2,644,458 1,200,239	1,160,963 178,209	98.913.307
CURRENT AS PROVIDED BY SECTION 204 OF			MENTS	Act of June 18, 1934 (1935 Fund)	年 1,064,960 998,032 857,024	1,999,203 871,502 420,868	230,849 1,043,543 1,278,373	4, 282, 450 4, 282, 273	1.590.000 1.330.595 1.557.503	838,953 445,012 1,067,934	920,000 1,613,142 1,470,324	354,023 2,363,922 942,434	991,091 852,000 261,593	460,000 735,425 3,693,000	1,700,340 734,742 1,966,253	1,171,295 777,096 2,639,003	254,040 1,342,000 761,911	1,075,748 3,658,000 533,173	241,354 893,188 776,603	570,083 1.743.354 571.928	792, 791	58,043,402
AS PROVIDI			APPORTIONMENTS	Sec. 204 of the Act of June 16, 1933 (1934 Fund)	# 2,032,452 599,423 1,449,634	3,450,440 1,718,632 659,120	481,113 1,302,816 2,320,973	1,121,562 5,760,033 731,872	2,413,358 2,522,401 1,837,926	1,426,879 842,479 831,132	428,185 3,184,057 2,376,415	1.744.669 2.923.273 1.559.937	1.957.240 1.136.479 477.336	55,099 1,272,129 3,608,768	2,380,573 1,451,112 3,871,148	2., 304. 199 1.526. 724 7.344.822	439.716 1.364.791 1.502.870	2,123,155 6,012,518 1,048,677	438,380 1,736,770 1,080,673	1,118,559 2,431,220 1,125,332	972.024 177.718	93.147.363
				STATE	Alabama Arizona Arkansas	California Colorado Connecticut	Dclaware Florida Georgia	Idaho Illinois Indiana	Iowa Kansas Kentucky	Louisiana	Massachusetts Michigan Minnesota	Mississippi Missouri Montana	Nebraska Nevada New Hampshire	New Jersey New Mexico New York	North Carolina North Dakota Ohio	Oklahoma Oregon Pennsylvania	Rhode Island South Carolina South Dakota	Tennessee Texas Utah	Vermont Virginia Washington	West Virginia Wisconsin Wyoming	District of Columbia Hawaii	TOTALS

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- Report of the Chief of the Bureau of Public Roads, 1934.

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## TRANSPORTATION SURVEY REPORTS

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- Report of a Survey of Transportation on the State Highways of Vermont (1927).
- Report of a Survey of Transportation on the State Highways of New Hampshire (1927).
- Report of a Plan of Highway Improvement in the Regional Area of Cleveland, Ohio (1928).
- Report of a Survey of Transportation on the State Highways of Pennsylvania (1928).
- Report of a Survey of Traffic on the Federal-Aid Highway Systems of Eleven Western States (1930).

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